

INFLUENCE OF GAS RADIATION MODELS ON  
OXY-COMBUSTION CHARACTERISTICS  
PREDICTIONS

BY

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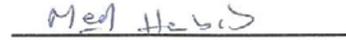
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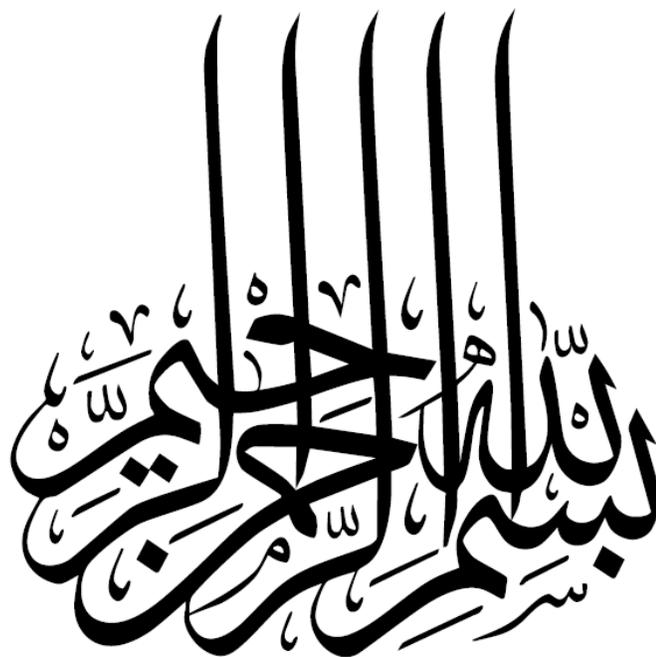
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STARTING BY THE NAME OF  
ALLAH  
THE MOST BENEFICIAL AND THE  
MOST MERCIFUL

*Dedicated to*

*My beloved parents,*

*My Wife*

*and*

*My daughter*

## **ACKNOWLEDGEMENT**

*“In the name of Allah, The Most Gracious and The Most Merciful”*

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# TABLE OF CONTENTS

ACKNOWLEDGEMENT.....	v
LIST OF TABLES.....	ix
LIST OF FIGURES.....	x
LIST OF ABBREVIATIONS.....	xv
THESIS ABSTRACT (ENGLISH).....	xvi
THESIS ABSTRACT (ARABIC).....	xvii
CHAPTER 1.....	1
INTRODUCTION.....	1
1.1 Research Background.....	1
1.2 Problem Statement.....	5
1.3 Objectives.....	6
1.4 Thesis Outline.....	7
CHAPTER 2.....	8
CARBON CAPTURE TECHNOLOGIES.....	8
2.1 Post-Combustion Capture (PCC) Technology.....	8
2.2 Pre-Combustion Capture Technology.....	9
2.3 Oxy-Fuel Combustion.....	10
2.4 Literature review.....	11
CHAPTER 3.....	28
GAS RADIATION MODELS.....	28
3.1 Introduction.....	28
3.2 Radiative Transfer Equation (RTE).....	30

3.3 Radiative Transfer Equation (RTE) Solution Methods.....	36
3.3.1 Introduction.....	36
3.3.2 Overview of computational methods in radiative transfer.....	37
3.3.3 Discrete Ordinate Method (DO).....	39
3.4 Modeling of spectral nature of radiative transfer.....	41
3.4.1 Introduction.....	41
3.4.2 Narrow-band models.....	43
3.4.3 Wide-band models.....	44
3.4.4 Total absorptivity-emissivity models.....	45
3.4.5 Absorption and emission coefficients.....	47
3.5 Simple grey gas model (SGG).....	48
3.6 Exponential wide band model (EWBM).....	50
3.7 Leckner model.....	52
3.8 Perry model.....	55
3.9 Weighted sum of grey gas model (WSGG).....	57
<b>CHAPTER 4.....</b>	<b>62</b>
<b>GAS RADIATION MODELS VALIDATION FOR COMBUSTION</b>	
<b>PROBLEM.....</b>	<b>62</b>
4.1 Introduction.....	62
4.2 Problem description.....	63
4.3 Numerical simulation.....	65
4.4 Results and Discussion.....	67

4.5 Assessment of gas radiation models.....	75
CHAPTER 5.....	85
CFD MODELING OF A WATER TUBE BOILER UNDER AIR AND OXY-FUEL COMBUSTION.....	85
5.1 Introduction.....	85
5.2 Boiler description.....	85
5.3 Problem setup.....	88
5.4 Grid independence study.....	91
5.5 Results and discussions.....	94
5.5.1 Comparisons between gas radiation models.....	94
5.5.2 Characteristics of Oxyfuel and Air-Fuel Combustion Processes.....	103
CHAPTER 6.....	117
CONCLUSIONS AND RECOMMENDATIONS.....	117
6.1 Conclusions.....	117
6.2 Recommendations for future work.....	118
NOMENCLATURE.....	120
REFERENCES.....	122
Vita.....	128

## LIST OF TABLES

Table 1 Coefficients to calculate the H <sub>2</sub> O emissivity with M=3 and N=3.....	54
Table 2 Coefficients to calculate the CO <sub>2</sub> emissivity with M=4 and N=5.....	54
Table 3 Pressure correction parameters used in equation (35) and equation (36).....	54
Table 4 Empirical constants used in equation (42).....	56
Table 5 Coefficients for the WSGG model [41].....	60
Table 6 Coefficients for the WSGG (4+1) model [26].....	61
Table 7 Coefficients for the WSGG (3+1) model [26].....	61
Table 8 Design data of boiler.....	86
Table 9 Different grids of the furnace.....	91

# LIST OF FIGURES

Figure 1 CO <sub>2</sub> emission level [1].....	2
Figure 2 NO emission level [1].....	3
Figure 3 Principles of the three main CO <sub>2</sub> capture options [2].....	4
Figure 4 Post-combustion capture (PCC) process [3].....	9
Figure 5 Pre-combustion capture process [3].....	10
Figure 6 Basic principle of oxyfuel technology [4].....	11
Figure 7 coordinates for derivation of the radiative transfer equation [54].....	32
Figure 8 The Chalmers 100 kW O <sub>2</sub> /CO <sub>2</sub> combustion test unit [42].....	64
Figure 9 fuel burner design [42].....	65
Figure 10 Temperature profile at distance 0.384m from the burner with different radiation models, air combustion.....	69
Figure 11 Temperature profile at distance 0.553m from the burner with different radiation models, air combustion.....	70
Figure 12 Temperature profile at distance 1.4m from the burner with different radiation models, air combustion.....	70
Figure 13 CO <sub>2</sub> concentrations at distance 1.4m from the burner, air combustion.....	71
Figure 14 Temperature profile at distance 0.553 m for different radiation models, OF21 combustion.....	72
Figure 15 Temperature profile at distance 1.4m from the burner with different radiation models, OF21 combustion.....	73

Figure 16 CO <sub>2</sub> concentrations at distance 0.384m from the burner, OF21 combustion.....	73
Figure 17 Temperature profile at distance 0.384m from the burner with different radiation models, OF27 combustion.....	74
Figure 18 CO <sub>2</sub> concentrations at distance 0.384m from the burner, OF27 combustion.....	75
Figure 19 Emissivity profiles as a function of gas temperature for different radiation models, air combustion.....	77
Figure 20 Emissivity profiles as a function of gas temperature for different radiation models, OF21 combustion.....	78
Figure 21 Emissivity profiles as a function of gas temperature for different radiation models, OF27 combustion.....	79
Figure 22 Discretization of a path length.....	80
Figure 23 Radiation intensity at distance 0.384m from the burner with different radiation models, air combustion.....	81
Figure 24 Radiation intensity at distance 0.384m from the burner with different radiation models, OF21 combustion.....	82
Figure 25 Radiation intensity at distance 0.384m from the burner with different radiation models, OF27 combustion.....	82
Figure 26 Radiation intensity at distance 1.4m from the burner with different radiation models, air combustion.....	83

Figure 27 Radiation intensity at distance 1.4m from the burner with different radiation models, OF21 combustion.....	84
Figure 28 Radiation intensity at distance 1.4m from the burner with different radiation models, OF27 combustion.....	84
Figure 29 Layout of furnace of the boiler.....	86
Figure 30 Side view of Furnace of the boiler .....	87
Figure 31 burner construction.....	88
Figure 32 General view of the boiler.....	89
Figure 33 Closer view of the burners.....	90
Figure 34 Temperature profile at line (x=3.264, z=6) along Y-axis.....	92
Figure 35 Temperature profile at line (x=4.539, y=-2.135) along Z-axis.....	93
Figure 36 Total heat flux profile at line (y=2, z=0) along X-axis.....	93
Figure 37 Total heat flux profile at line (x=0, y=3) along Z-axis.....	94
Figure 38 Temperature profile at line (x=4.539, y=4) along Z-axis, air combustion.....	96
Figure 39 Temperature profile at line (x=3.264, y=-1.0675) along Z-axis, air combustion.....	96
Figure 40 Temperature profile at line (x=4.539, y=4) along Z-axis, OF21 combustion.....	98
Figure 41 Temperature profile at line (x=3.264, y=-1.0675) along Z-axis, OF21 combustion.....	98

Figure 42 Temperature profile at line ( $x=3.264$ , $y=-1.0675$ ) along Z-axis, OF27 combustion.....	99
Figure 43 Temperature profile at line ( $x=4.539$ , $y=4$ ) along Z-axis, OF27 combustion.....	100
Figure 44 CO <sub>2</sub> mole fraction profile at line ( $x=3.264$ , $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion.....	101
Figure 45 H <sub>2</sub> O mole fraction profile at line ( $x=3.264$ , $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion.....	101
Figure 46 Incident radiation profile at line ( $x=3.264$ , $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion.....	102
Figure 47 Total heat flux profile at line ( $x=0$ , $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion.....	102
Figure 48 Temperature profile at line ( $x=3.264$ , $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion.....	103
Figure 49 Temperature profiles at line ( $x=3.264$ , $y=-1.0675$ ) along Z-axis for different combustion cases, EWBM.....	104
Figure 50 Specific thermal capacity profiles at line ( $x=3.264$ , $y=-1.0675$ ) along Z-axis for different combustion cases, EWBM.....	105
Figure 51 Contours for temperature of vertical plane passing through the middle burners 2 & 5 ( $x=4.539$ m), EWBM.....	106
Figure 52 Contours for temperature of horizontal plane passing through the upper burners ( $y=0$ m), EWBM.....	107

Figure 53 Contours for CH <sub>4</sub> mass fraction of vertical plane passing through the middle burners 2 & 5 (x=4.539m), EWBM.....	109
Figure 54 Turbulent viscosity profiles at line (x=3.264, y=-1.0675) along Z-axis for different combustion cases, EWBM.....	110
Figure 55 Contours for CO <sub>2</sub> mass fraction of vertical plane passing through the middle burners 2 & 5 (x=4.539m), EWBM.....	111
Figure 56 Density profiles at line (x=3.264, y=-1.0675) along Z-axis for different combustion cases, EWBM.....	112
Figure 57 Contours of radiation heat flux for air-fuel and oxyfuel combustion at the front wall of the furnace (x=0m), EWBM.....	113
Figure 58 Contours of total heat flux for air-fuel and oxyfuel combustion at the front wall of the furnace (x=0m), EWBM.....	114
Figure 59 Total heat flux profiles at line (x=0, y=-1.0675) along Z-axis for different combustion cases, EWBM.....	115
Figure 60 Incident radiation profiles at line (x=3.264, y=-1.0675) along Z-axis for different combustion cases, EWBM.....	116

## LIST OF ABBREVIATIONS

DO	Discrete ordinates method
RTE	Radiative transfer equation
LBL	Line-by-line model
SNB	Statistical narrow band model
WBM	Wide band model
CFD	Computational fluid dynamics
WSGG	Weighted sum of grey gases model
EWBM	Exponential wide band model
SLW	Spectral line-based weighted-sum-of-grey-gases model
SGG	Simple grey gas model
BA	Block approximation method
BEA	Band energy approximation
HITRAN	High-resolution transmission molecular absorption database
HITEMP	High-temperature molecular spectroscopic database
CCS	Carbon Capture and Storage
DTM	Discrete Transfer Method
MCM	Monte Carlo Method
BEM	Boundary Element Method
ZM	Zone Method
FVM	Finite Volume Method
MFM	Multi-Flux Methods

## THESIS ABSTRACT (ENGLISH)

**NAME:** MOHAMMED AHMAD RAJHI  
**TITLE:** INFLUENCE OF GAS RADIATION MODELS ON OXY-COMBUSTION CHARACTERISTICS PREDICTIONS  
**MAJOR:** MECHANICAL ENGINEERING  
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Determination of thermal radiation energy plays an important role in the Computational Fluid Dynamic (CFD) modeling of combustion devices. For this purpose, many gas radiation models have been developed and used. These models vary in their accuracy and complexity. In this study, CFD modeling of a typical industrial water tube boiler is conducted for three different combustion environments; air-fuel combustion ( $O_2$  @ 21 % Vol and  $N_2$  @79% Vol), OF21 ( $O_2$  @ 21 % Vol and  $CO_2$  @79% Vol) and OF27 ( $O_2$  @ 27 % Vol and  $CO_2$  @73% Vol). Simple grey gas model (SGG), Exponential wide band model (EWBM), Leckner model, Perry model and Weighted sum of grey gases model (WSGG) were examined and their influence on the results were evaluated. Among the models the exponential wide band model (EWBM) is the most accurate and is considered the benchmark model. Finally, the characteristics of oxyfuel combustion are compared to those of air-fuel combustion in this boiler.

MASTER OF SCIENCE DEGREE  
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## THESIS ABSTRACT (ARABIC)

الاسم: محمد أحمد راجحي

العنوان: تأثير نماذج حساب الطاقة الإشعاعية للغاز على التنبؤ بخصائص الإحتراق بالأكسجين  
التخصص: الهندسة الميكانيكية .

التاريخ: مايو 2012 .

تحديد طاقة الإشعاع الحراري يلعب دور مهم في عمل نماذج حساب الموائع الديناميكي ( Computational Fluid Dynamic ) لأجهزة الإحتراق. من أجل ذلك، العديد من نماذج حساب الطاقة الإشعاعية للغاز تم تطويرها وإستخدامها. هذه النماذج تختلف في دقتها وتعقيدها. في هذه الدراسة تم استخدام طريقة حساب الموائع الديناميكي ( Computational Fluid Dynamic ) لعمل نموذج خاص لغلاية (مولد بخار) صناعية نموذجية بإستخدام ثلاث طرق للإحتراق، الطريقة الأولى بإستخدام الهواء ( air-fuel combustion )، الطريقة الثانية بإستخدام الأوكسجين بنسبة 21 % مع ثاني أكسيد الكربون بنسبة 79 % (OF21) والطريقة الثالثة بإستخدام الأوكسجين بنسبة 27 % مع ثاني أكسيد الكربون بنسبة 73 % (OF27). تم إختبار وتقييم عدة نماذج لحساب الطاقة الإشعاعية للغاز وهي : Weighted sum of grey gases ، Perry model ، Leckner model ، Simple grey gas model model (WSGG) و Exponential wide band model (EWBM). النموذج الأخير يعتبر النموذج المعياري في هذه الدراسة. أخيراً خصائص الإحتراق بالأوكسجين تمت مقارنتها بخصائص الإحتراق بالهواء.

درجة الماجستير في العلوم  
جامعة الملك فهد للبترول و المعادن  
الظهران المملكة العربية السعودية

# CHAPTER 1

## INTRODUCTION

### 1.1 RESEARCH BACKGROUND

Global warming is a serious problem in which the combustion of coal, oil and other fossil fuel causes the atmospheric concentrations of greenhouse gases, such as carbon dioxide (CO<sub>2</sub>) to increase. As an inevitable result, the atmospheric temperature will increase causing climate change. Rise of sea levels, changes in the rain-fall patterns (draught in some places and severe flooding in others) are thought to be the consequences of this change. Beside carbon dioxide (CO<sub>2</sub>), combustion of fossil fuel emits many other pollutants e.g. nitrogen dioxide (NO<sub>2</sub>) and nitric oxide (NO). Industrial smog, acid precipitation and lung diseases (asthma) may be the results of the high concentration of these pollutants. Public awareness and legislation have led to a policy of reduction of greenhouse gas emissions in most economically well-developed countries, an international environmental treaty is adopted under the united nation umbrella known as koyoto protocol aimed to reduce the industrialized countries collective emissions of greenhouse gases by 5.2% by the year 2012 compared to the year 1990. Figure 1 and 2 shows the rapid increasing of CO<sub>2</sub> and NO emission from 1978 up to 2010 respectively.

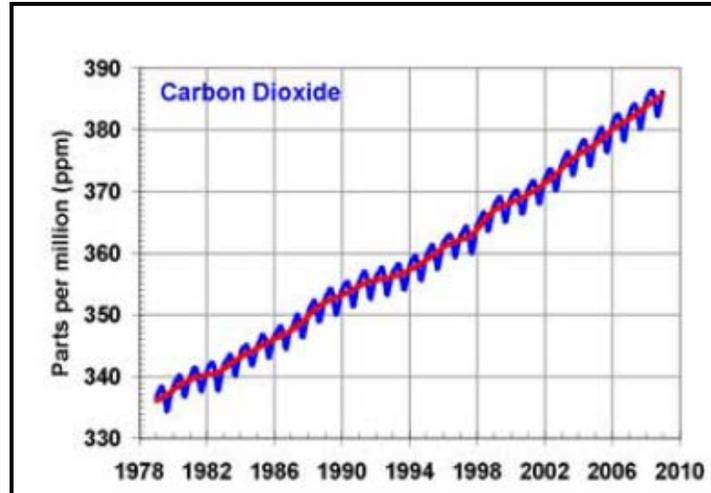


Figure 1 CO<sub>2</sub> emission level [1]

In order to achieve the targeted reduction of greenhouse gases, alternative sources of energy are encouraged to be used widely, nuclear power and renewable energy sources (solar and wind energy) are examples of these alternatives. Renewable energies are carbon neutral and present a favorable solution to the problem of greenhouse gas emissions. Unfortunately renewable energy technologies are currently not mature enough in comparison to fossil fuel based technologies. Much work is required before such energy sources will produce a major portion of our energy. Until these sources become reliable the fossil fuel is used to fill a gap of the increasing demand on energy.

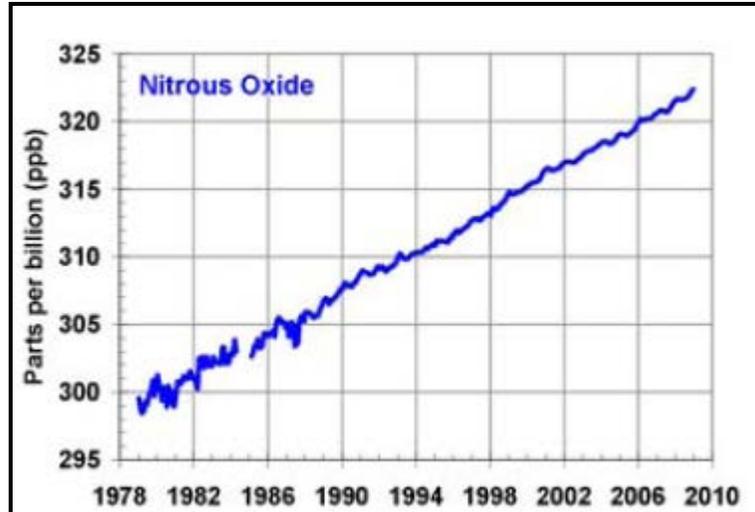


Figure 2 NO emission level [1]

Combustion of fossil fuel is used extensively in power generation process. Reduction of greenhouse gas emission in power plants can be accomplished through one of the following strategies: 1) Improving power plants efficiency, 2) Employ the combined cycle (gas + steam turbines), 3) Use natural gas instead of coal ( having a lower carbon content) and 4) Enhance CO<sub>2</sub> capture and storage technologies.

The first three options may lead to incremental reduction of greenhouse gas emission, to make a step-change reduction in emission, the CO<sub>2</sub> produced by combustion needs to be captured and stored (or sequestered). Over the past decades, because of its wide availability, stability of supply and low cost; coal is considered to be the main source of energy for many countries. However coal emits more carbon dioxide than other fuels and the need for carbon dioxide capture technology is increased. Nowadays, there are three carbon capture technologies (CCS) are available (see Figure 3): 1) Post-combustion capture technology, 2) Pre-combustion capture technology and 3) Oxy-fuel combustion technology.

The development of these technologies has been rapid in recent years, to apply them on large scale basis we still need to focus on many related issues (i.e. cost reduction including O<sub>2</sub> production and CO<sub>2</sub> capture). Each one of the above technologies has its own advantages and limitation.

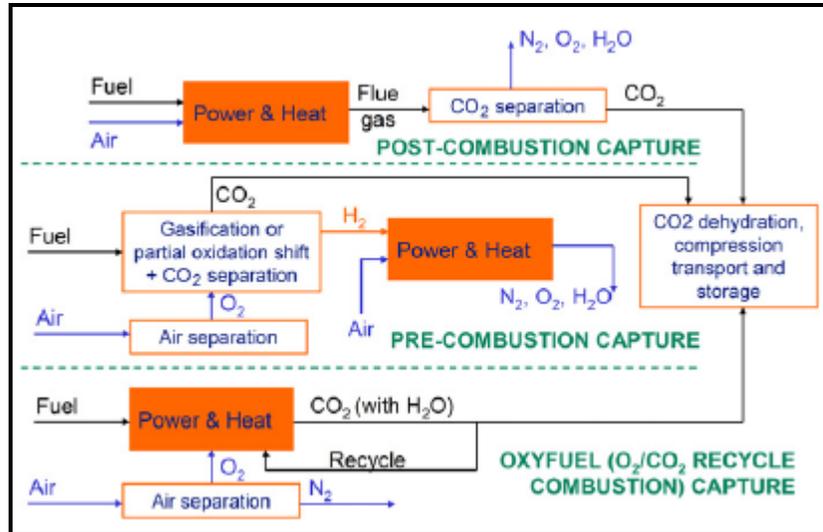


Figure 3 Principles of the three main CO<sub>2</sub> capture options [2].

In this study, the first two technologies will be covered briefly, since it is considered a promising option to capture carbon dioxide (CO<sub>2</sub>), oxy-fuel combustion technology will be given a major part of the study. In addition to its effectiveness for carbon capture, the NO<sub>x</sub> emission can be controlled and reduced by the process. Thermal radiation plays an important role in thermal and combustion engineering; therefore there is a necessity to determine the radiation field and calculate the radiant flux divergence (i.e. radiation sources/sinks) in the thermal energy equation. Several gas radiation models are proposed and used for air combustion case. However, these models need to be checked and validated for oxyfuel combustion. During oxy-fuel combustion, concentrations of CO<sub>2</sub> and H<sub>2</sub>O gases increase compared to air-fired conditions. Unlike N<sub>2</sub>, which is transparent to thermal radiation, both CO<sub>2</sub> and H<sub>2</sub>O emit and absorb thermal radiation. High

concentrations of absorbing/emitting gases increase the emissivity of the flue gas with subsequent effects on the radiative heat transfer.

## **1.2 PROBLEM STATEMENT**

The influence of different gas radiation models on the results of air-fuel and oxyfuel combustion is considered in this study. Results of the CFD modeling of a typical industrial water tube boiler based on different gas radiation models are obtained and evaluated. Fluent package software is utilized to simulate the combustion cases. The study will focus on the following:

1. Discuss the importance of the oxy-fuel combustion as a promising technology to lower greenhouse gases emission especially carbon dioxide  $\text{CO}_2$  and  $\text{NO}_x$ .
2. Review a number of gas radiation models in terms of the range of applicability and accuracy.
3. Simulate the combustion processes under air-fuel and oxyfuel environments with different  $\text{CO}_2/\text{O}_2$  ratios.
4. Make a grid independence survey of the findings by using three different grids.
5. Validation of the results should be warranted through the comparisons with the experimental measurements.

## 1.3 OBJECTIVES

The objectives of the thesis are listed below:

1. Evaluating the influence of gas radiation models on the results of the CFD modeling of air-fuel and oxyfuel combustion.
2. Analyzing the combustion performance of the typical industrial water tube boiler under air-fuel and oxyfuel environments.
3. Investigating the effect of replacing  $N_2$  by  $CO_2$  on the combustion characteristics.
4. Determining the proper  $CO_2/O_2$  ratio that would give the same temperature and heat transfer characteristics as in the air-fuel combustion.
5. Checking the validity and accuracy of different gas radiation models for air-fuel and oxyfuel combustion modeling.
6. Computing the error introduced in the results of oxyfuel combustion when the weighted-sum-of-grey-gases model (WSGG), which developed by Smith for air-fuel combustion, is used in the modeling.
7. Assessing the suitability of retrofitting the boiler to oxyfuel combustion.

## 1.4 THESIS OUTLINE

This thesis contains six chapters.

**Chapter 1** introduces the subject of Carbon capture and Storage and importance of Oxy-fuel combustion to mitigate greenhouse gases specially CO<sub>2</sub>. The problem statement and objectives of the present thesis are discussed.

**Chapter 2** is mainly a literature review and is divided in four sections. The first three sections are about carbon capture technologies. The last section is the literature review on oxyfuel combustion and gas radiation models.

**Chapter 3** is an overview on the gas radiation models. It consists of nine sections; first section is an introduction about gas radiation, second section is the derivation of the radiative transfer equation, third section is about the solution methods of radiative transfer equation, fourth section is the modeling of spectral nature of radiative transfer, and finally the last five sections explain different gas radiation models.

**Chapter 4** is the numerical solution of the combustion problem in the pilot scale furnace under air-fuel and oxyfuel combustion conditions. The validation of radiation models through comparison of the results with the measurements is presented.

**Chapter 5** is the CFD modeling of a typical industrial water tube boiler under air-fuel and oxyfuel combustion. Grid independence of the boiler have been checked and presented. Different gas radiation models have been evaluated and the characteristics of air-fuel and oxyfuel combustions are compared.

**Chapter 6** presents the conclusions of this study and some recommendation for future work.

# CHAPTER 2

## CARBON CAPTURE TECHNOLOGIES

### 2.1 Post-Combustion Capture (PCC) Technology

In post-combustion capture method (see figure 4), a new final processing stage is applied to remove most of the CO<sub>2</sub> from the combustion products just before they are vented to atmosphere. The most commercially advanced methods use wet scrubbing with aqueous amine solutions. Amines-based solvents are suited to the lean combustion CO<sub>2</sub> concentrations of flue gas (12–15 v/v% for coal and 4–8% for gas fired plants), but require a large amount of energy to regenerate the solvent (in the solvent stripper), this being as much as 80% of the total energy of the process. To compensate the energy used in regeneration process, the additional fuel is required. CO<sub>2</sub> is removed from the waste gas by the amine solvent at relatively low temperatures (order 50 °C). The solvent is then regenerated for re-use by heating (to around 120 °C), before being cooled and recycled continuously. The CO<sub>2</sub> removed from the solvent in the regeneration process is then dried, compressed and transported to safe geological storage.

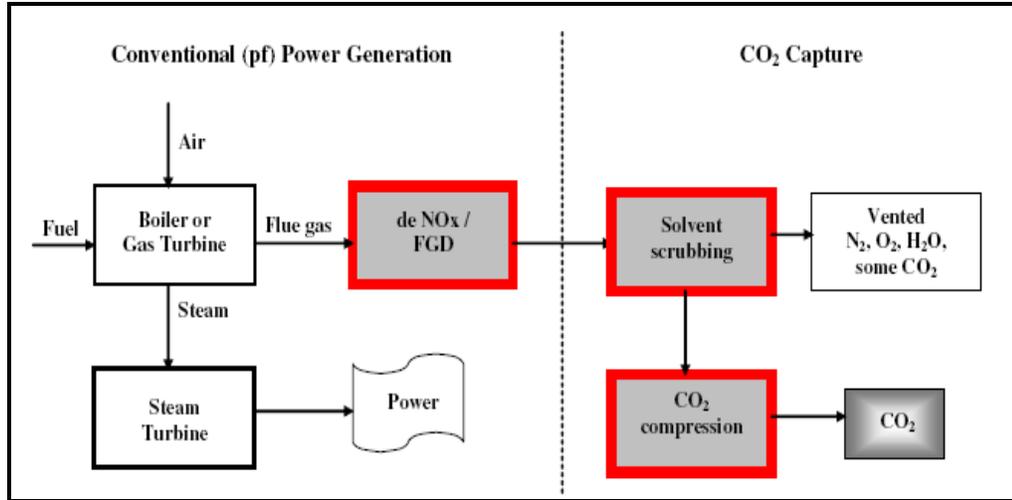


Figure 4 Post-combustion capture (PCC) process [3].

## 2.2 Pre-Combustion Capture Technology

In pre-combustion capture of CO<sub>2</sub> (see figure 5), all types of fossil fuels can, however, be gasified (partially combusted, or reformed) with sub-stoichiometric amounts of oxygen (and usually some steam) at elevated pressures (typically 30–70 atmospheres) to give a 'synthesis gas' mixture of predominantly CO and H<sub>2</sub>. Additional water (steam) is then added and the mixture is passed through a series of catalyst beds for the 'water–gas shift' reaction to approach equilibrium:  $\text{CO} + \text{H}_2\text{O} \leftrightarrow \text{CO}_2 + \text{H}_2$  (adding steam and reducing the temperature promotes CO conversion to CO<sub>2</sub>). The CO<sub>2</sub> can be separated to leave a hydrogen-rich fuel gas. The separation process typically uses a physical solvent such as pressure-swing-absorption, and methanol or polyethylene glycol (with commercial brands called Rectisol and Selexol); CO<sub>2</sub> is dissolved at higher pressure and then released as the pressure is reduced. Because no heat is required to regenerate the solvent and the CO<sub>2</sub> can be released at above-atmospheric pressure, the energy requirements for CO<sub>2</sub> capture and compression in pre-combustion capture systems may be of the order of half

that required post-combustion capture. But pre-combustion capture systems have to pay an efficiency penalty for the shift reaction.

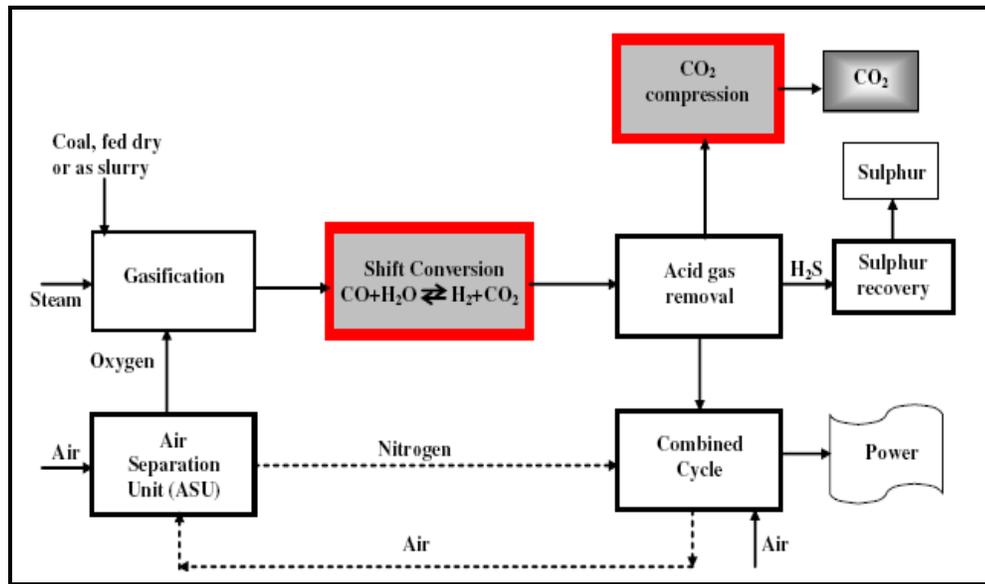


Figure 5 Pre-combustion capture process [3]

### 2.3 Oxy-Fuel Combustion

Oxy-fuel or O<sub>2</sub>/CO<sub>2</sub> recycle combustion is a relatively new Carbon Capture and Storage (CCS) technology and is still at an early development stage compared to either the post- or the pre-combustion CCS technologies. However, oxy-fuel technology promises to be an economical alternative coupled with the possibility of higher CO<sub>2</sub> capture rates. Oxy-fuel combustion is defined as the combustion of fossil fuels, in nearly pure oxygen environment instead of air which is the conventional method employed in steam power plants. Figure 6 illustrates power plant with oxy-fuel technology. Specifically, 95% pure oxygen or higher is fed to the boiler via a cryogenic distillation air separation unit. In addition to oxygen, the major part of the CO<sub>2</sub>-rich

exhaust flue gas is also recycled back to the boiler as a form of diluent, in order to control combustion temperature. This is necessary, since combustion of coal in pure oxygen gives a high flame temperature which will enhance the formation of NO<sub>x</sub> and may cause damage to the boiler. Pulverized coal or other hydrocarbon fuel is then combusted in this mixture of O<sub>2</sub> and CO<sub>2</sub> within the boiler.

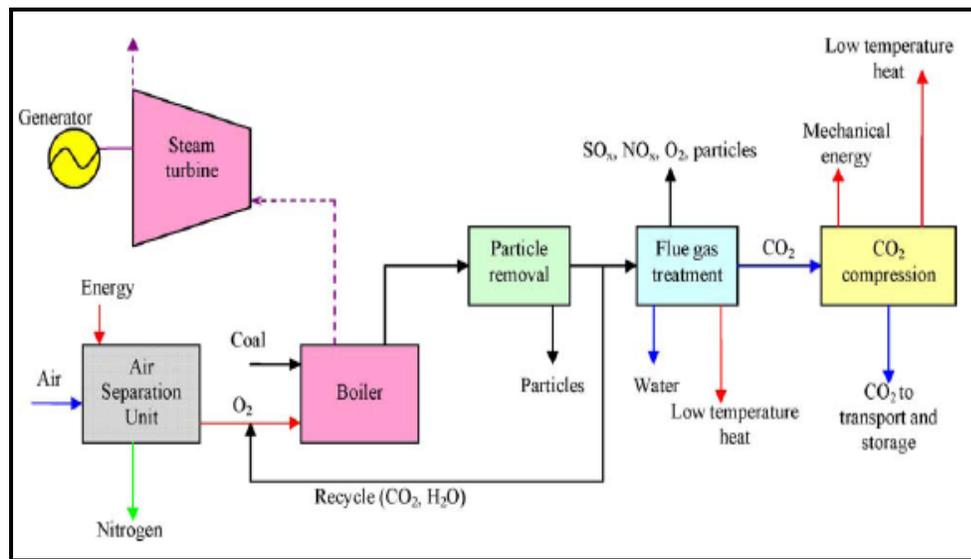


Figure 6 Basic principle of oxyfuel technology [4].

## 2.4 Literature review

Radiative heat transfer is a prevailing phenomenon in many areas, devices and equipment. That is especially the case if the temperatures are very high such as in a combustion furnace. Radiative heat transfer characteristics of the flames, flue gases and the particulate matter are very important to be considered in a boiler design. There is a great necessity for mathematical models and computer codes that solve various radiation problems. The real gases as carbon dioxide, water vapor and their mixtures are mainly

considered within these radiation problems. One of the important issues of the gas radiation is the description of the radiative properties of real gases or of so-called non-grey gases. The models used for defining the radiative properties of combustion gases in radiation calculations can be roughly sorted in three groups [5]:

- (1) Spectral line-by-line models,
- (2) Spectral band models
- (3) Global models.

The complexity of the models decreases from the line model to the global model. Each of these models has its merits and drawbacks and consequently its area of application. Historically, the oldest and the simplest concept for the prediction of radiation in gas is grey gas model [6]. It assumes that the gas absorption coefficient is constant within entire region of wavelengths. This model compared with others gives the predictions of very poor accuracy [7]. The line-by-line (LBL) model provides the best accuracy. In this method radiative transfer equation (RTE) is integrated over detailed molecular spectrum for the gases. Because of the enormous amount of computational requirements, this model is used only for benchmark solutions.

The statistical narrow band model (SNB) is very accurate in the prediction of radiative transfer in high temperature gases. It gives the spectral transmissivity averaged over a narrow band. Because of that, it is difficult to couple statistical narrow band model (SNB) to the solution method of radiative transfer equation such as finite volume method (FVM) where the values of spectral (monochromatic) absorption coefficient or its average over a wavelength interval is needed. Also the disadvantage of statistical narrow band model (SNB) is that it requires a large number of bands, and therefore is very

expensive for computation. Wide band model (WBM) is a simplification of the statistical narrow band model (SNB). Instead of spectral lines, it works with bands and is more economical. Wide band model (WBM) is less, but still reasonable accurate. It yields wide band absorptance while the solution of radiative transfer equation (RTE) by finite volume method (FVM) operates either with spectral or with averaged absorption coefficient. Therefore Wide band model (WBM) cannot easily and simply be incorporated in finite volume method (FVM). In addition, the Wide band model (WBM) requires the knowledge of the path length in the model as well as the spectral parameters associated with this path length.

Line-by-line models (LBL), statistical narrow band model (SNB) and Wide band model (WBM) are very accurate gas radiative properties models, but they require enormous amounts of computer time. This is undesirable even nowadays when powerful supercomputers are available since a gas radiative properties model is only a part of a radiation calculation. A radiation calculation is in turn only a part of a complex computational fluid dynamics (CFD) code that describes fire and combustion comprises 3-D turbulent multiphase fluid flow, chemical reactions and heat transfer by radiation, convection and conduction.

The global gas radiative properties models such as those of Hottel and Sarofim's [6], Steward and Kocafe's [8, 9] and Smith et al.'s [10] are based on the concept of weighted sum of grey gases model (WSGG). The weighted sum of grey gases model (WSGG) was first established by Hottel and Sarofim [6]. It replaces the radiative properties of real gases or of non-grey gases with an equivalent finite number of grey gases. The weighted sum of grey gases model (WSGG) is used for evaluation of total gas

emissivity and total gas absorptivity for a given path length. For each of the grey gases, the absorption coefficient taken as temperature independent, and the weighting factors taken as temperature dependent in polynomial form, should be evaluated. Even weighted sum of grey gases model (WSGG) is established for the evaluation of gas emissivity and absorptivity, when weighted sum of grey gases model (WSGG) is coupled with the solution of radiative transfer equation (RTE) only absorption coefficients and weighting factors of grey gases are used and never the total gas radiative properties.

The radiative transfer were calculated in 2-D, i.e. in axisymmetric cylindrical enclosures [11, 12] by using statistical narrow band model (SNB) with a ray tracing method. Also the statistical narrow band model (SNB) combined with a ray tracing method was used in axisymmetric cylindrical enclosure in [13]. The predictions based on the ray tracing method are extremely expensive and cannot be used to standard engineering calculations due to difficulties related to the linking of the radiative transfer equation (RTE) with narrow band model (SNB) and Wide band model (WBM).

Modest has shown [14] that weighted sum of grey gases model (WSGG) can be generalized for use with any arbitrary solution method of radiative transfer equation (RTE). This method needs evaluation of the grey gas weighting factors and absorption coefficients (for each grey gas) for any used gas species or their mixtures. The Spectral line weighted sum of grey gases model (SLW) [15, 16] appeared as a more accurate version of weighted sum of grey gases model (WSGG). Within the Spectral line weighted sum of grey gases model (SLW) the weights of classical weighted sum of grey gases model (WSGG) were evaluated by using the absorption-line black-body distribution function. This function is calculated directly from the high-resolution

molecular spectrum of gases. Instead of integration of radiative transfer equation (RTE) over wavenumber, now the quadrature is carried out over absorption cross section. Very similar to the Spectral line weighted sum of grey gases model (SLW) is the absorption distribution function (ADF) concept [17, 18]. It differs from the Spectral line weighted sum of grey gases model (SLW) only how the weighting factors are calculated. The weighting factors are evaluated in such a way that emission of an isothermal gas is strictly calculated for actual spectra.

Trivic [19] has developed a 3-D mathematical model for predicting turbulent flow with combustion and radiative heat transfer within a furnace. The model consists of two sections:

- (1) Transport equations that are non-linear partial differential equations solved by a finite difference scheme.
- (2) The radiative heat transfer that was analyzed by the zonal method.

The link between these two sections is a source-sink term in the energy equation. The radiative properties of combustion gases were represented by two different models:

- (1) Hottel and Sarofim's model and,
- (2) Steward's model.

The Monte Carlo method was used to evaluate total radiative interchange in the system between zones. The Mie equations were used for the determination of the radiative properties of particles suspended in the combustion gases. The mathematical model was validated against experimental data collected on two large furnaces:

- (1) A tangentially pulverized coal fired boiler of 220 MW

- (2) An oil fired boiler of 345 MW, with symmetrically positioned burners at the front and rear wall.

The tests and calculations were performed for several loads for both boilers. The measured and predicted gas temperature and heat flux distributions within the boilers were compared. The results gave reasonable agreement between measurements and values predicted by the model.

Coelho [20] has carried out 3-D numerical simulations of radiative heat transfer from non-grey gases by the discrete ordinate method (DO) and the discrete transfer method (DTM). He has used several gas radiative properties models as CK model, the Spectral line weighted sum of grey gases model (SLW) and weighted sum of grey gases model (WSGG). The predictions were compared against the ray tracing statistical narrow band model (SNB) results [21] that were used as benchmark data. Coelho has concluded, inter alia, the following:

- (1) The weighted sum of grey gases model (WSGG) is computationally economical and has moderate accuracy.
- (2) The Spectral line weighted sum of grey gases model (SLW) is a compromise between accuracy and numerical requirements but it needs additional work for an optimization of the coefficients.
- (3) The CK model is the most accurate but too time consuming for engineering applications.

The comparison of the predictions calculated by nine total emissivity models against the calculations done by exponential wide band model (EWBM) was carried out [7]. One of

the main conclusion was that Smith et al. weighted sum of grey gases model [10] indicates the advantage.

Trivic [22] has developed a new 3-D non-grey gas mathematical model and computer code for radiation, based on the coupling of finite volume method (FVM) and weighted sum of grey gases model (WSGG) convenient for incorporation within CFD codes. Since it is showing the advantage compared with others, Smith et al.'s weighted sum of grey gases model (WSGG) was incorporated in this model.

During oxy-fuel combustion, concentrations of CO<sub>2</sub> and H<sub>2</sub>O gases increase compared to air-fired conditions. Unlike N<sub>2</sub>, which is transparent to thermal radiation, both CO<sub>2</sub> and H<sub>2</sub>O emit and absorb thermal radiation. High concentrations of absorbing/emitting gases increase the emissivity of the flue gas with subsequent effects on the radiative heat transfer.

Buhre et al [23] have reviewed the status and the research needed for oxy-fuel technology. The understanding of oxy-fuel combustion has been established primarily from pilot-scale studies, as it has not been applied at practical scale, and there have been few fundamental studies. Buhre stated that the oxy-fuel combustion and CO<sub>2</sub> capture from flue gases is a near-zero emission technology that can be adapted to both new and existing pulverized coal-fired power stations. In the oxy-fuel technology the concentration of carbon dioxide in the flue gas is increased from approximately 17 to 70% by mass. The carbon dioxide can then be captured by cooling and compression for subsequent transportation and storage.

Wall [3] has stated that the heat transfer prediction are critical for oxy-fuel technology, since there are changes in gas properties due to CO<sub>2</sub> recycling from outlet back to the

furnace inlet. These changes are attributed to alteration of gas radiative properties and gas heat capacity. During oxy-fuel combustion, the flue gas tri-atomic molecules concentration increases drastically and this will lead to change in the gas emissivity.

Hottel et al [6] showed that the major contributor of heat transfer from a flame from conventional fuels is thermal radiation from water vapor, carbon dioxide, fly ash, soot and carbon monoxide. Any change of CO<sub>2</sub> and H<sub>2</sub>O concentrations yield to change in radiative heat transfer. For air combustion with conventional partial pressures of CO<sub>2</sub> and H<sub>2</sub>O, the heat transfer calculation carried out by using a 3 grey-one clear gas model to estimate flame emissivity. In oxy-fuel combustion the 3 grey-one clear gas model should be validated and/or modified or replaced by a more accurate model [6, 24], carbon dioxide and water vapor also have high thermal heat capacities compared to nitrogen. Kumar et al [25] have conducted some measurements on coal devolatilisation, they concluded that devolatilisation in an atmosphere of O<sub>2</sub>/CO<sub>2</sub> is greater than in O<sub>2</sub>/N<sub>2</sub> due to char gasification by CO<sub>2</sub>. The elevated CO<sub>2</sub> concentration surrounding the burning char particles could also result in gasification reactions contributing to the char mass loss. Measurements by Kumar et al [25] indicate that coal and char reactivity can differ at the same O<sub>2</sub> levels in O<sub>2</sub>/N<sub>2</sub> and O<sub>2</sub>/CO<sub>2</sub> environments. Even if the reactivities in the two environments are identical, coal burnout will improve in an oxy-fuel retrofit due to:

- The higher O<sub>2</sub> partial pressures experienced by the burning fuel,
- Possible gasification by CO<sub>2</sub>
- Longer residence times due to the lower gas volumetric flows.

Haynes et al [26] discussed the influence of gaseous additives on the formation of soot in diffusion flames. They concluded that addition of CO<sub>2</sub> and H<sub>2</sub>O to the oxidizer, as well as replacement of N<sub>2</sub> by CO<sub>2</sub> in the oxidizer, will reduce soot formation in the flame.

Shaddix et al [27] found that gasification reaction of the char by CO<sub>2</sub> becomes significant under oxygen-enriched char combustion at temperatures prevailing in practical processes.

Gaseous pollutant formation and emissions are changing during oxy-fuel combustion, as a result of sulfur retention in ash and deposits; the SO<sub>x</sub> emissions per energy of fuel combusted may be lowered by less than 20%, depending on ash composition and NO<sub>x</sub> emissions generated per unit of energy are reduced by up to 70% [3]. Kimura et al [28] stated that during oxy-fuel combustion, the amount of NO<sub>x</sub> exhausted from the system can be reduced to less than one-third of that with combustion in air.

According to Okazaki et al [29] the mechanisms lead to NO<sub>x</sub> reduction are:

- Decrease of thermal NO<sub>x</sub> due to the very low concentration of N<sub>2</sub> from air in the combustor.
- The reduction of recycled NO<sub>x</sub> as it is reburnt in the volatile matter release region of the flame.
- The reaction between recycled NO<sub>x</sub> and char.

They concluded by experiments that the reduction of recycled NO<sub>x</sub> is the dominant mechanism for the reduction in NO<sub>x</sub> emissions. They estimated that more than 50% of the recycled NO<sub>x</sub> was reduced when 80% of the flue is recycled.

Croiset et al [30, 31] noted that the conversion of coal sulphur to SO<sub>2</sub> decreased from 91% for the air case to about 64% during oxy-fuel combustion. In the oxy-fuel combustion, SO<sub>2</sub> concentration is higher than that from air combustion due to flue gas recirculation.

Molburg et al [32] stated that there is a need for desulphurization of the recycled flue gas for oxy-fuel combustion to reduce corrosion impact on the furnace.

Khare et al [33] noted that in order to achieve the same O<sub>2</sub> fraction in the flue gas as in air firing, the excess O<sub>2</sub> from 3% to 5% for dry and wet recycle is required for oxy-fuel combustion. Also, oxygen concentrations at the burner inlet range from 25% to 38% (v/v) to achieve matched furnace heat transfer depending on the use of dry or wet recycle with some impact of furnace size. Changes to the plant or its operation may be required to maintain design output by achieving a satisfactory balance for heat transfer in the different sections of the furnace. The balancing of heat transfer appears to depend on the extent of drying of the recycle stream [34].

Preliminary heat transfer calculations for retrofitted boiler have also been performed by the University of Newcastle [35]. The calculations revealed that retrofitting of existing boilers with oxy-fuel technology results in different heat transfer impacts. For the same adiabatic flame temperature, furnace heat transfer increases and convective pass transfer decreases. As the furnace heat transfer is dependent on the furnace size, the impact is scale (i.e. boiler size) dependent.

A study has been done by Payne et al. [36] including experiments in a 3 MW test furnace; they investigated the efficiency of the recycle rate as a tuning parameter with the purpose of establishing similar heat transfer characteristics in oxy-fuel as in air-firing in

retrofits of existing boilers. In addition, a modeling study of a 50 MW furnace has been conducted in the same work; the results showed that similar overall heat transfer (radiative + convective) could be established during both dry and wet recycling of flue gases at certain recycle rates and to get the same adiabatic flame temperature as for air case the oxygen concentrations should be lower than 30%. Finally, they concluded that the recycle rate is an appropriate tuning parameter to achieve similar boiler performance for air and oxy-fuel combustion.

CFD models for oxy-fuel combustion have been published for flame models by Chui and Tan [37] to predict  $\text{NO}_x$  formed in a pilot-scale furnace.

Tan et al. [38] performed natural-gas-fired oxy-fuel experiments in a 300 kW refractory lined combustor. Total heat flux measurements showed that an oxy-fuel flame with 28% vol oxygen and an air flame produced similar total heat flux profiles. The temperature conditions were almost identical for air and oxy-fuel conditions. Zheng et al [39, 40] modeled the heat transfer in a boiler to assess the suitability of retrofitting an air-fired boiler to oxy-fuel combustion. The gas emissivity was calculated from the correlations for total gas emissivity for the water vapor and carbon dioxide suggested by Leckner [41]. The studies indicated that the lower and upper part of an air-fired boiler can be made to perform properly without major modification when converting from air firing to oxy-fuel combustion.

Andersson and Johnsson [42] have studied experimentally the combustion of propane fuel for 3 test cases: air-combustion (referenced one) and other two oxy-combustion cases (OF21 @ 21%  $\text{O}_2$  and OF27 @ 27%  $\text{O}_2$ ). The difference in total radiation intensity between these cases is determined and compared. The mean emissivity is then

determined based on the experimental data and as a comparison the gas emissivity is calculated by an existing gas emissivity model given by Leckner [41]. In addition the difference in combustion environment is described. One of the results they have reported is in case of the OF 21 combustion, the temperature levels were lower compared to air-fired condition and this is returned to two reasons; 1) the higher specific heat capacity of  $\text{CO}_2$  compared to  $\text{N}_2$ , and 2) increase in radiation losses caused by the  $\text{CO}_2$  with increase the gas emissivity. In addition, the physical volume of the high temperature zone of OF21 case is smaller compared to the two other cases. The gas composition of the flame envelope is clearly different from the rest; the amount of unburned hydrocarbons is significantly higher. In spite of the lower temperature levels of the OF21 case compared to the air case, the OF21 radiation intensities are approximately similar to those of air case. Furthermore, the temperature levels in the OF27 combustion approach those of the air case and the radiation intensity is higher than that in the air case despite the fact that the temperature levels are lower or similar to those of air. It is also found that, the flame radiation intensity increases with up to 30% and the incident wall intensity by 20% compared to air fired conditions. Finally, they concluded that the difference between oxyfuel combustion and air combustion can be partly explained by the high  $\text{CO}_2$  content in the combustion gases, i.e.  $\text{CO}_2$  partial pressure approximately eight times the pressure level of normal air-fired condition, and the increase in the total emissivity of oxyfuel combustion flame is attributed to the increased emissivity of the  $\text{CO}_2$ . In general, changes in soot volume,  $\text{CO}_2$  and  $\text{H}_2\text{O}$  concentration levels are very important factors in modeling the radiative heat transfer for  $\text{O}_2/\text{CO}_2$  flames.

In the most recent study, Johnsson et al [43] have evaluated several approximate gas radiative property models for oxy-fuel environments in large atmospheric boiler. These models were Statistical narrow band (SNB) models, weighted-sum-of-grey-gases model (WSGG), spectral line-based weighted-sum-of-grey-gases model (SLW) and two grey-gas approximations model. The parameters of statistical narrow band models (SNB) are taken from three sets of databases:

- 1) Leckner [41, 44]
- 2) Grosshandler (RADCAL) [45]
- 3) Taine et al [46, 47, 48] at EM2C lab

The model uses the last database is considered the benchmark model which all other model are compared to. The weighted-sum-of-grey-gases model (WSGG) coefficients have been fitted to emissivities calculated with the EM2C SNB model in the temperature range of 500-2500 K and path lengths between 0.01 and 60 m and atmospheric pressure. The two grey models; one is based on weighted-sum-of-grey-gases model WSGG (4+1) and the other based on EM2C SNB model are tested.

The accuracy of the models is investigated in a number of cases; (uniform and non-uniform) and (wet and dry flue gas) both in terms of intensity of single path, radiative source term and wall flux for a domain between two infinite plates. Temperature and gas concentration profiles used in the investigation for non-isothermal paths are based on experiments conducted in a test furnace, where air and O<sub>2</sub>/CO<sub>2</sub> combustion of a propane flame were studied [49]. The O<sub>2</sub> concentration in the feed gas for an oxy-fuel flame was 27% and the recycle flue gas was dry, given a mole fraction of CO<sub>2</sub> around 0.8 in the

furnace. Radial profiles of gas temperature and concentration of O<sub>2</sub>, CO and CO<sub>2</sub> and total hydrocarbons were measured at different heights in the furnace. The radiation intensity was recorded with a narrow angle radiometer. It is noted that, for the total intensity of the uniform paths, the discrepancies between the statistical narrow band models (SNB) are below 5%. The Spectral line weighted sum of grey gases model (SLW) and weighted sum of grey gases model (WSGG) (3+1) are in a good agreement with the benchmark model deviating less than 10%. The grey model gives a different shape of the intensity profile, although the intensity at the end of the path is correct. For the total intensity of the non-uniform paths, it is found that, the difference between the statistical narrow band models (SNB) is as high as 25% at the end of the path in the case of wet flue gas and within 10% during dry flue gas. The maximum discrepancy of weighted sum of grey gases model (WSGG) (3+1) is 15% but less than 10% in most regions and the deviation between the spectral line-based weighted-sum-of-grey-gases model (SLW) and the EM2C statistical narrow band model (SNB) is 5-10%. The grey models give both an incorrect shape of the profile and a significant error at the end of the path. The comparison of the incident wall flux between the models is carried out in the study, it is shown that the difference between the statistical narrow band models (SNB) is 30% at the longest path in the wet flue gases case whereas 15% for dry flue gases. The weighted sum of grey gases model (WSGG) (4+1) deviates less than 20 % at recycling of wet flue gas and less than 10% at dry flue gas recycling. The spectral line-based weighted-sum-of-grey-gases model (SLW) with its larger number of gases agrees very well with the EM2C statistical narrow band model (SNB). Finally, in term of radiative source quantity, the difference between the statistical narrow band model (SNB) is about

5-20%. The deviation of the weighted-sum-of-grey-gases model (WSGG) models is with a few exceptions less than 20%.

The accuracy of the radiation models is determined by the importance of the radiative source term. Lallemand et al [7] pointed out that in regions of intensive combustion, the heat released by combustion dominates over the radiative release, and the energy radiated away from non-luminous flames hardly exceeds 10-15% of the total heat release in the flame. In addition, they compared gas radiation models with measurements of total radiation intensity in a 300 kW non-sooting natural-gas flame. The agreement between measurements and computed exponential wide band model (EWBM) data is excellent for measurements against cold furnace walls.

Habib et al [50] have compared the characteristics of oxyfuel combustion to those of air-fuel combustion in a typical natural gas fired package boiler. The percentages of recycled  $\text{CO}_2$  considered in their study were 83.8% and 77% by mass. The first corresponds to 21%  $\text{O}_2$  and the second corresponds to 29%  $\text{O}_2$  by volume. Results indicated that the temperature levels are reduced in oxyfuel combustion. As the percentage of recirculated  $\text{CO}_2$  is increased, the temperature levels are greatly reduced. They found that the fuel and oxygen consumption rates are slower in oxyfuel combustion relative to air-fuel combustion. Furthermore, heat transfer from the burnt gases to the water jacket along the different surfaces of the furnace was calculated. It is shown that the energy absorbed is much higher in the case of air-fuel combustion along all surfaces except for the end part of the furnace close to the furnace rear wall. The heat transfer in the return chamber (tube bank) was also calculated and the results indicated higher heat transfer in the oxyfuel

case in comparison with the air fuel case as a result of ignition delay in the vicinity of the furnace entrance region.

The weighted sum of grey gases model (WSGG) is recommended for CFD applications when the radiative source term has a significant influence and reasonable estimates of wall fluxes are required. The drawback of the model is the need for specific sets of coefficients at different  $H_2O/CO_2$  ratios and most of the available coefficients are fitted to air-fired conditions.

Porter et al [51] have used the spherical harmonic (P1) and the discrete ordinates method (DO) to solve the radiative transfer equation (RTE) in multi-dimensional problem. The spectral nature of radiation has been treated by using non-grey gas full spectrum k-distribution method (FSCK) and a grey method. The simulation results of air-fuel and oxyfuel combustion are compared with the statistical narrow band model (SNB). They concluded that the non-grey full spectrum k-distribution method was in good agreement with the statistical narrow band model (SNB). Significant errors are introduced in the wall heat flux when a grey model is used. In addition, using the smith grey weighted sum of grey gases model (WSGG) leads to substantial errors in the radiation source calculation which changes the predicted temperature.

Weighted sum of grey gases model (WSGG), which was developed by Smith [10], is fitted for air-fuel combustion. This model was modified by Johansson et al [52] to be suitable to cover the temperature range of 500–2500 K, pressure path lengths of 0.01–60 bar m and molar ratios of  $H_2O$  and  $CO_2$  between 0.125 and 2. This range of molar ratios covers both oxy-fuel combustion of coal, with dry- or wet flue gas recycling, as well as

combustion of natural gas. They have reported that the modified weighted sum of grey gases model (WSGG) is a computationally efficient option for CFD simulations.

Hjærtstam et al [53] have utilized the CFD method to investigate the influence of gas and gas-soot radiation mechanisms in air and oxyfuel flames when Propane fuel is burnt in a 100 kW test furnace. In their study, a grey model was examined in order to know if it is sufficient to generate a reliable solution when applied in CFD simulation of oxyfuel combustion. They have shown that in the case of Oxyfuel combustion both grey and non-grey weighted sum of grey gases models (WSGG) overestimate the peak temperature, however, the results of non-grey model are closer to the measurements than that of grey model. In addition, they stated that the inclusion of soot modeling in both grey and non-grey gas radiation models has a significant impact in the CFD calculations.

# CHAPTER 3

## GAS RADIATION MODELS

### 3.1 Introduction

Thermal radiation plays an important role in thermal and combustion engineering. This is especially true for applications involving high-temperature chemically reacting flows where radiation can dominate the total energy transfer in the system. A combustion device is a concrete example where one typically needs to calculate the flame structure and heat transfer in the system in multiple dimensions systems. For such a purpose one must determine the radiation field and calculate the radiant flux divergence (i.e. radiation sources/sinks) in the thermal energy equation [54]. The calculation of the radiant energy quantities is coupled to the transport equations for the momentum, energy and species concentration distributions. Nowadays engineers are confronted with complex combustion phenomena that depend on interrelated processes of thermodynamics, chemical kinetics, fluid mechanics, heat and mass transfer, turbulence and radiative transfer. Computational fluid dynamics (CFD) tools are needed to help understand, design, optimize and operate high-temperature combustion devices where radiative transfer is an important mode of energy transport. Owing to the nature of the radiative transfer equation (RTE) such CFD computations are very time-consuming, because there is a need to integrate the radiation field (intensity) over all directions and the entire

spectrum [19]. Research efforts have been under way for the past several decades to make the computations more efficient and cost effective so that computational fluid dynamics tools could be used to perform design and optimization calculations. During the last several decades, considerable progress has been made in predicting radiative transfer in multidimensional “enclosures” containing absorbing, emitting and scattering media such as mixtures of gases and gases/polydisperse particles [58]. Because the radiative transfer equation (RTE) is similar to the equations used for simulating neutron transport and flow of rarified gases, researchers have often adopted and assimilated the methods developed by scientists in these fields to solve radiation heat transfer problems encountered in thermal/combustion systems [56]. When calculating radiation heat transfer in a combustion device, probably the first and most important decision one has to make is about the choice of the radiative transfer model, which is dictated by the specific application and the level of detail needed to analyze the problem. For example, the level of detail required to determine the local temperature and species concentration distributions in a flame is much greater than that required to calculate the local radiant heat flux on a load in a furnace or on the furnace wall. Next, one must decide if detailed spectral calculations of radiation transport is warranted or some approximate band model or grey treatment of radiative transfer will be adequate. One must also choose a numerical method for solving the radiative transfer equation (RTE) that is compatible with the technique employed for solving the transport equations. Finally, the model should also be accurate, robust and predict the desired radiant energy quantities in a computationally efficient manner [54].

The accuracy of radiative transfer predictions in combustion systems cannot be better than the accuracy of the radiative properties of the combustion products used in the analysis. These products usually consist of combustion gases such as water vapor, carbon dioxide, carbon monoxide, sulfur dioxide, and nitrous oxide, and particles, like soot, fly-ash, pulverized-coal, char or fuel droplets. Before attempting to tackle radiation heat transfer in practical combustion systems, it is necessary to know the radiative properties of the combustion products. Considering the diversity of the products and the probability of having all or some of these in any volume element of the system, it can easily be perceived that the prediction of radiative properties in combustion systems is not an easy task. The wavelength dependence of these properties and uncertainties about the volume fractions and size and shape distribution of particles cause additional complications [6].

### **3.2 Radiative Transfer Equation (RTE)**

Two theories have been developed for the study of the propagation and interaction of electromagnetic radiation with matter, namely, the classical electromagnetic wave theory and the radiative transfer theory. The theories were developed independently and there is no similarity in their basic formulations. Conceptually, they are completely distinct; however, both theories describe the same physical phenomenon. The classical electromagnetic theory has approached the study of propagation and interaction of matter with radiation from the microscopic point of view and the radiative transfer theory from the macroscopic (or phenomenological) point of view. The study of the detailed interaction between electromagnetic radiation and matter on the microscopic level from both the classical and quantum mechanics point of view yields the interaction cross-

sections of the particles making up the matter [57]. This fundamental approach predicts the macroscopic properties of the media, and these properties appear as coefficients in the radiative transfer equation. The quantitative study, on the phenomenological level, of the interaction of radiation with matter that absorbs, emits, and scatters radiant energy is the concern of the radiative transfer theory. The theory ignores the wave nature of radiation and visualizes it in terms of light rays or photons. These are concepts of geometrical optics. The geometrical optics theory is the study of electromagnetism in the limiting case of extremely small wavelengths or of high frequency. The detailed mechanism of the interaction process involving atoms or molecules and the radiation field is not considered. Only the macroscopic problem consisting of the transformation suffered by the field of radiation passing through a medium is examined. Thus, there is a considerable simplification over the electromagnetic wave theory. The radiative transfer equation (RTE) forms the basis for quantitative study of the transfer of radiant energy in a participating medium [24]. The equation is a mathematical statement of the conservation principle applied to a monochromatic pencil (bundle) of radiation and can be derived from many viewpoints. The RTE is based on application of an energy balance on an elementary volume taken along the direction of a pencil of rays and confined within an elementary solid angle. The detailed mechanism of the interaction processes involving particles and the field of radiation is not considered here. On the phenomenological level only the transformation suffered by the radiation field passing through a participating medium is examined. The derivation accounts mathematically for the rate of change of radiation intensity along the path in terms of physical processes of absorption, emission, and scattering.

Consider a cylindrical volume element, figure 7, of cross-section  $dA$  and length  $ds$  in an absorbing, emitting, and scattering medium characterized by the spectral absorption coefficient  $\kappa_\nu$ , scattering coefficient  $\sigma_\nu$  [54]. The axis of the cylinder is in the direction of the unit vector  $\vec{s}$ , i.e.  $ds$  is measured along  $\vec{s}$ . The spectral intensity of radiation (spectral radiance) in the  $\vec{s}$ -direction incident normally on one end of the cylinder is  $I_\nu$  and the intensity of radiation emerging, through the second end in the same direction is  $I_\nu + dI_\nu$ . Here,  $\nu$  is the frequency and is related to the wavelength  $\lambda$  by  $\nu = c / \lambda$ , where  $c$  is the speed of electromagnetic wave in vacuum.

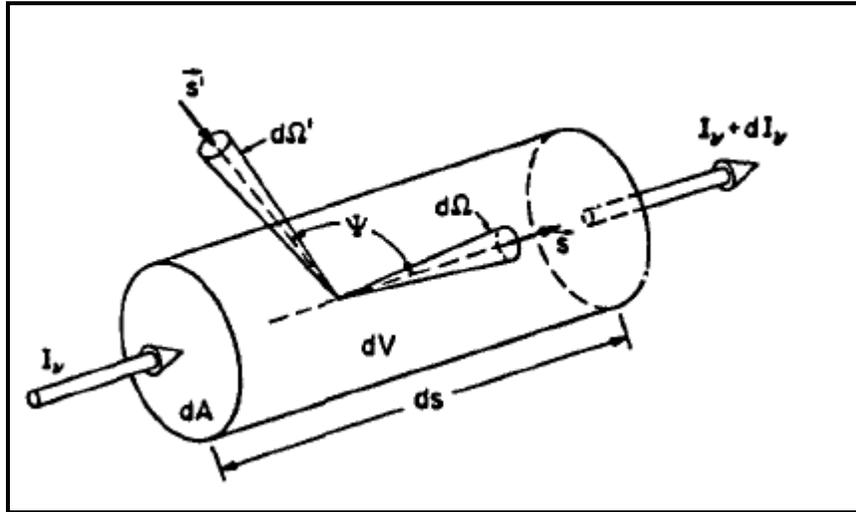


Figure 7 coordinates for derivation of the radiative transfer equation [54].

It follows from the definition of the spectral intensity  $I_\nu$  that radiant energy incident normally on the infinitesimally small cross-section  $dA$  during time interval  $dt$ , in frequency range  $d\nu$  and within the elementary solid angle  $d\Omega$  about the direction of the unit vector  $\vec{s}$  is

$$I_\nu dA d\Omega d\nu dt \tag{1}$$

The emerging radiant energy at the other face of the cylinder in the same direction equals

$$(I_\nu + dI_\nu) dA d\Omega d\nu dt \tag{2}$$

The net gain of radiant energy, i.e. the difference between energy crossing the two faces of the cylinder, is then given by

$$(I_\nu + dI_\nu - I_\nu) dA d\Omega d\nu dt = dI_\nu dA d\Omega d\nu dt \quad (3)$$

The loss of energy from this pencil of rays due to absorption and scattering in the cylinder is

$$(\kappa_\nu + \sigma_\nu) I_\nu ds dA d\Omega d\nu dt \quad (4)$$

The emission by the matter inside the cylindrical volume element  $dV$ , in the time interval  $dt$ , in the frequency range  $d\nu$  confined in the solid angle  $d\Omega$  about the direction  $\vec{s}$  equals

$$\kappa_\nu n_\nu^2 I_{b\nu} dV d\Omega d\nu dt \quad (5)$$

Where  $I_{b\nu}$  is Planck's spectral blackbody intensity of radiation, and  $n_\nu$  is the spectral index of refraction of the medium. The increase in energy of the pencil of rays  $(\vec{s}, d\Omega)$  due to in-scattering of radiation by the matter into the elementary cylindrical volume from all possible directions  $\vec{s}'$  is

$$\sigma_\nu ds \left[ \frac{1}{4\pi} \int_{\Delta\nu} \int_{\Omega'=4\pi} \Phi_\nu(\vec{s}' \rightarrow \vec{s}; \nu' \rightarrow \nu) \times I_{\nu'}(\vec{s}') d\Omega' d\nu' \right] dA d\Omega d\nu dt \quad (6)$$

In this expression the phase function  $\Phi_\nu(\vec{s}' \rightarrow \vec{s}; \nu' \rightarrow \nu) d\Omega' d\nu' / 4\pi$  represents the probability that radiation of frequency  $\nu'$  propagating in the direction  $\vec{s}'$  and confined within the solid angle  $d\Omega'$  is scattered through the angle  $(\vec{s}', \vec{s})$  into the solid angle  $d\Omega$  and the frequency interval  $d\nu$ . This probability is determined by the scattering mechanism. For coherent scattering the phase function is independent of frequency  $\nu'$  and reduces to  $\Phi_\nu(\vec{s}' \rightarrow \vec{s})$ .

Since in this case the sum of probability over all directions must equal unity, we must have

$$\frac{1}{4\pi} \int_{\Omega=4\pi} \Phi_{\nu}(\vec{s} \rightarrow \vec{s}') d\Omega' = \frac{1}{4\pi} \int_{\Omega=4\pi} \Phi_{\nu}(\vec{s}' \rightarrow \vec{s}) d\Omega = \frac{1}{4\pi} \int_{\Omega=4\pi} \Phi_{\nu}(\Psi) d\Omega = 1 \quad (7)$$

This implies that for coherent scattering the spectral phase function is normalized to unity. The scattering angle  $\Psi$ , i.e. the angle between  $\vec{s}'$  and  $\vec{s}$  can be expressed as

$$\cos \Psi = \cos \theta \cos \theta' + \sin \theta \sin \theta' \cos(\phi - \phi') \quad (8)$$

Or

$$\cos \Psi = \xi \xi' + \eta \eta' + \mu \mu' \quad (9)$$

Where

$$\xi = \sin \theta \cos \phi, \quad \eta = \sin \theta \sin \phi, \quad \mu = \cos \theta \quad (10)$$

are the direction cosines in any orthogonal coordinate system. Reference to figure 1,  $\vec{s}'$  ( $\theta'$ ,  $\phi'$ ) represents the incoming direction of the pencil of rays, and  $\vec{s}$  ( $\theta$ ,  $\phi$ ), is the direction of the pencil after scattering.

Equating the change of energy in the cylindrical volume element to the net gain or loss of energy along the traversal path of the cylinder in terms of the processes of attenuation, emission and in-scattering yields [57]

$$dI_{\nu} ds dA d\Omega dv dt = -(\kappa_{\nu} + \sigma_{\nu}) I_{\nu} ds dA d\Omega dv dt + \kappa_{\nu} n_{\nu}^2 I_{b\nu} dV d\Omega dv dt + \sigma_{\nu} ds \left[ \frac{1}{4\pi} \int_{\Delta\nu'} \int_{\Omega'=4\pi} \Phi_{\nu}(\vec{s}' \rightarrow \vec{s}; \nu' \rightarrow \nu) \times I_{\nu'}(\vec{s}') d\Omega' d\nu' \right] dA d\Omega dv dt \quad (11)$$

Dividing this equation by  $ds dA d\Omega dv dt$  and recalling that the distance  $ds$  traversed by the pencil of rays is  $cdt$ , where  $c$  is the velocity of light in the medium, yields the equation of transfer in a Lagrangian coordinate system

$$\frac{1}{c} \frac{dI_{\nu}}{dt} = -(\kappa_{\nu} + \sigma_{\nu}) I_{\nu} + \kappa_{\nu} n_{\nu}^2 I_{b\nu} + \frac{\sigma_{\nu}}{4\pi} \int_{\Delta\nu'} \int_{\Omega'=4\pi} \Phi_{\nu}(\vec{s}' \rightarrow \vec{s}; \nu' \rightarrow \nu) I_{\nu'}(\vec{s}') d\Omega' d\nu' \quad (12)$$

Clearly, the left-hand side of this integro-differential equation represents the net change in  $I_\nu$  per unit length along the path  $ds = c dt$ . Equation (12) is a statement of the conservation of energy principle for a monochromatic pencil of radiation (in the direction  $\vec{s}$ ) and is generally called the "radiative transfer equation" (RTE) [57].

The substantial derivative  $d/dt$  refers to the rate of change of spectral intensity as seen by an observer propagating along with the velocity of radiation (Lagrangian coordinates). In terms of a coordinate system fixed in space (Eulerian coordinates), radiative transfer equation (RTE) may be written as

$$\frac{1}{c} \frac{dI_\nu}{dt} = \frac{1}{c} \frac{\partial I_\nu}{\partial t} + (\nabla \cdot \vec{s}) I_\nu = \beta_\nu (S_\nu - I_\nu) \quad (13)$$

where the source function  $S_\nu$  represents the sum of emitted and in scattered radiation and is defined as radiant energy leaving an element of volume of matter in the direction  $(\vec{s}, d\Omega)$  per unit volume, per unit solid angle, per unit frequency, and per unit time,

$$S_\nu = (\kappa_\nu n_\nu^2 I_{b\nu} / \beta_\nu) + (\sigma_\nu / \beta_\nu) (1/4\pi) \times \int_{\Delta\nu'} \int_{\Omega'=4\pi} \Phi_\nu(\vec{s}' \rightarrow \vec{s}; \nu' \rightarrow \nu) I_\nu(\vec{s}') d\Omega' d\nu' \quad (14)$$

In general the intensity is a function of three spatial coordinates, two angles and time; of course, the seventh independent variable required to define the radiation intensity is the wavelength or the frequency of radiation.

The radiative transfer equation (12) is an integro- differential equation, and because of this it is very difficult to solve exactly for multidimensional geometries. Therefore, some simplifications of this equation are necessary. A close look at the source term given in equation (14) reveals that the in-scattering term (the second term of the right hand side) yields the integral nature of the radiative transfer equation (RTE). If scattering is negligible in the medium, then the equation (14) will be a differential equation, which is

much easier to solve than the integro-differential equation. Calculation of radiative transfer requires two types of models [58]:

- (1) Models to account for directional nature of radiation; and
- (2) Models to describe the spectral nature of radiation.

Since, the directional and spectral models are not coupled or directly related, they can be discussed separately. Numerous approaches for both types of models have been proposed, and they will be divided into several groups for more logical and coherent discussion.

### **3.3 Radiative Transfer Equation (RTE) Solution Methods**

#### **3.3.1 Introduction**

The radiative transfer equation is an integro-differential equation, and its solution even for a one dimensional, planar, grey medium is quite difficult. Most engineering systems, on the other hand, are multidimensional. In addition, spectral variation of the radiative properties must be accounted for in the solution of the radiative transfer equation (RTE) for accurate prediction of radiation heat transfer. These considerations make the problem even more complicated [54]. Therefore, it is almost necessary to introduce some simplifying assumptions for each application before attempting to solve the radiative transfer equation (RTE) in its general form. It is not possible to develop a single general solution method for the equation which would be equally applicable to different systems. Consequently, several different solution methods have been developed over the years.

According to the nature of the physical system, characteristics of the medium, the degree of accuracy required, and the available computer facilities, one of several different methods can be adopted for the solution of the problem considered. Before choosing one solution method over another, it is important to know the advantages and disadvantages of each method. In this section, several radiative transfer models of interest to combustion systems and their features will be discussed.

### **3.3.2 Overview of computational methods in radiative transfer**

During the last four decades numerous methods have been developed. These methods for calculating radiative transfer in multidimensional geometries can be grouped into four broad categories [58];

- (1) Directional averaging approximations;
- (2) Differential approximations (moment, modified moment, spherical harmonics, etc.);
- (3) Energy balance (zone, Monte Carlo, finite volume, finite element, boundary element, etc.) methods; and
- (4) Hybrid (discrete transfer, zone-Monte Carlo, ray tracing, etc.) methods.

The asymptotically thin and thick approximations are simple but unfortunately their ranges of validity are very limited; therefore, only the general optically self-absorbing situations are considered here. All of the directional averaging approximations involve some type of averaging of the radiance field with direction. The two-, four-, six-, and multi-flux methods (MFM) are probably the simplest to apply. The shortcomings of MFMs are primarily twofold [58]. First, the accuracy of the predictions depends on the

arbitrary choice of the solid angle subdivision over which the intensities are integrated to obtain the fluxes. Second, the fluxes for one direction are not coupled with those of another direction if the medium is not scattering.

The first order differential approximations of the radiative transfer equation (RTE) (moment and spherical harmonics) are capable of treating radiatively participating media with scattering. They yield radiative transfer predictions of reasonable accuracy and can be adopted for band calculations. The discrete ordinates method (DO) for approximating the radiative transfer equation (RTE) and the finite volume method (FVM) for solving the radiative transfer equation (RTE) numerically are compatible with CFD methods for solving transport problems in combustion systems.

Energy balance methods such as the zone method (ZM) and its extensions as well as the boundary element method (BEM), which can be considered as a variant of the zone method (ZM), require exceedingly large storage for the local exchange factors if the methods are to be used to calculate the local quantities. The method is suitable for calculating heat fluxes or global heat transfer rates in a furnace if the temperatures and radiating species concentrations are known. The disadvantage of the zone method (ZM) and boundary element method (BEM) solvers is the difficulty of generating fine meshes in regions of large temperature and species concentration gradients (i.e. flame structure) as well as the difficulty in evaluating integrals whose integrands are singular. The statistical Monte Carlo method (MCM) is incompatible with CFD procedures for solving the transport equations.

The discrete transfer method (DTM) is considered a hybrid method because it is derived from the combination of flux methods and the Monte Carlo method (MCM) for choosing

the finite number of directions of the radiative transfer equation (RTE). The method is well suited for CFD calculations. The disadvantages of the other hybrid methods are the same as those of the energy balance methods.

### **3.3.3 Discrete Ordinate Method (DO)**

The discrete ordinates method (DO) is also in a way a multi-flux method (MFM), but is considered separately because of its importance. The method does not require any assumptions concerning the directional dependence of the intensity. It was originally suggested by Chandrasekhar [59] for astrophysical applications, and detailed derivation of the relevant model equations were discussed in application to neutron transport problems. The method has been applied to solve radiation heat transfer problems including those arising in combustion systems. Currently, there are two types of radiative transfer equation (RTE) used in discrete ordinates method (DO). One is the conventional approach which approximates the radiative transfer equation (RTE) by a system of first-order ordinary differential equations and the other is the even parity formulation (EPF-DO) which is a second-order partial differential equation. The approximation is obtained by discretizing the entire solid angle ( $\Omega = 4 \pi$ ) using a finite number of ordinate directions and corresponding weight factors. The radiative transfer equation (RTE) is written for each ordinate, and the integral terms are replaced by a quadrature summed over each ordinate. A simpler but more accurate version of this method obtained later using Gaussian or Lobatto quadrature is called the  $S_n$ -approximation. The approximation was obtained by dividing the spherical space into  $N$  equal solid angles or  $N(N + 2)$  angular subdomains, where  $N$  is determined by the order of quadrature scheme used. For axisymmetric systems the number of angular subdivisions required is reduced by a factor

to be uniform within each of these angular domains whose extent is determined by the weights of the quadrature scheme employed. The  $S_n$  approximation denotes the discrete ordinates in that there are  $N$  discrete values of direction cosines  $(\xi_n, \eta_n, \mu_n)$  which always satisfy the identity

$$\xi_n^2 + \eta_n^2 + \mu_n^2 = 1 \quad (15)$$

As a concrete example, the radiative transfer equation (RTE) for a 3-D enclosure in Cartesian ordinates is written for each quadrature point  $n$  as:

$$\xi_m \frac{\partial I_m}{\partial x} + \eta_m \frac{\partial I_m}{\partial y} + \mu_m \frac{\partial I_m}{\partial z} = -(\kappa + \sigma)I_m + \kappa I_b(T) + \frac{\sigma}{4\pi} \sum_{m'=1}^N w_{m'} \Phi(m', m) I_{m'} \quad (16)$$

Where  $\mu_m, \eta_m$  and  $\xi_m$  represent direction cosines for the discrete angular direction  $S_m$ ;  $w_{m'}$  is the angular (Gaussian) weights associated with the  $S_{m'}$  direction, and the summation is over the total number of discrete angular directions,  $N$ , used in the approximation. Appropriate boundary conditions need to be imposed to complete the mathematical statement of the problem. If a surface bounding the medium, say, at  $x = 0$  emits and reflects radiation diffusely, then the boundary condition for equation (16) is given by:

$$I_m = \varepsilon I_b + \left( \frac{\rho}{\pi} \right) \sum_{m'=1}^M w_{m'} |\mu_m| I_{m'}; \quad \mu_m > 0 \quad (17)$$

where  $\varepsilon$  is the emissivity and  $\rho$  reflectivity of the surface. Similar expressions can be written for the other five surfaces in the 3-D enclosure.

## 3.4 Modeling of spectral nature of radiative transfer

### 3.4.1 Introduction

Every combustion process produces combustion gases, such as water vapor, carbon dioxide, carbon monoxide, and others. The partial pressures of these gases in the combustion products are determined by the type of the fuel used and the conditions of the combustion environment, such as fuel-air ratio, total pressure and ambient temperature [24]. These gases do not scatter radiation significantly, but they are strong selective absorbers and emitters of radiant energy. Consequently, the variation of the radiative properties with the electromagnetic spectrum must be accounted for. Spectral calculations are performed by dividing the entire wavelength (or frequency) spectrum into several bands and assuming that the absorption/emission characteristics of each species remain either uniform or change smoothly in a given functional form over these bands. As one might expect, the accuracy of the predictions increases as the width of these bands becomes narrower. Exact results, however, can be obtained only with line-by-line calculations which require the analysis of each discrete absorption-emission line produced as a result of the transitions between quantized energy levels of gas molecules. Line-by-line calculations are not practical for most engineering purposes but are usually required for the study of radiative transfer in the atmosphere [55].

Prediction of radiative transfer in combustion systems requires modeling of the radiation characteristics of combustion products: gases ( $\text{CO}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{CO}$ ,  $\text{NO}_2$ ,  $\text{CH}_4$ , etc.), particles (soot, liquid fuel droplets, pulverized coal, char, fly-ash, dust, etc.) and their mixtures. Use of the global (total) gas and gas plus particle emittance as well as mean-beam-length

concepts may be adequate for radiant heat transfer but not for temperature and flame structure calculations in flames and combustion chambers [6]. In the order of decreasing detail the models are: line-by-line (LBL), narrow band (NB) and wide band (WB). The line-by-line (LBL) models are the most accurate approaches for calculation radiative transfer when based on precise spectroscopic data bases such as HITRAN or HITEMP [54]. They yield accurate results, but are computationally too expensive for engineering application of radiative transfer in combustion systems since the entire spectrum consists of millions of single lines. In combustion applications global (integrated over spectrum) radiative transfer calculations in 3-D geometries, which of necessity are based on spatial, directional and spectral discretization, require the use of approximate models. Narrow band (NB) and wide band (WB) models cannot be applied to multidimensional radiative transfer problems which require solution of the differential radiative transfer equation (RTE), where the knowledge of the spectral absorption coefficient or its average over a spectral interval is required [19]. The narrow band (NB) model gives the spectral transmittance over a NB, and the wide band (WB) model yields the WB absorptance. These models can be used only with the integral representation of the radiative transfer equation (RTE). Unfortunately; these approaches are prohibitively expensive for practical applications. For this reason it is necessary to turn to more efficient methods to solve the radiative transfer equation (RTE) based on the differential formulation. The total emittance and absorptance concepts are useful for one-dimensional radiative transfer analyses or for zonal predictions of global radiant energy quantities. In addition, the emittance / absorptance concepts cannot be readily adopted to combustion systems containing mixtures of radiatively participating gases and absorbing and scattering

particles [54]. For multidimensional radiative transfer calculation absorption and scattering coefficients as well as single scattering albedo are needed. Unfortunately, the mean absorption coefficients calculated from total emittances / absorptances depend on temperature, path length, pressure and ratio of CO<sub>2</sub> and H<sub>2</sub>O partial pressures. A detailed assessment of the effective absorption coefficients calculated from different total emittance correlations has been reported by Lallemand et al [7], but only in the optically thin and thick limiting cases can the mean absorption coefficients be defined uniquely and treated as properties (i.e. functions of composition, pressure and temperature).

### **3.4.2 Narrow-band models**

Narrow-band models are constructed from spectral absorption-emission lines of molecular gases by postulating a line shape and an arrangement of lines. The shape (profile) of spectral lines is quite important as it yields information for the effect of pressure, temperature, optical path length, and intrinsic properties of radiating gas on the absorption and emission characteristics. The Lorentz profile is the most commonly used line shape to describe gases at moderate temperatures under the conditions of the local thermodynamic equilibrium [55], and it is also known as a collision-broadened line profile. If the temperature is high and the pressure is low, the Doppler line profile would be more appropriate to use. If there are ionized gases and plasmas in the medium and they are influenced by interactions between the radiating particles and surrounding charged particles, then the Stark profile yields a more accurate representation of the spectral line radiation. Note that it is also possible to superpose these line profiles to incorporate the effects of different physical conditions on the line radiation.

There are basically two different line arrangements for narrow band models used extensively in the literature. The Elsasser or regular model assumes that the lines are of uniform intensity and are equally spaced. The Goody or statistical model postulates a random exponential line intensity distribution and a random line position selected from a uniform probability distribution [24]. For practical engineering calculations both of these models yield reasonably accurate results. Usually there is less than 8% discrepancy between the predictions of these two models. Narrow-band model predictions generally require an extensive and detailed library of input data, and the calculations cannot be performed with reasonable computational effort. On the other hand, as long as the concentration distributions of gaseous species are not accurately known the high accuracy obtained for the spectral radiative gas properties from narrow band models would not necessarily increase the accuracy of radiation heat transfer predictions. Also, it is not always convenient to use detailed, complex models for the spectral radiative gas properties.

### **3.4.3 Wide-band models**

Since gaseous radiation is not continuous but is concentrated in spectral bands, it is possible to define wide-band absorptivity and/or emissivity models. The radiation absorption characteristics for each band of any gas can be obtained from experiments and then empirical relations can be fitted to those data. The profile of the band absorption may be box or triangular shaped or an exponentially decaying function can be used by curve fitting. These types of empirical models are known as wide-band models, and among them the exponential wide band model (EWBM) is commonly used. For an isothermal medium, several approximate expressions for the total band absorptivity and

emissivity as well as the reviews of the wide-band models are available in the literature [60]. The spectral band absorptivity distributions from a narrow band model are compared with those from a wide-band model for two isothermal media. It is found that, the wide band model is in good agreement with the narrow band model, especially for 2.7  $\mu\text{m}$  H<sub>2</sub>O and CO<sub>2</sub> bands, 4.3  $\mu\text{m}$  CO<sub>2</sub> band, and 6.3  $\mu\text{m}$  H<sub>2</sub>O band. In addition, the relative importance of short wavelength band radiation (i.e. from 1.38  $\mu\text{m}$  and 1.89  $\mu\text{m}$  H<sub>2</sub>O bands) becomes larger as the temperature of the medium increases. The error introduced by approximating the short wavelength band absorption by wide-band models is marginal, since the temperature of a typical combustion chamber is usually not as high as 2000 K, and the other gas bands absorb radiation more strongly than short wavelength bands.

It should be mentioned that some of the detailed spectral properties of the combustion gases will be suppressed when they are combined with those of the particles. Because of this, use of very accurate spectral properties of gases may not increase the accuracy of radiative transfer predictions.

#### **3.4.4 Total absorptivity-emissivity models**

A detailed modeling of the radiative properties of combustion gases may not be warranted for the accuracy of total heat transfer predictions in combustion chambers, but definitely increase the computational effort. For engineering calculations it is always desirable to have some reliable yet simple models for predicting the radiative properties of the gases. Here, we review some of the available models. One way of obtaining radiative properties easily is to use Hottel's charts which are presented as functions of

temperature, pressure and concentration of a gas [6]. Some scaling rules for the total absorptivity and emissivity of combustion gases can be used to extend the range of applicability of Hottel's charts. Of course, in order to use these charts in computer models, curve-fitted correlations are desirable. Other sources for continuous expressions are the narrow and wide band models. The spectral or band absorptivities from these models are first integrated over the entire spectrum for a given temperature and pressure to obtain total absorptivity and emissivity curves. Afterwards, appropriate polynomials are curve-fitted to these families of curves at different temperatures and pressures using regression techniques. Sometimes, these curve-fitted expressions can be so arranged that the resulting expressions would be presented as the sum of total emissivity or absorptivity of clear and grey gases [6]. These are known as the "weighted sum-of-grey-gases" models. There are several curve-fitted expressions available in the literature for use in computer codes. Some of them are given in terms of polynomials and the others are expressed in terms of the weighted sum-of-grey gases. All of these models are restricted to the total pressure of one atmosphere, except that of Leckner [41], and all of them are for the gas radiation along a homogeneous path, i.e. uniform temperature and/or uniform pressure. If the path is inhomogeneous then the equivalent line model or the total transmittance nonhomogeneous method can be used to predict radiation transmitted along the path [45]. However, in multidimensional geometries or if scattering particles are present in the system, the use of these models for practical calculations becomes prohibitive as the equations are much more complicated.

### 3.4.5 Absorption and emission coefficients

The total absorptivities and emissivities are useful for zero or one-dimensional radiative transfer analyses as well as zonal methods for radiative transfer. However, for differential models of radiative transfer the absorption and emission coefficients are required rather than the total absorptivities and emissivities [19]. Since scattering is not important for combustion of gases (and soot particles), the grey absorption / emission coefficient can be obtained from the Bouguer's or Beer-Lambert's law [54]. For a given mean beam length  $L_m$  one can write

$$\bar{\kappa} = (-1/L_m) \ln(1 - \varepsilon) \quad (18)$$

The mean absorption coefficients obtained from spectral calculations as well as curve-fitted continuous correlations were compared with measurements from a smoky ceiling layer formed in a room fire and very good agreement was found [9]. It is possible to determine the so called "grey" absorption and emission coefficients for each temperature, pressure, and path-length, which yield approximately the same total absorptivity or emissivity of the CO<sub>2</sub>-H<sub>2</sub>O mixture. In order to determine the absorption and emission coefficients from total absorptivity-emissivity data, the corresponding mean beam length must be properly evaluated. The definition of the mean beam length for a volume of a gas radiating to its entire surface is given as [24]

$$L_m = 4 C V/A \quad (19)$$

where C is the correction factor and for an arbitrary geometry, its magnitude is 0.9. In general, the absorption and emission coefficients are functions of the medium temperature, pressure and gas concentrations. If these coefficients are obtained from the

total emissivity and absorptivity models, they will also be functions of the mean beam length. Therefore, if the total emissivity of a gas volume is fixed, then the corresponding absorption coefficient decreases with increasing physical path length or pressure. The use of absorption/emission coefficients related to the mean beam length is convenient for the scaling of radiation heat transfer in practical systems. As the characteristic dimension of the enclosure decreases the gas becomes thinner, and eventually in the limit of optically thin gas the mean absorption coefficient becomes identical to the Planck's mean absorption coefficient. With an increasing size of the enclosure, the gas becomes optically thicker and the mean absorption coefficient approaches Rosseland's mean absorption coefficient.

### 3.5 Simple grey gas model (SGG)

The simple grey gas model (SGG) approximation assumes that the radiant absorption and emission by gas molecules are independent of the frequency of the radiation regardless of the strong dependency on frequency found in reality. Consequently the radiative properties of a gas mixture are represented by a single parameter, the effective absorption coefficient [61]. The advantages of this model are its simplicity and minimal resultant execution times and it therefore has commonly been used in generic CFD programs for modeling radiative transfer. Under the grey gas assumption the radiative transfer equation (RTE) for the radiation intensity (integrated over the entire spectrum) in three-dimensional Cartesian coordinates, neglecting scattering of radiation, is

$$\xi \frac{\partial I}{\partial x} + \eta \frac{\partial I}{\partial y} + \mu \frac{\partial I}{\partial z} = -\kappa_e I + \kappa_e I_b \quad (20)$$

In the full modeling of a gas-fired furnace, the radiative transfer equation (RTE) is solved for known temperature field and species concentration that are determined by fundamental equations of motion, energy and mass of species.

An optimal value of the simple grey gas model (SGG) effective absorption coefficient for a given situation can be found by trial and error by comparing the numerical results obtained using the simple grey gas model (SGG) with experimental data or with more accurate numerical results based on a more rigorous treatment of gas radiation [61]. The optimal value may be defined as that gives agreement on the total heat flux to sinks when the gas temperature field is specified. However, it will be found that predictions with the optimal coefficient do not, in general, accurately match the distribution of heat flux density over the sinks, or the temperature distribution over the refractories, or the temperature distribution in the gas. Moreover, the optimal value for any one situation is not generally optimal for another.

The general method for making a realistic best estimate of the effective absorption coefficient from the known properties of emission by the gas is based on interpretation of the total emissivity of the gas upon the bounding surface. The estimate involves use of a mean beam length  $L_m$  and a characteristic gas temperature. The characteristic temperature can be the volume-average gas temperature  $T_m$ , in which case

$$\kappa_e = -(1/L_m) \ln[1 - \varepsilon_g(T_m, L_m)] \quad (21)$$

Or the local temperature,  $T$ , giving a locally varying value,

$$\kappa_e = -(1/L_m) \ln[1 - \varepsilon_g(T, L_m)] \quad (22)$$

The mean beam length is estimated as

$$L_m = 3.6V / S \quad (23)$$

Where  $V$  represents volume of furnace and  $S$  is the wall surface area.

### 3.6 Exponential wide band model (EWBM)

The exponential wide band model was developed by Edwards in 1976 [60]. It is based on a physical analysis of gas absorption. It provides a set of semi-empirical expressions to predict the total band absorptance of infrared active molecules. It is believed that this model can be used to predict radiative properties in the temperature range  $T=300$  to 2500-3000 K, total pressure range  $P_t = 0.5$ -20 atm, volumetric fraction  $X_i = 0$  - 1 and path length range  $L = 10^{-4}$  - 100 m. In this model, it is assumed that the total absorption of a vibration-rotation band can be approximated with correlations dependent on three parameters: the integrated band intensity,  $\alpha$ , the band width parameter,  $\omega$ , and the mean line-width-to-spacing parameter,  $\beta$ . These parameters depend on temperature, but pressure effects are also accounted for through the equivalent broadening pressure,  $P_e$ . The parameters yield asymptotic relations for the total band absorptance  $A_k$ . These relations are known as the four-region expression, which is defined as

$$\text{Linear region if } (\tau_H \leq 1 \text{ and } \tau_H \leq \eta): A_k = \omega \tau_H; \tau_k = 0.9$$

$$\text{Square root region if } (\eta \leq \tau_H \leq 1/\eta \text{ and } \eta \leq 1): A_k = \omega[(4 \eta \tau_H)^{1/2} - \eta];$$

$$\tau_k = 0.5[1 + \omega \eta / A_k]$$

$$\text{Log- root region if } (\tau_H > 1/\eta \text{ and } \eta \leq 1): A_k = \omega[\ln(\eta \tau_H) + 2 - \eta]; \tau_k = \omega / A_k$$

$$\text{Logarithmic region if } (\tau_H > 1 \text{ and } \eta > 1): A_k = \omega[\ln(\tau_H) + 1]; \tau_k = \omega / A_k$$

The expressions of the band transmissivity,  $\tau_k$ , are derived from the relation

$$\tau_k = \frac{\tau_H}{A_k} \frac{\partial A_k}{\partial \tau_H} \quad (24)$$

with an upper limit of the transmissivity equal to 0.9, as recommended by Edwards [60].

The optical depth at the band head  $\tau_H$  and the pressure correction parameter  $\eta$  are calculated from

$$\tau_H = \frac{\alpha X}{\omega} \quad (25)$$

$$\eta = \beta P_e \quad (26)$$

where  $X$  is the density path-length. The parameters  $\alpha$  and  $\beta$  are calculated according to the simplified relations presented by Lallemand and Weber [62], while  $\omega$  is given by a correlation dependent on temperature, as presented by Edwards [60]. In order to calculate the total emissivity of H<sub>2</sub>O-CO<sub>2</sub> mixtures, band energy approximation (BEA) is used. In this method, it is assumed that the blackbody emissive power is constant over each absorption band. Lallemand [62] concluded that the band energy approximation (BEA) method is computationally more attractive than the block approximation (BA) method for implementation in CFD computer codes and in most cases the maximum discrepancy between emissivity calculation using the two methods is insignificant. The total emissivity is then given by equation (27)

$$\varepsilon_g \cong \sum_{k=1}^N \frac{\bar{E}_{\nu_{c,k}}^o}{\sigma \cdot T^4} \cdot A_k - \Delta \varepsilon_{c+w} \quad (27)$$

Where  $\bar{E}_{\nu_{c,k}}^o$  is the blackbody emissive power, the correction term,  $\Delta \varepsilon_{c+w}$ , accounts for the overlap between the 2.7  $\mu\text{m}$  and 15  $\mu\text{m}$  bands for mixtures of CO<sub>2</sub> and H<sub>2</sub>O. The

simplified procedure proposed by Modak [63] is used in the present study to calculate this overlapping and is given by the following expression:

If  $(p_c + p_w) l \geq 0.1$  atm-m

$$\Delta \varepsilon_{cw} = \left\{ \frac{\zeta}{10.7 + 101\zeta} - \frac{\zeta^{10.4}}{111.7} \right\} \times \left\{ \log_{10} [101.3(p_c + p_w)l] \right\}^{2.76} \cdot F(T) \quad (28)$$

If  $(p_c + p_w) l < 0.1$  atm-m

$$\Delta \varepsilon_{cw} = 0$$

Where  $\zeta$  and  $F(T)$  are defined by equations (29) and (30), respectively.

$$\zeta = \frac{p_w}{p_w + p_c} \quad (29)$$

$$F(T) = -1.0204 \times 10^{-6} T^2 + 2.2449 \times 10^{-3} T - 0.23469 \quad (30)$$

### 3.7 Leckner model

The model is developed by Leckner in 1972 [41, 44]. It includes the radiation from H<sub>2</sub>O and CO<sub>2</sub>. The CO<sub>2</sub> and H<sub>2</sub>O emissivities are treated separately after which the total emissivity of the mixture is determined by taking a total band overlap correction term into consideration. The advantage of this model is that it can be applied for any arbitrary partial pressures of CO<sub>2</sub> and H<sub>2</sub>O. The polynomial correlations of the model produce a maximum error of 5% compared to the spectral data for water vapor H<sub>2</sub>O and carbon dioxide CO<sub>2</sub> emissivities at a temperature above 400 K. The total emissivity for a mixture of CO<sub>2</sub> and H<sub>2</sub>O is calculated from equation (31):

$$\varepsilon_g = \varepsilon_c + \varepsilon_w - \Delta \varepsilon_{cw} \quad (31)$$

In the model a zero partial pressure emissivity for either CO<sub>2</sub> or H<sub>2</sub>O is given by

$$\ln \varepsilon_o = a_o + \sum_{i=1}^M a_i \lambda^i \quad (32)$$

In which

$$a_i = C_{oi} + \sum_{j=1}^N C_{ji} \tau^j \quad (33)$$

and

$$\lambda = \log_{10}(p_i L) \quad (34)$$

where  $p_i L$  is the pressure path length of either H<sub>2</sub>O or CO<sub>2</sub> (given in bar.cm) and  $\tau = T/1000$  (the gas temperature  $T$  is given in K). The coefficients  $C_{ji}$  are listed in Table 1 and Table 2. To account for different pressure conditions, a pressure correction term is introduced, which is a function of the total pressure of the mixture and the temperature, path length and partial pressure of the gas. The pressure correction term is determined using the relation

$$\left[ \frac{\varepsilon_i}{\varepsilon_o} - 1 \right] / \left[ \frac{\varepsilon_i}{\varepsilon_o} - 1 \right]_{\max} = \exp\{-\xi(\lambda_{\max} - \lambda)^2\} \quad (35)$$

Here,  $\varepsilon_i$  is the emissivity of either CO<sub>2</sub> or H<sub>2</sub>O. At high pressures the emissivity follows an asymptotic behavior according to

$$\left( \frac{\varepsilon_i}{\varepsilon_o} \right)_{\max} = \frac{A \cdot P_E + B}{P_E + A + B - 1} \quad (36)$$

Table 1 Coefficients to calculate the H<sub>2</sub>O emissivity with M=3 and N=3 [41]

<i>i</i>	$C_{i1}$	$C_{i2}$	$C_{i3}$
1	-2.2118	-1.1987	0.035596
2	0.85667	0.93048	-0.14391
3	-0.10838	-0.17156	0.045915

Table 2 Coefficients to calculate the CO<sub>2</sub> emissivity with M=4 and N=5 [41]

<i>i</i>	$C_{i1}$	$C_{i2}$	$C_{i3}$	$C_{i4}$	$C_{i5}$
1	-3.9781	2.7353	-1.9882	0.31054	0.015719
2	1.9326	-3.5932	3.7247	-1.4535	0.20132
3	-0.35366	0.61766	-0.84207	0.39859	-0.063356
4	-0.080181	0.31466	-0.19973	0.046532	-0.0033086

The parameters  $\zeta$ ,  $\lambda_{max}$ ,  $A$  and  $B$  are given in Table 3. The effective pressure  $P_E$  is defined in equations (37) and (38) for H<sub>2</sub>O and CO<sub>2</sub> respectively.

$$P_E = P_T \left( 1 + 4.9 \frac{P_w}{P_T} \sqrt{273/T} \right) \quad (37)$$

$$P_E = P_T \left( 1 + 0.28 \frac{P_c}{P_T} \right) \quad (38)$$

Table 3 Pressure correction parameters used in equation (35) and equation (36) [41]

	CO <sub>2</sub>	H <sub>2</sub> O
$\zeta$	1.47	0.5
$\lambda_{max}$	$\log_{10}(0.225\tau^2)$ ; T > 700 K $\log_{10}(0.054\tau^{-2})$ ; T < 700 K	$\log_{10}(13.2\tau^2)$
$A$	$0.1\tau^{-1.45}+1$	$1.888-2.053\log_{10}(\tau)$ $\tau=0.75$ if T < 750 K
$B$	0.23	$1.1\tau^{-1.4}$

The total emissivity of the gas mixture of H<sub>2</sub>O and CO<sub>2</sub> for the desired partial and total pressures is equal to the sum of emissivities of the two gases minus a correction term  $\Delta\varepsilon_{cw}$  due to overlap in some spectral regions. The overlap between the two gases is approximated by equation (39):

$$\Delta\varepsilon_{cw} = \left\{ \frac{\zeta}{10.7 + 101\zeta} - 0.0089\zeta^{10.4} \right\} \cdot \lambda^{2.76} \quad (39)$$

In which the partial pressure ratio  $\zeta$  is given by

$$\zeta = \frac{P_w}{P_w + P_c} \quad (40)$$

and the parameter  $\lambda$  is defined as

$$\lambda = \log_{10}((p_w + p_c) \cdot L) \quad (41)$$

### 3.8 Perry model

Empirical correlations for the emissivities of water vapor, carbon dioxide, and four mixtures of the two gases are developed in the 8th edition of the Perry's Chemical Engineering Handbook [64]. They are valid for pressure-path-length of range 0.005-10 m-atm. The emissivities at three temperatures of 1000, 1500 and 2000 K at six values of partial pressure ratios of H<sub>2</sub>O/CO<sub>2</sub>, namely, 0, 0.5, 1.0, 2.0, 3.0 and  $\infty$  can be calculated by equation (42). Empirical constants for different partial pressure ratios of H<sub>2</sub>O/CO<sub>2</sub> are listed in Table 4. These correlations were developed based on the data in Hottel emissivity charts [6] and were adjusted to the more recent data from RADCAL [65]. To obtain the emissivity at different temperature, linear interpolation or extrapolation for the

emissivities determined at 1000, 1500 and 2000 K is used and given by equation (43).

The constants available at H<sub>2</sub>O/CO<sub>2</sub> partial pressure ratios of 0.5, 1.0, 2.0 and 3.0 were used to cover the ranges 0.25-0.67, 0.67-1.5, 1.5-2.33 and 2.33-4 respectively as recommended in the handbook. In this study, constants at partial pressure ratios of 0 and ∞ were used to cover the ranges of 0-0.25 and 4-∞ respectively.

$$\log_{10} \varepsilon_g T_g = a_0 + a_1 \log pL + a_2 \log^2 pL + a_3 \log^3 pL \quad (42)$$

$$\varepsilon_g T_g = \frac{\varepsilon_g T_H (T_g - T_L) + \varepsilon_g T_L (T_H - T_g)}{500} \quad (43)$$

Where T<sub>H</sub> and T<sub>L</sub> are the higher and lower temperature, respectively.

Table 4 Empirical constants used in equation (42) [64]

$p_w/p_c$	T (K)	$a_0$	$a_1$	$a_2$	$a_3$
0	1000	2.2661	0.1742	-0.039	0.0040
	1500	2.3954	0.2203	-0.0433	0.00562
	2000	2.4104	0.2602	-0.0651	-0.00155
0.5	1000	2.5754	0.2792	-0.0648	0.0017
	1500	2.6451	0.3418	-0.0685	-0.0043
	2000	2.6504	0.4279	-0.0674	-0.0120
1	1000	2.6090	0.2799	-0.0745	-0.0006
	1500	2.6862	0.3450	-0.0816	-0.0039
	2000	2.7029	0.4440	-0.0859	-0.0135
2	1000	2.6367	0.2723	-0.0804	0.0030
	1500	2.7178	0.3386	-0.0990	-0.0030
	2000	2.7482	0.4464	-0.1086	-0.0139
3	1000	2.6432	0.2715	-0.0816	0.0052
	1500	2.7257	0.3355	-0.0981	0.0045
	2000	2.7592	0.4372	-0.1122	-0.0065
∞	1000	2.5995	0.3015	-0.0961	0.0119
	1500	2.7083	0.3969	-0.1309	0.00123
	2000	2.7709	0.5099	-0.1646	-0.0165

### 3.9 Weighted sum of grey gas model (WSGG)

There are several gas radiative properties models, called global models; they are based on the concept of weighted sum of grey gases. They are Hottel and Sarofim's gas radiative properties model [6], Steward and Kocafe's model [8, 9], Smith et al weighted sum of grey gases model (WSGG) [10] etc. The equation (44) is used to evaluate total emissivity in terms of the weighted sum of grey gases and it is useful especially for the zonal method of analysis of radiative transfer.

$$\varepsilon = \sum_{i=0}^I a_{\varepsilon,i}(T) \left[ 1 - e^{-\kappa_i PL} \right] \quad (44)$$

Where  $a_{\varepsilon,i}$  is the emissivity weighting factors for the grey gas  $i$ . The weighting factors are dependent on gas temperature  $T$ . The expression in the brackets in equation (44) is the  $i$ th grey gas emissivity, having absorption coefficient  $\kappa_i$ , partial pressure  $P$  and the thickness of gas layer, or path length  $L$ . For a gas mixture  $P$  is the sum of partial pressures of absorbing gases. The physical interpretation of the weighting factor  $a_{\varepsilon,i}$  is the fractional amount of black-body energy in the regions of spectrum where grey gas having absorption coefficient  $\kappa_i$  exists. A value of zero is assigned to absorption coefficient for  $i = 0$ . This gas is called clear gas. It accounts for windows in the spectrum between the spectral regions that have absorption. The total emissivity is an increasing function of partial pressure–path length product approaching unity in the limit. Therefore the weighting factors must sum to unity and also must have positive values. The weighting factor for clear gas, i.e. for  $i = 0$ , is defined as

$$a_{\varepsilon,0}(T) = 1 - \sum_{i=1}^I a_{\varepsilon,i} \quad (45)$$

So that,  $i$  values of the weighting factors should be evaluated. A very common and convenient representation of the temperature dependency of the weighting factors is a polynomial of order  $J-1$  given as

$$a_{\varepsilon,i}(T) = \sum_{j=1}^J b_{\varepsilon,i,j} T^{j-1} \quad (46)$$

Here  $b_{\varepsilon,i,j}$  is referred to as the emissivity gas temperature polynomial coefficients. The absorption coefficients  $\kappa_i$  and the polynomial coefficients  $b_{\varepsilon,i,j}$  are obtained by fitting equation (44) to a table of total emissivities previously calculated from the exponential wide band model (EWBM). (The total emissivities for various gas mixtures can be obtained by different means: by experimental measurements, by calculation based on spectral lines or bands based on Hottel's charts etc.). Having  $i$  grey gases and  $J-1$  polynomial order,  $i$  ( $J-1$ ) coefficients should be evaluated. For total absorptivity, the irradiation temperature of surfaces surrounding the gas,  $T_s$ , must be also taken into account. So the expression for total absorptivity is

$$\alpha = \sum_{i=1}^I a_{\alpha,i}(T, T_s) [1 - e^{-\kappa_i PL}] \quad (47)$$

Here the absorptivity weighting factors,  $a_{\alpha,i}$ , are the functions of gas temperature,  $T$ , and of surface irradiation temperature,  $T_s$ , as well. The weighting factors must be positive and the sum of all weighting factors must be equal to unity. The value of weighting factor for  $i = 0$  is evaluated as

$$a_{\alpha,0} = 1 - \sum_{i=1}^I a_{\alpha,i} \quad (48)$$

The dependency of the weighting factors on gas and irradiation temperatures is expressed by polynomials of orders  $J-1$  and  $K-1$ , respectively as

$$a_{\alpha,i} = \sum_{j=1}^J \left[ \sum_{k=1}^K c_{\alpha,i,j,k} T_s^{k-1} \right] T^{j-1} \quad (49)$$

where  $c_{\alpha,i,j,k}$  are the absorptivity polynomial coefficients. The absorption coefficients,  $\kappa_i$ , for total emissivity and absorptivity are taken to be identical in order to reduce the computational efforts. But the orders of polynomials for emissivity and absorptivity may differ. The number of coefficients needed for total absorptivity is equal to  $I * J * K$ . One issue should be stressed here. Irradiation temperature  $T_s$  plays important role for the evaluation of total gas absorptivity, or in this model for the evaluation of absorptivity weighting factors.

Usually the use of two or three grey gas components and a clear gas is sufficient to accurately represent the total emissivity and absorptivity of a CO<sub>2</sub>/H<sub>2</sub>O/clear gas mixture over a wide range of temperature and partial pressure path length product. For a CO<sub>2</sub>/H<sub>2</sub>O/clear gas mixture with  $P_{co2}=0.1P$ ,  $P_{H2O}=0.2 P$ , and  $P=101.3kPa$ , weighting factors have been reported by Smith et al [10] and Truelove [66] among others. Smith et al employed third order polynomials for both  $a_{\varepsilon,i}(T)$  and  $a_{\alpha,i}(T,T_s)$ , i.e.,

$$a_{\varepsilon,i}(T) = \sum_{j=1}^4 b_{\varepsilon,i,j} T^{j-1} \quad (50)$$

$$a_{\alpha,i} = \sum_{j=1}^4 \left[ \sum_{k=1}^4 c_{\alpha,i,j,k} T_s^{k-1} \right] T^{j-1} \quad (51)$$

The absorption coefficients  $\kappa_i$  and the polynomial coefficients  $b_{\varepsilon,i,j}$ , for mixture of CO<sub>2</sub> and H<sub>2</sub>O with  $P_w/P_c = 1$  and  $P_w/P_c = 2$  shown in Table 5. They are valid at total pressure of one atmosphere, pressure path length of 0.001-10 atm-m and temperature range of 600 to 2400 K.

Table 5 Coefficients for the WSGG model [10]

$i$	$k_i$	$b_{\epsilon,i,1} * 10$	$b_{\epsilon,i,2} * 10^4$	$b_{\epsilon,i,3} * 10^7$	$b_{\epsilon,i,4} * 10^{11}$
Mixture, $P_w/P_c = 1$					
1	0.4303	5.15	-2.303	0.9779	-1.494
2	7.055	0.7749	3.399	-2.297	3.77
3	178.1	1.907	-1.824	0.5608	-0.5122
Mixture, $P_w/P_c = 2$					
1	0.4201	6.508	-5.551	3.029	-5.353
2	6.516	-0.2504	6.112	-3.882	6.528
3	131.9	2.718	-3.118	1.221	-1.612

All the weighted sum of grey gases model (WSGG) parameters, i.e. weighting factors and absorption coefficients, provided by Smith et al [10] and others are intended for air-combustion conditions, and their validity to be used in oxy-fuel combustion is limited. Pressure path-lengths and ratios of H<sub>2</sub>O to CO<sub>2</sub> in oxy-fuel combustion are drastically changed from that of conventional combustion. Therefore, there is a need to derive new parameters to be used in these models for oxyfuel combustion conditions. Johansson et al [43] have developed two weighted sum of grey gases models (WSGG), the first consists of three grey gases and one clear gas, and the second is of four grey gases and one clear gas. In each model, parameters have been fitted to oxy-fired conditions with wet ( $Y_{H_2O}/Y_{CO_2} = 1$ ) and dry flue-gas recycling ( $Y_{H_2O}/Y_{CO_2} = 0.125$ ). The absorption coefficients are constant for each gas, while the weights are calculated from temperature-dependent relations normalized to  $T_{ref} = 1200$  K,

$$a_j = \sum_i c_{j,i} (T/T_{ref})^{i-1} \quad (52)$$

The coefficients  $c_{j,i}$ , listed in Table 6 and Table 7, have been fitted to emissivities calculated with the EM2C SNB model in the temperature range of 500–2500 K and path lengths between 0.01 and 60 m with a total pressure of almost 1 bar.

Table 6 Coefficients for the WSGG (4+1) model [43]

$j$	1	2	3	4
$Y_{H_2O}/Y_{CO_2} = 0.125$ (dry)				
$\kappa_j$	0.0408	0.4217	5.201	122.48
$c_1$	0.2719	0.3677	0.2324	0.1058
$c_2$	0.0896	-0.1284	-0.1214	-0.0602
$c_3$	-0.0327	-0.003	0.017	0.008
$Y_{H_2O}/Y_{CO_2} = 1$ (wet)				
$\kappa_j$	0.0668	0.6818	5.9261	86.014
$c_1$	0.2887	0.133	0.361	0.1843
$c_2$	-0.076	0.303	-0.2112	-0.1545
$c_3$	0.0604	-0.1426	0.0332	0.0347

Table 7 Coefficients for the WSGG (3+1) model [43]

$j$	1	2	3
$Y_{H_2O}/Y_{CO_2} = 0.125$ (dry)			
$\kappa_j$	0.0992	2.6589	88.1078
$c_1$	0.4995	0.3418	0.1273
$c_2$	-0.017	-0.1701	-0.0726
$c_3$	-0.0393	0.0196	0.0101
$Y_{H_2O}/Y_{CO_2} = 1$ (wet)			
$\kappa_j$	0.1281	2.4256	59.557
$c_1$	0.2621	0.3898	0.2603
$c_2$	0.1798	-0.0232	-0.2193
$c_3$	-0.0491	-0.0523	0.0502

# CHAPTER 4

## GAS RADIATION MODELS VALIDATION FOR COMBUSTION PROBLEM

### 4.1 Introduction

The scope of this chapter is to check the validity of some approximate gas radiative properties models for CFD modeling of combustion in a pilot scale test furnace. The gas radiation models were evaluated against the experimental measurements and their influence on the results was investigated. Computational Fluid Dynamic (CFD) method is used to simulate three different combustion environments (i.e. air combustion, oxyfuel combustion with 21% O<sub>2</sub> (OF21) and oxyfuel combustion with 27% O<sub>2</sub> (OF27)) based on different gas radiation models. Propane fuel (C<sub>3</sub>H<sub>8</sub>) is combusted in the Chalmers 100 kW O<sub>2</sub>/CO<sub>2</sub> combustion test unit. The models of radiation that were examined are exponential wide band model (EWBM), weighted sum of grey gases model (WSGG) and simple grey gas model (SGG). The validation of results is obtained through comparison with Andersson experimental measurements [42]. Furthermore, some data of experiment such as temperature profile and gas concentration profile (i.e. CO<sub>2</sub> and H<sub>2</sub>O) are used to calculate radiation intensities and gas emissivities at different distance from the burner. Then, these computed intensities were evaluated against the measured intensities across

the furnace. The discrepancies in the findings and their probably source of errors were discussed and highlighted.

## 4.2 Problem description

Andersson [42, 67] have conducted an experiment on the 100 kW oxy-fuel combustion test unit that connected to the Chalmers 12 MW research boiler. The combustor is down-fired with a cylindrical refractory lined reactor which measures 0.8 m in inner diameter and 2.4 m in inner height. Figure 8 illustrates the schematic of the combustion test unit used in the study. Figure 9 depicts the top-mounted burner which consists of a fuel lance ( $D_i = 34$  mm) and primary and secondary feed-gas registers. The primary register ( $D_o = 52$  mm) is swirled with a fin angle of  $45^\circ$ , whereas the secondary register ( $D_o = 92$  mm) has a more moderate swirl with a fin angle of  $15^\circ$ . In order to reduce the time for the combustor to reach thermal steady state conditions, cooling tubes with variable length and an outer diameter of 50 mm are inserted from the top of the combustion chamber at four positions close to the wall, equally distanced from the flame. The ports along the side of the combustor, R1 to R7 are situated at distances 46 to 1400 mm from the burner inlet. These are used for the combustor profile measurements of gas concentration, temperature and radiation intensity profiles. The combustor temperature measurements were performed with water-cooled suction pyrometers. A narrow angle radiometer is used for the radiation measurements. Gas composition data consisting of  $O_2$ , CO,  $CO_2$  and total hydrocarbon (HC) concentrations were measured by a gas suction probe in the probe ports, at the furnace exit and in the recycle loop.

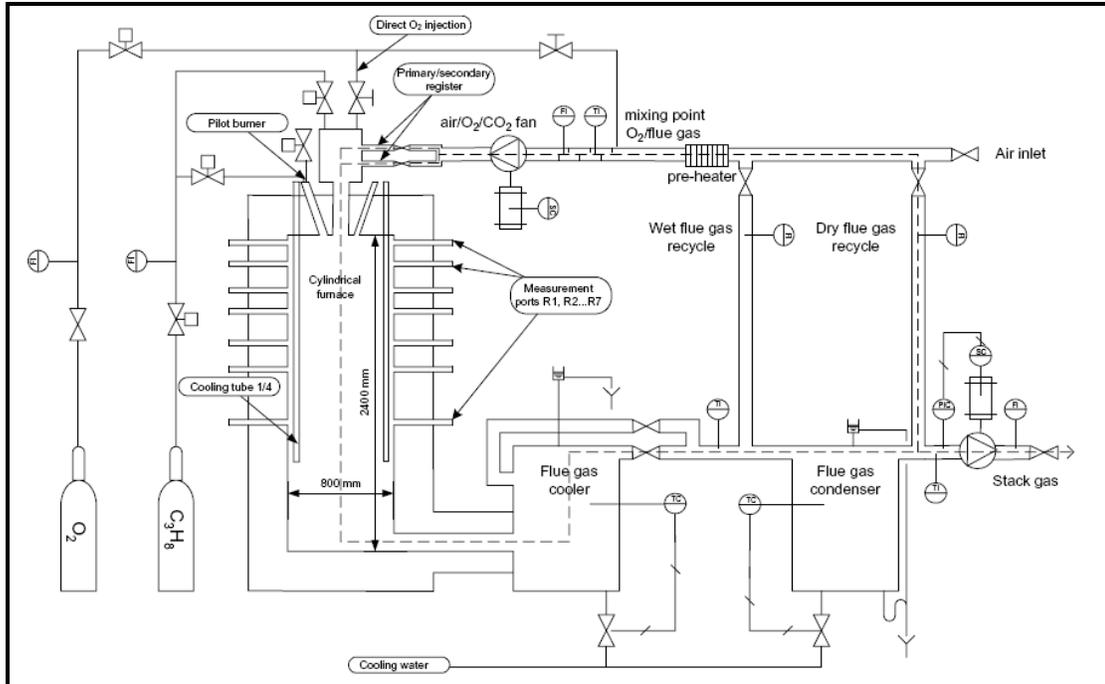


Figure 8 The Chalmers 100 kW O<sub>2</sub>/CO<sub>2</sub> combustion test unit [42].

There are three test conditions are considered in the work, named as:

- Air-firing ( O<sub>2</sub> @ 21 % Vol and N<sub>2</sub> @79% Vol )
- OF21 ( O<sub>2</sub> @ 21 % Vol and CO<sub>2</sub> @79% Vol )
- OF27 ( O<sub>2</sub> @ 27 % Vol and CO<sub>2</sub> @73% Vol )

All tests were performed for propane gas fuel C<sub>3</sub>H<sub>8</sub> combusted at a stoichiometric ratio of 1.15 (15% excessive air). A feed gas temperature is between 25-30 °C. Heat input is adjusted to be 80 Kwth. The fuel mass flow rate through the fuel lance is 1.74 g/s. The volume flow rate of 37m<sup>3</sup>/h and 54m<sup>3</sup>/h are fed through primary and secondary air lances, respectively. The total heat absorption is 48.7 kW in the air case whereas 45.8 kW for OF21.

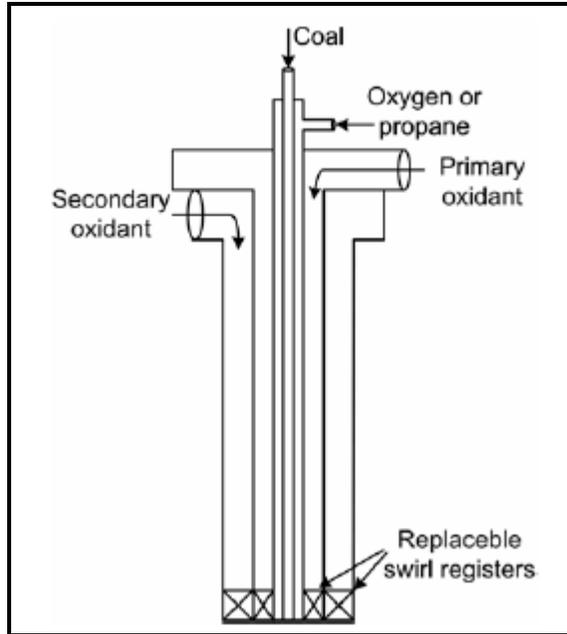


Figure 9 fuel burner design [42]

### 4.3 Numerical simulation

The set of governing differential equations, i.e. continuity, momentum, energy and conservation of a chemical species, together with the boundary conditions were solved numerically by an iterative procedure utilizing the CFD Fluent package software [68]. The finite volume method is used for representing and evaluating partial differential equations in the form of algebraic equations. The CFD calculations normally provide detailed results including velocity, temperature, species concentrations and heat flux that are not easily obtained through experimental measurements. In this study, a 2-D down-firing cylindrical furnace is created for the purpose of simulation. The inner diameter and length are 0.8 m and 2.4 m respectively. The furnace is discretized into quadrilateral cells and the grids were concentrated in the regions of high flow gradients in order to minimize the errors due to false diffusion. All parameters are set up according to the

experiment input data. Total heat absorption is achieved by exposing a part of the furnace wall of 1.4 m length to a negative heat flux. Since the furnace is symmetrical, only half of the cylinder will be considered to reduce the computational time. The solution was converged when the summation of the residual in the governing equation at all the domain nodes was less than 0.01%. The models that used in the solution are

- Incompressible ideal gas assumption
- Standard  $k$ - $\varepsilon$  turbulent model
- Temperature-dependent properties
- One step volumetric reaction for species transport
- Finite-rate/Eddy-dissipation turbulence-chemistry interaction
- Discrete Ordinate Radiation Model (DO)

Since the scope of the study focuses on the radiation models, different radiation models are used along with Discrete Ordinate method (DO) in the solution of the radiative transfer equation (RTE) which is given by

$$\xi \frac{\partial I}{\partial x} + \eta \frac{\partial I}{\partial y} + \mu \frac{\partial I}{\partial z} = -\kappa_e I + \kappa_e I_b \quad (53)$$

The solution of radiative transfer equation (RTE) can not be obtained unless the radiative properties such as absorption coefficients are known. Gas absorption coefficients can be determined from the computed emissivity at each computational cell by employing the Beer-Lambert's law and taking the mean beam length of the geometry as the path length. As previously discussed in chapter 3, gas radiation is a function of frequency and to overcome this obstacle many approximate model were proposed. The simplest one is simple grey gas model (SGG). In this model a single value is assumed for absorption coefficient  $\kappa$  in the (RTE) and therefore gas radiation is independent of frequency. In

addition, the wavelength dependency can be eliminated with an appropriate choice of a total property model. The weighted-sum-of-grey-gases model (WSGG) domain-based is a reasonable compromise between the oversimplified grey gas model and a complete model which takes into account particular absorption bands. Smith et al. [6] weighted-sum-of-grey-gases model (WSGG) is one of the default total radiative property models available in Fluent. Here the emissivity of a gas mixture is represented in terms of temperature independent absorption coefficients and temperature dependent emissivity weighting factors for three grey and one clear gas. These coefficients were fit to the emissivities obtained from exponential wide band model calculations (EWBM). Coefficients are presented for H<sub>2</sub>O/CO<sub>2</sub> ratios of 2 and 1. Unfortunately, this model is intended for air combustion only and can not be applied for oxy-fuel combustion scenarios. So that, user defined functions (UDFs) should be compiled in oxy-combustion solution. Utilizing the new coefficients developed by Andersson [43], two UDFs are coded and built. First one is using three grey one clear weighted-sum-of-grey-gases models and the other is four grey one clear weighted-sum-of-grey-gases models. Furthermore, the exponential wide band model (EWBM), which was explained in the previous chapter, was implemented as a UDF. It is employed in this study to provide a benchmark solution to evaluate the performances of the simple grey gas model (SGG) and weighted-sum-of-grey-gases models (WSGG).

#### **4.4 Results and Discussion**

In order to validate the computational procedure including turbulence, combustion and radiation models, the present computational results were compared with experimental

data of Andersson et al [42]. Three gas radiative properties models, i.e. the exponential wide band model (EWBM), weighted sum of grey gases model (WSGG) and simple grey gas model (SGG), are evaluated against experimental measurements. Figure 10 illustrates radial temperature distribution at distance 0.384m from the burner. It is found that the temperatures are over-estimated by all models. The maximum discrepancy is about 20%. The effective absorption coefficient of 0.4 (1/m) predicts higher temperature by 17% compared to the measurements whereas the absorption coefficient of 0.2 (1/m) is in good agreement with exponential wide band model (EWBM) except in the region near the center of the furnace. The difference between the exponential wide band model (EWBM) and the weighted sum of grey gases model (WSGG) is less than 1%. It is worth noting that the weighted-sum-of-grey-gases model (WSGG) introduced by Smith [6] is fitted to exponential wide band model (EWBM) to find the model's coefficients. At distance 1.4m from the burner, temperature profile is shown in figure 12; it appears that the computed results are over-predicted by 18% in case of the exponential wide band model (EWBM) compared to the measurements. The weighted sum of grey gases model (WSGG) and the simple grey gas model (SGG) of absorption coefficient equal to 0.4 (1/m) are in good agreement and both lead to 15% error in the temperature results. It is found that the absorption coefficient of 0.31 (1/m) reflects the same behavior of the exponential wide band model (EWBM). The error of 27% is noted when the absorption coefficient is 0.2 (1/m). As shown in figure 11, the discrepancies in temperature did not exceed 9% in case of the exponential wide band model (EWBM) and the weighted sum of grey gases model (WSGG). The result of the simple grey gas model (SGG) with absorption coefficient of 0.4 (1/m) is the most accurate compared to the measurement, the error less than 5%. In

addition, CO<sub>2</sub> concentrations are considered in the validation, they are under-predicted in numerical solution. However, the difference is about 3% as shown in figure 13. In general, temperature, velocity and species concentration can be predicted by using of numerical models with error of about 20% in the air combustion process.

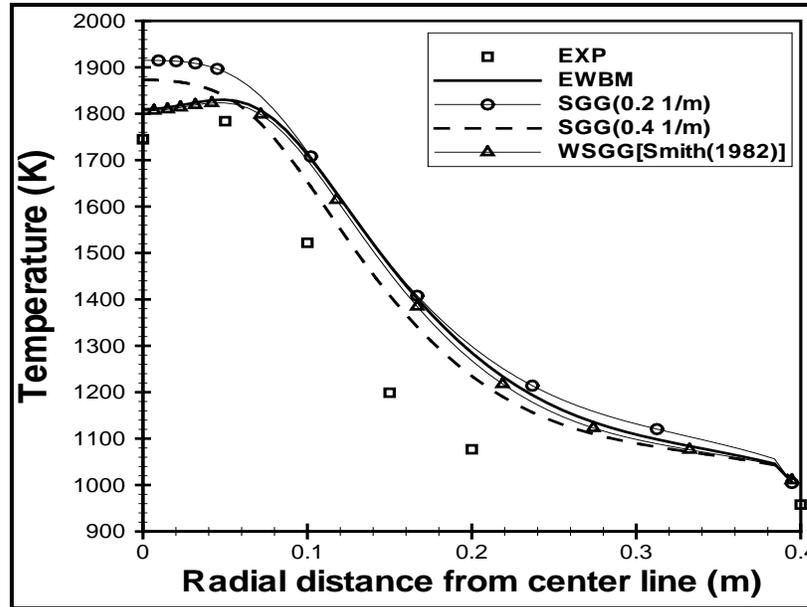


Figure 10 Temperature profile at distance 0.384m from the burner with different radiation models, air combustion.

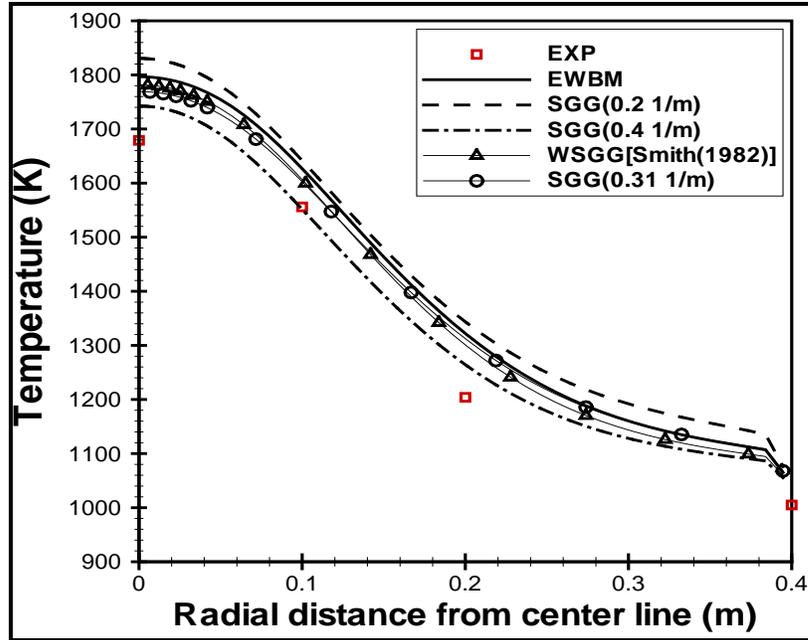


Figure 11 Temperature profile at distance 0.553m from the burner with different radiation models, air combustion.

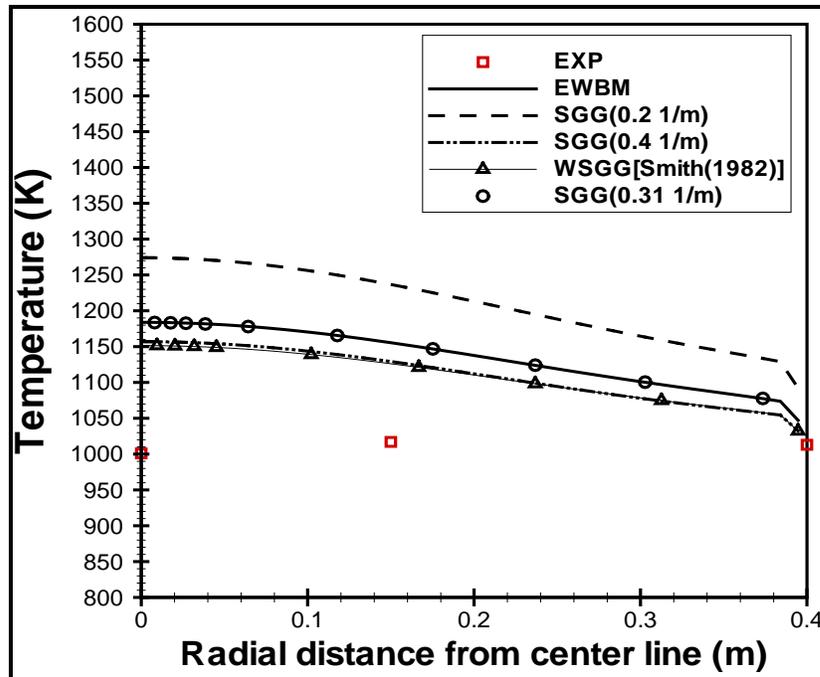


Figure 12 Temperature profile at distance 1.4m from the burner with different radiation models, air combustion.

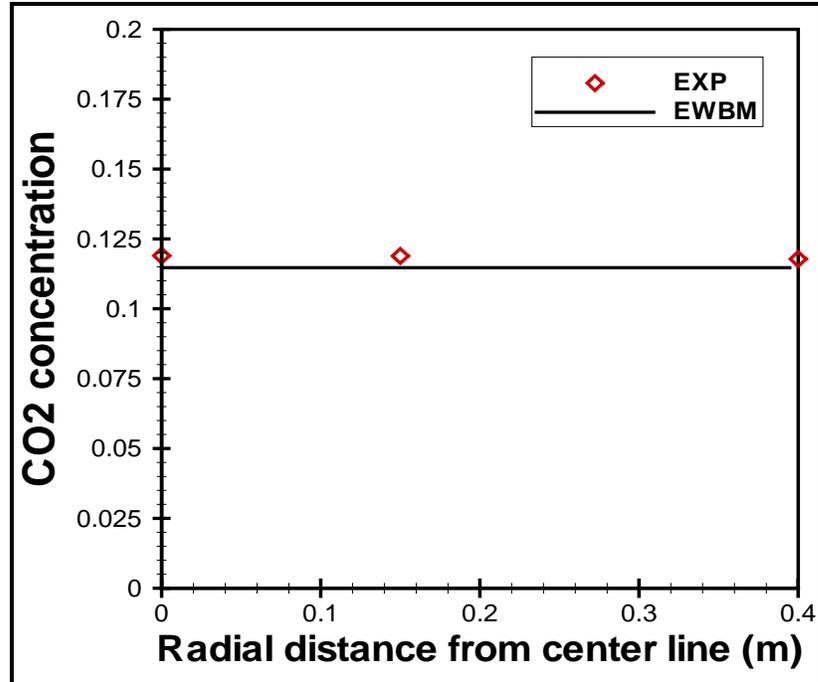


Figure 13 CO<sub>2</sub> concentrations at distance 1.4m from the burner, air combustion.

For the oxy-fuel combustion, OF21, figures 14 and 15 show the temperature profile at distance 0.553m and 1.4m from the burner, respectively. It is found that the temperature predicted by the exponential wide band model (EWBM) is above the measured values by 3.8% at 0.553 m whereas 15% at 1.4 m. In addition, three grey one clear gas weighted sum of grey gases model (WSGG), which is introduced by Andersson [43], leads to temperature results with error of 4% and 12% at distance of 0.553m and 1.4m from the burner respectively. As shown in figure 15, the accuracy of four grey one clear gas weighted sum of grey gases model (WSGG) is about +16 %. The temperature computed by the simple grey gas model (SGG) represented by the effective absorption coefficient of 0.31 (1/m) deviate by 13% compared to experimental data. The absorption coefficient of 0.2 (1/m) over-predicts the temperature by 7.5% and 23% at distance 0.553m and 1.4m from the burner, respectively. The more accurate results are obtained if the

absorption coefficient of 0.4 (1/m) is used in the solution, the difference did not exceed +4% at distance 0.553m and +15% at distance 1.4m from the burner. Smith weighted sum of grey gases model (WSGG) over-predicts the temperature by 5.5% and 19% at distance 0.553m and 1.4m respectively. It is noted in figure 16 that the computed CO<sub>2</sub> concentrations are in good agreement with experimental data and the error less than 5%.

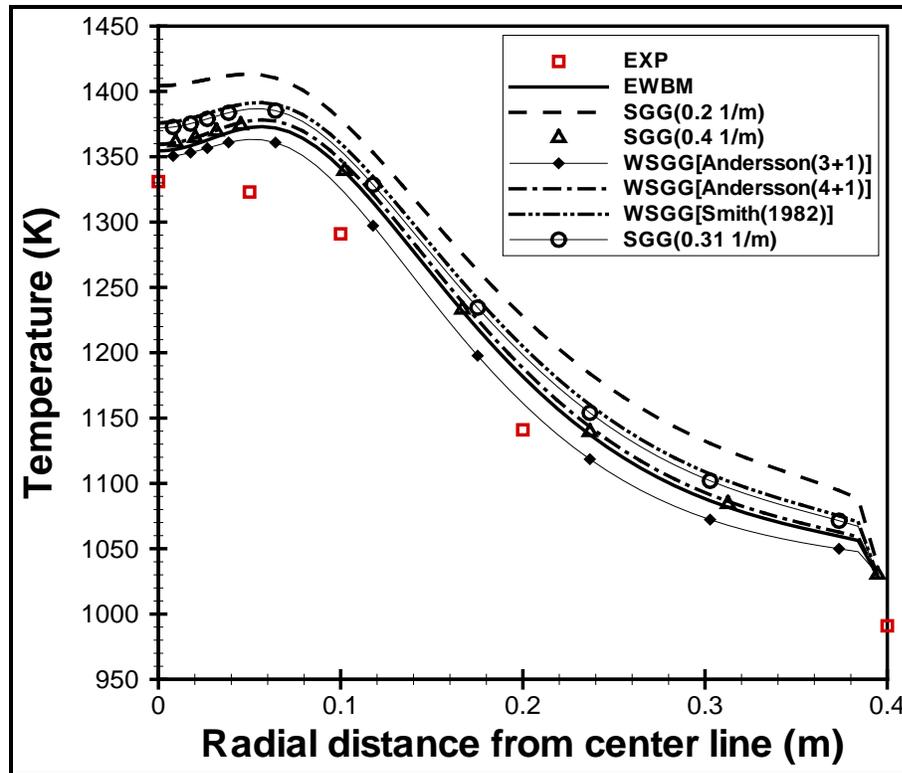


Figure 14 Temperature profile at distance 0.553 m for different radiation models, OF21 combustion.

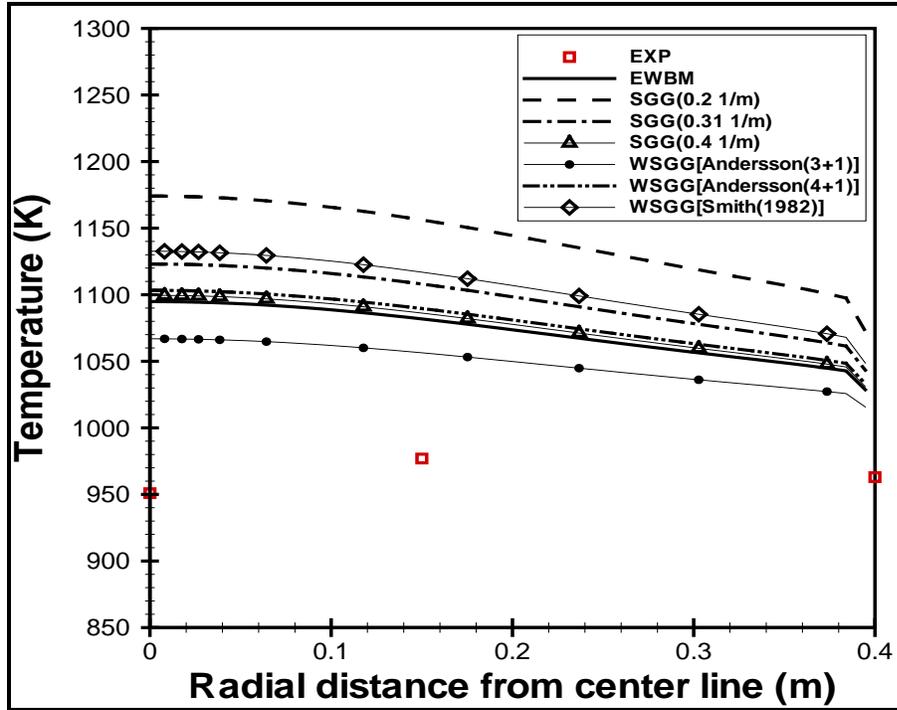


Figure 15 Temperature profile at distance 1.4m from the burner with different radiation models, OF21 combustion.

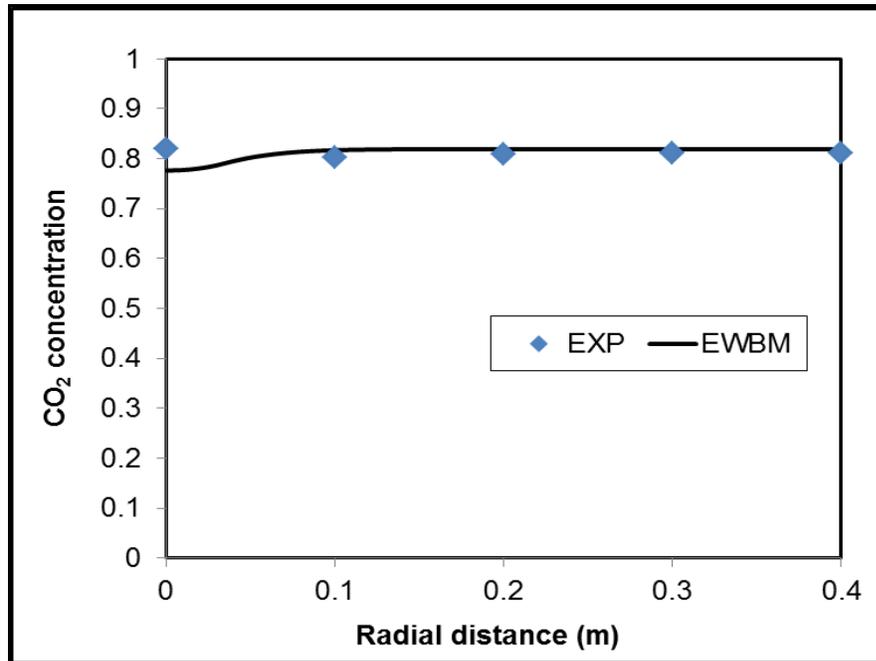


Figure 16 CO<sub>2</sub> concentrations at distance 0.384m from the burner, OF21 combustion.

In case of OF27 combustion, it is clear from figure 17 that the exponential wide band model (EWBM) and the weighted sum of grey gases model (WSGG) over predict the gas temperature. The maximum deviation in the temperature is found to be 8% in case of the exponential wide band model (EWBM). It is increased to 22% for three grey one clear gas weighted sum of grey gases model. However, a little improvement in the accuracy is noted if four grey one clear gas weighted sum of grey gases model is used and in this case the error is calculated to be 19%. The results of simple grey gas model (SGG) model with absorption coefficient of 0.4 (1/m) are in good agreement with the measurements, the error less than 5%. The absorption coefficient of 0.2 (1/m) over-predicts the results by 13%. It should be mentioned that the weighted sum of grey gases model (WSGG) developed by Smith is predicting higher temperature by 13%.

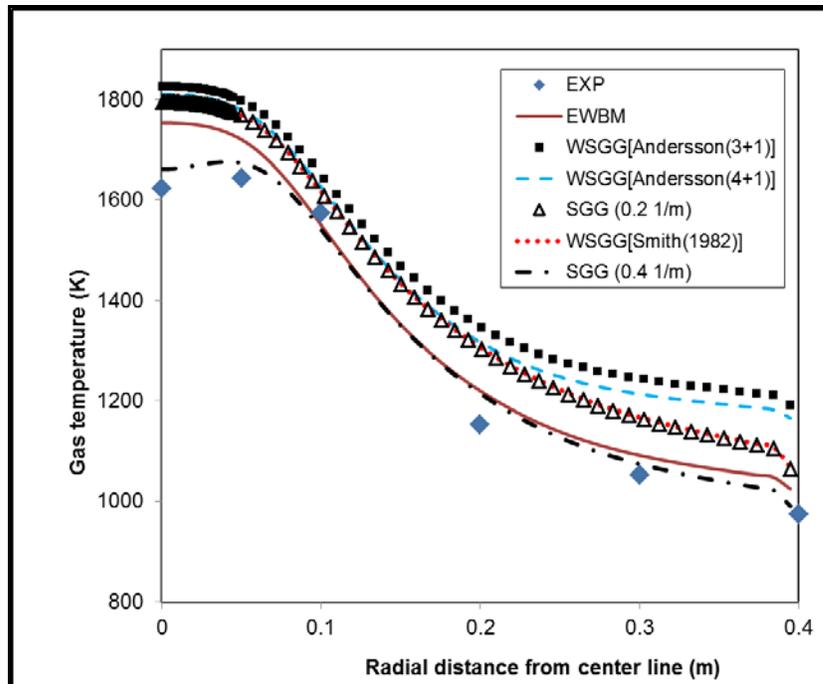


Figure 17 Temperature profile at distance 0.384m from the burner with different radiation models, OF27 combustion.

It is clear from figure 18 that the CO<sub>2</sub> concentrations are over-estimated compared to the measured data and the maximum discrepancy in the results does not exceed 3.7%.

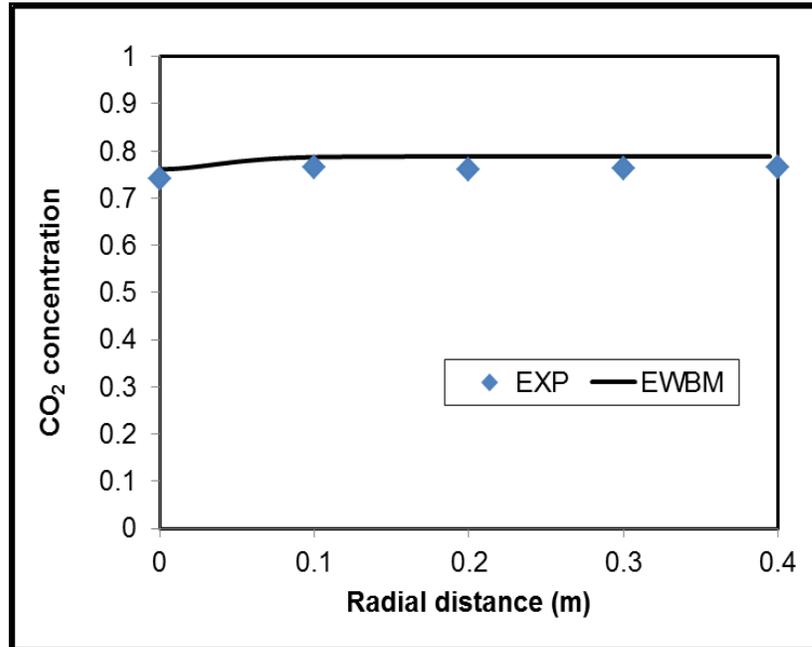


Figure 18 CO<sub>2</sub> concentrations at distance 0.384m from the burner, OF27 combustion.

#### 4.5 Assessment of gas radiation models

In the previous section, the results of CFD solution method have been checked and evaluated. CFD solution includes modeling of fluid flow, turbulence, radiation and combustion. CFD findings were validated by comparing them with experimental measurements. The maximum discrepancies of the results were about 20%. There is a necessity to investigate the behaviors of gas radiation models exclusively. For this purpose, the measured data of Andersson experiment [42, 49] (i.e. temperature profile, species concentration and radiation intensities) were used in the analysis. Once the above

mentioned properties are known, one can calculate the emissivities and radiation intensities by different models. Then these computed results should be compared to the ones measured in the experiment. The gas radiative properties models that used in the study are;

- Leckner model
- Perry model
- Exponential wide band model (EWBM)
- Weighted-sum-of-grey-gases model (WSGG)

The above models are coded and executed for this purpose. Initially, total gas emissivities as a function of gas temperature for air-firing, OF21 and OF27 were computed and plotted in figure 19, 20 and 21 respectively. In air combustion situation, it is noted that the Leckner model predicts the emissivities with error not exceed 5% compared to exponential wide band model (EWBM). The deviation at temperature above 1000 °C is less than 2%, however, this deviation increases up to 5% at temperature of 700 °C. The Perry model under-estimates the emissivities by 15%. It over-predicts the emissivities at temperature below 700 °C. For the last model, weighted-sum-of-grey-gases model (WSGG) introduced by Smith, it is always over-estimating the emissivities with maximum difference of 8%. It should be mentioned that the Leckner model is under-predicting the total emissivities at temperature higher than 1200 °C and over-predicting them below this temperature, the minimum deviation is found at 1200 °C and it is about 0.3%.

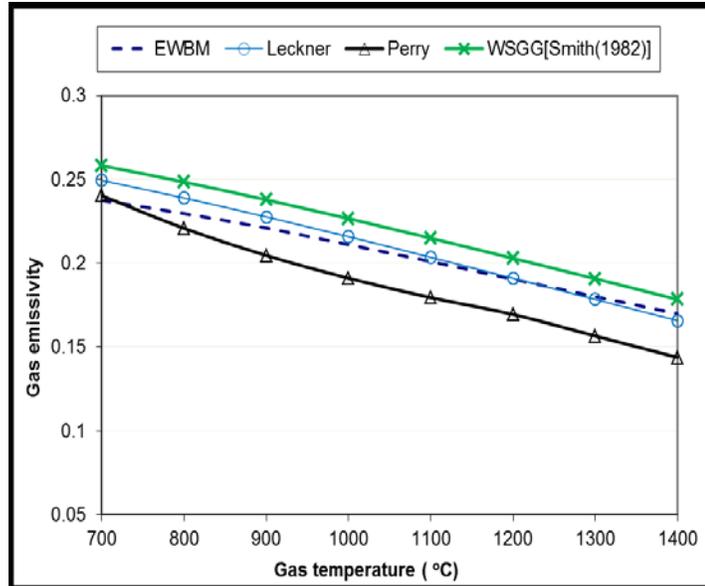


Figure 19 Emissivity profiles as a function of gas temperature for different radiation models, air combustion

In case of OF21, as depicted in figure 20, the difference between exponential wide band model (EWBM) and Leckner is about 4.6% however, this percentage decreases as temperature increases and reaches to 0.5% at 1400 °C. The emissivities calculated by Andersson four grey one clear gas WSGG model are under-predicted by 12.7% at temperature 1400 °C, however, this error reduced at lower temperatures to reach 0.35% at 700 °C. In the same trend as in air case, the Perry model under-predicts the total gas emissivities with error of 39%.

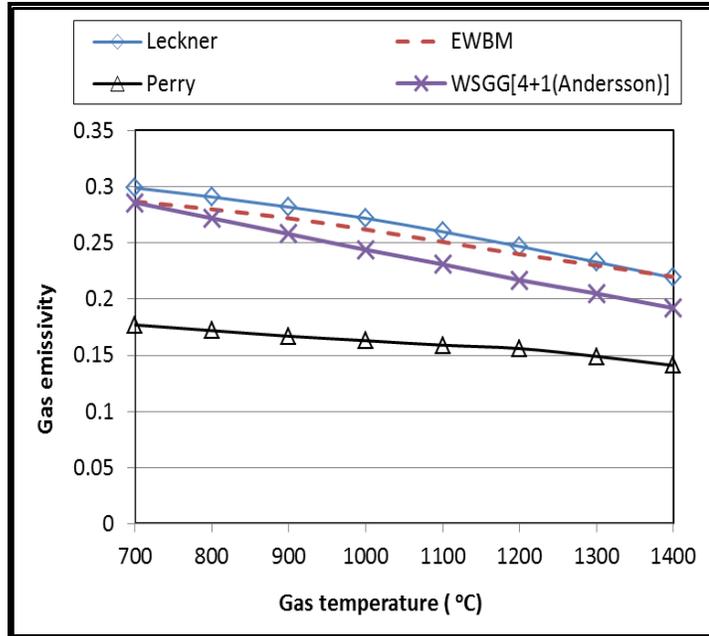


Figure 20 Emissivity profiles as a function of gas temperature for different radiation models, OF21 combustion

Figure 21 demonstrates the total gas emissivities as a function of gas temperature for OF27 combustion. It appears that the Leckner model gives higher emissivities than of exponential wide band model (EWBM). The maximum deviation is about 4%. The Perry model predicts the lower emissivities by 43.5% at 700 °C. The emissivities of four grey one clear weighted sum of grey gas model are lower by 20% at temperature of 1400 °C compared to exponential wide band model. At temperature 700 °C, the error is around 8%.

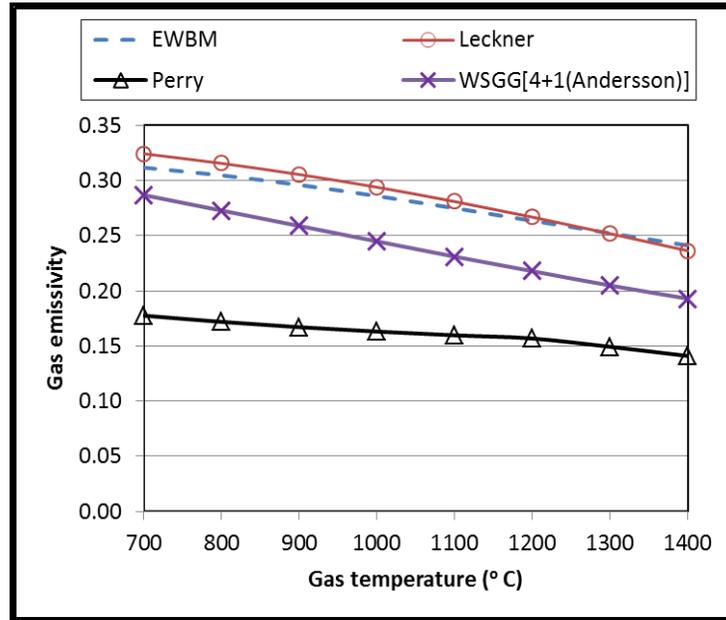


Figure 21 Emissivity profiles as a function of gas temperature for different radiation models, OF27 combustion

In addition, the weighted-sum-of-grey-gases models (WSGG) are used to determine the radiation intensities for air-combustion, OF21 and OF27. The non-correlated formulation of radiative transfer equation (RTE) given by equation (54) is chosen in the solution.

$$\bar{I}_{vk,n} = \bar{I}_{vk,n-1} \bar{\tau}_{vk,n-1 \rightarrow n} + \bar{I}_{bv,k,n-1/2} (1 - \bar{\tau}_{vk,n-1 \rightarrow n}) \quad (54)$$

The path length is divided into n cells as shown in figure 22 in which the radiation intensity for each cell can be computed from the neighboring cells.

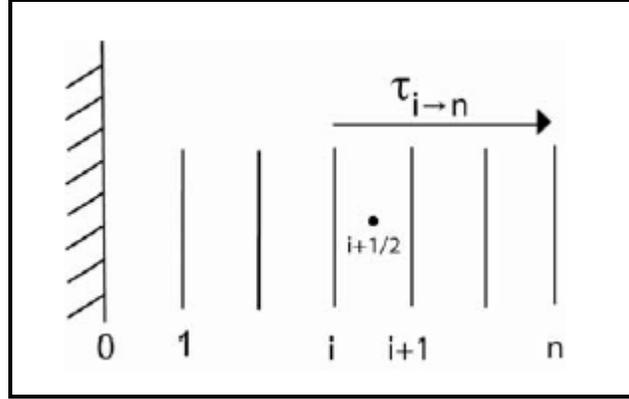


Figure 22 Discretization of a path length.

The weighted-sum-of-grey-gases models (WSGG) are coupled with the non-correlated formulation of radiative transfer equation (RTE) in order to find the radiation intensities.

For gas  $j$  the discretized radiative transfer equation (RTE) becomes

$$I_{j,n} = I_{j,n-1} \exp(-\kappa_j S_m P(Y_{CO_2} + Y_{H_2O})) + a_j I_{b,j,n-1/2} (1 - \exp(-\kappa_j S_m P(Y_{CO_2} + Y_{H_2O}))) \quad (55)$$

Where  $S_m$  is the length of the computational cell ranging from  $n-1$  to  $n$ . The total intensity is the sum of the individual gas intensities of the  $j$  gases.

$$I_n = \sum_j I_{j,n} \quad (56)$$

Figures 23, 24 and 25 present the radiation intensities at distance 0.384m from the burner for air, OF21 and OF27 respectively. It appears that the modeled intensities are under-predicted for all flames. In the air flame the computed radiation intensities by weighted-sum-of-grey-gases model (WSGG) are lower by 60% from the measured values. The modeled radiation intensities in case of OF21 flame found to be more accurate than in air case. As seen in figure 24 the maximum differences between the calculated and the measured intensities are about 33% and 30% for three grey one clear weighted sum of

grey gas model and four grey one clear weighted sum of grey gas model respectively. The model of three grey one clear weighted sum of grey gas under-predicts the intensities by approximately 58% for OF27 and this percentage reduced to 56% if four grey one clear weighted sum of grey gas model is used.

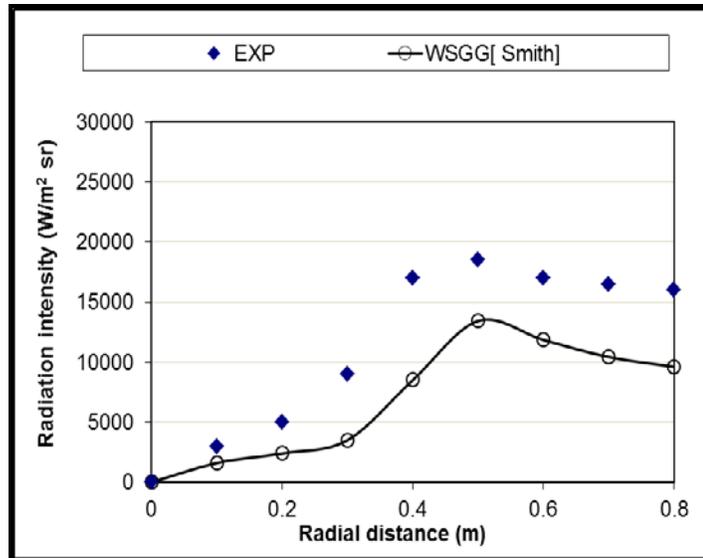


Figure 23 Radiation intensity at distance 0.384m from the burner with different radiation models, air combustion.

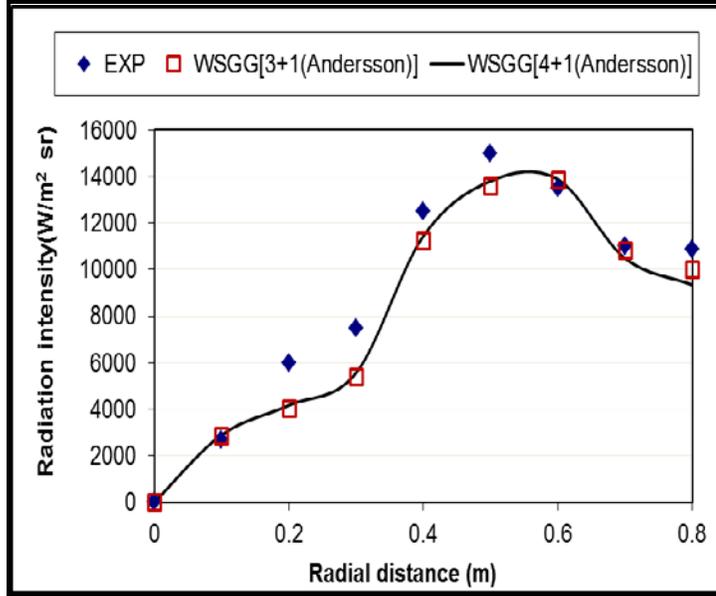


Figure 24 Radiation intensity at distance 0.384m from the burner with different radiation models, OF21 combustion.

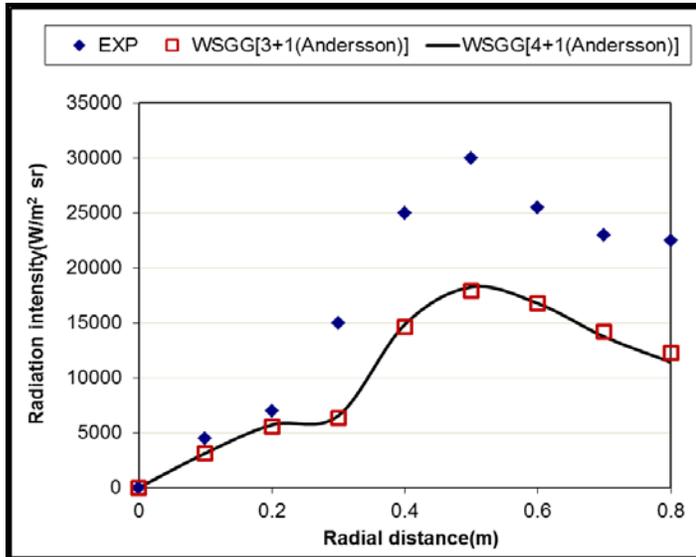


Figure 25 Radiation intensity at distance 0.384m from the burner with different radiation models, OF27 combustion.

At distance 1.4m from the burner the radial radiation intensities are plotted for air, OF21 and OF27 as shown in figure 26, 27 and 28. It is noted that for the air flame the weighted-sum-of-grey-gases model (WSGG) over-predicts the intensities by 20.7% at the

furnace wall. In the OF21 as depicted in figure 27 the modeled intensities by four grey one clear weighted sum of grey gas model are higher than the measured ones by about 21% and 23% in case of three grey one clear weighted sum of grey gas model. Finally, OF27 flame intensities are computed and plotted in figure 28, it is found that the discrepancy in the results of the two models does not exceed 27% compared to the measured data.

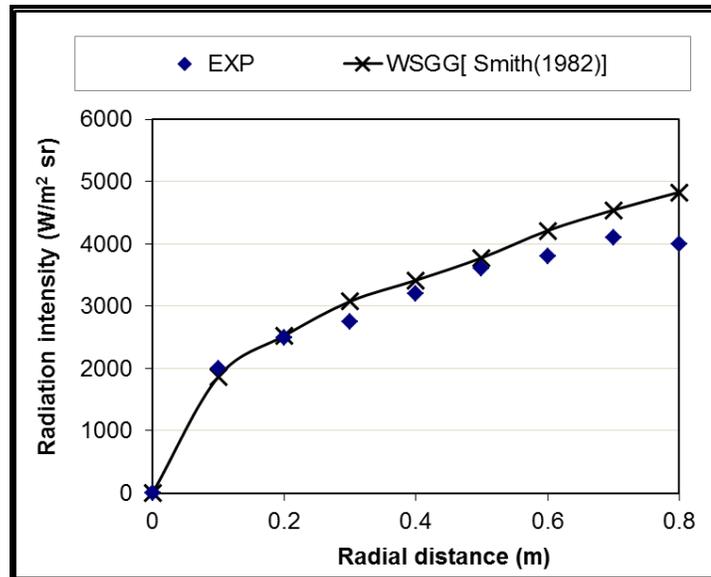


Figure 26 Radiation intensity at distance 1.4m from the burner with different radiation models, air combustion.

It should be mentioned that the source of errors in the calculated intensities for air and OF27 at distance 0.384 m is attributed to the soot formation and the high temperature in this region. As reported by Andersson [49], both the air and OF27 flame display a luminous appearance due to the soot formed near the burner. Furthermore, it is found that the volume fraction of the soot particles increases from air to OF27 conditions. On the other hand, the soot volume fraction is drastically reduced from the air and OF27 flames

to the OF21 flame. It can be stated that the differences between the measured total intensity and the modeled gas intensity is the influence of radiation from soot.

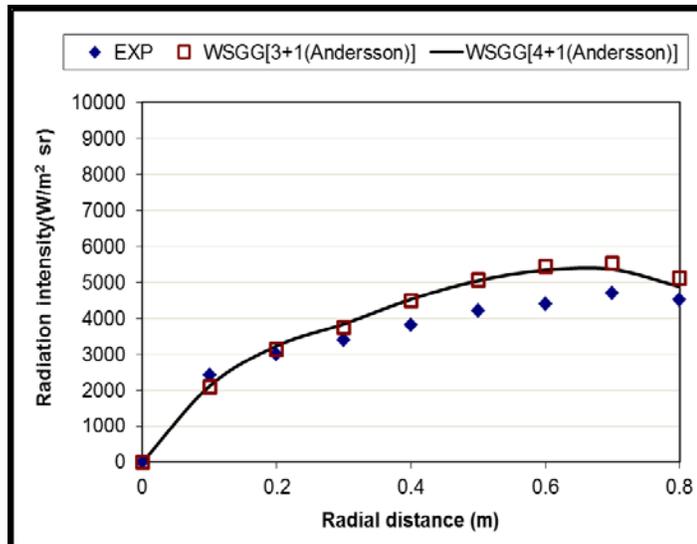


Figure 27 Radiation intensity at distance 1.4m from the burner with different radiation models, OF21 combustion.

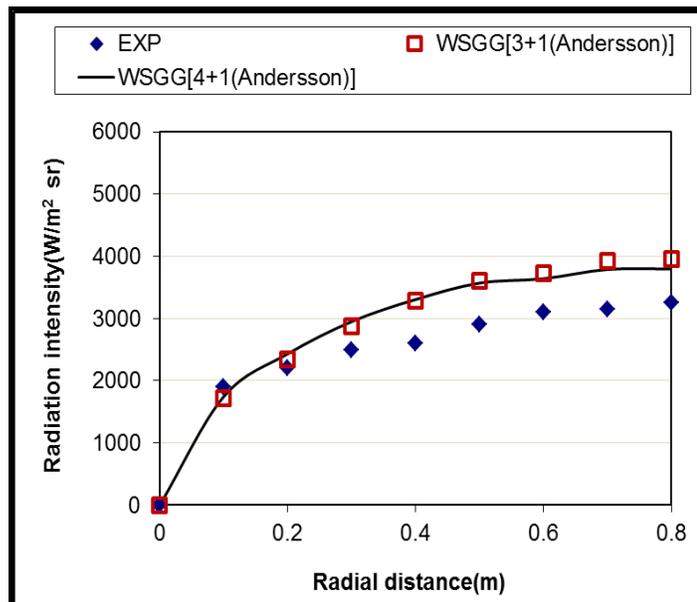


Figure 28 Radiation intensity at distance 1.4m from the burner with different radiation models, OF27 combustion.

# CHAPTER 5

## CFD MODELING OF A WATER TUBE BOILER UNDER AIR AND OXY-FUEL COMBUSTION

### 5.1 Introduction

This chapter is aimed at investigating the influence of gas radiation models on combustion characteristics for full scale fossil fuel furnace. Methane ( $\text{CH}_4$ ) is burnt under three different scenarios (i.e. air, OF21 and OF27). CFD solution of these scenarios based on different gas radiation models are obtained and discussed. Grid independency of the results is verified. The error introduced in the results of oxyfuel combustion is calculated when weighted sum of grey gas model (WSGG), which developed by Smith for air-fuel combustion, is used. Finally, the characteristics of oxyfuel combustion are compared to those of air-fuel combustion in this boiler.

### 5.2 Boiler description

CFD modeling of a full scale fossil-fuel fired furnace is conducted in this study. The objective of this modeling is to investigate the influence of radiation models on combustion characteristics and to analyze the impact of replacing air with a mixture of  $\text{O}_2$ - $\text{CO}_2$  in the combustion. The boiler is manufactured by Mitsubishi Heavy Industries

(MHI) and is operated in Saudi Aramco refineries at Abqaiq and Ras Tanura. A brief summary of design data is given in Table 8. The layout of this boiler is depicted in figures 29 and 30.

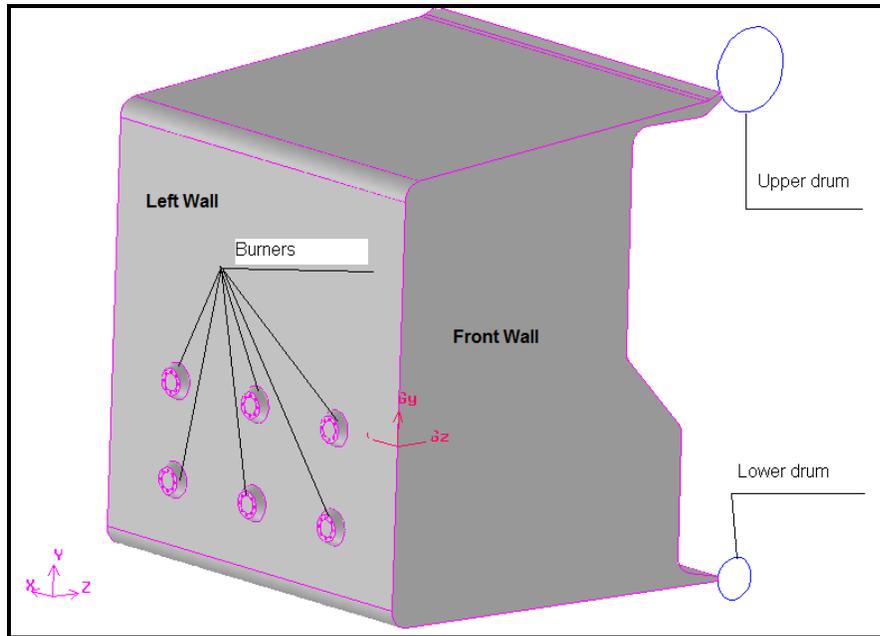


Figure 29 Layout of furnace of the boiler

It consists of D-type tubes extending from the mud drum (lower drum) through the bottom wall and forms the burner wall. The top part of the D-shaped tubes forms the top wall of the furnace and ends at the upper drum. Tubes on the front wall extend from lower to upper drum while the tubes forming the side walls extend from a lower header to an upper header.

Table 8 Design data of boiler

Mass flow rate of steam, lb/h	Drum pressure, psig	Steam temperature, F	Number of burners
750,000	750	750	6 on two levels

The water-steam mixture inside the tube is normally saturated and has a saturation temperature corresponding to the drum pressure. Accordingly, the front, side, rear and top walls of the furnace provide a constant temperature (538.6 K) boundary condition. The tubes forming the bottom wall are thermally insulated.

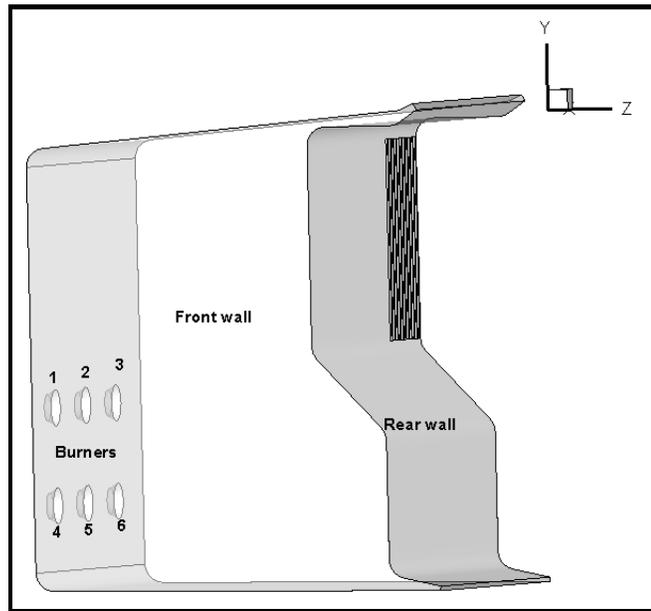


Figure 30 Side view of Furnace of the boiler

It is clear that the boiler consists of 6 burners located in two levels. Each burner has eight fuel nozzles with 24 fuel jets in each nozzle. The burner construction is shown in Figure 31. This indicates that evaporation may be assumed to start at the beginning of the sidewalls of the riser tubes. The heat flux imparting on the sidewall causes the quality of the water/steam mixture to increase as the mixture flows up the riser tubes and also at the top part of the tubes.

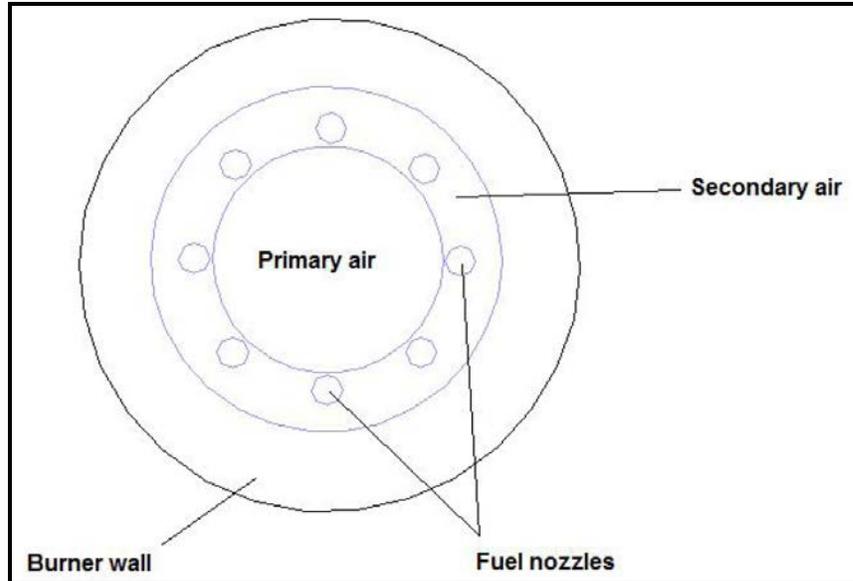


Figure 31 burner construction

### 5.3 Problem setup

The main purpose of this study is to evaluate different gas radiation models and their influences on the result of air and oxy-fuel combustion scenarios. The study of the full scale furnace includes three different cases of combustion;

- Air combustion (  $O_2$  @ 21 % Vol and  $N_2$  @79% Vol )
- OF21 (  $O_2$  @ 21 % Vol and  $CO_2$  @79% Vol )
- OF27 (  $O_2$  @ 27 % Vol and  $CO_2$  @73% Vol )

Methane fuel ( $CH_4$ ) is burnt in the furnace with stoichiometric ratio of about 1.16 (16% excessive air). It is assumed that all fuel is converted to combustion products of  $CO_2$  and  $H_2O$  and the total heat input into the furnace is 295 MW. The fuel mass flow rate through each burner is about 0.98 kg/s. Primary and secondary air passing through each burner are equal to 8.87 kg/s and 10.7 kg/s respectively. The combustion process takes place in environment of atmospheric pressure. Computational Fluid Dynamic (CFD) method is

used in the solution procedure. Gambit software was used to build a 3-D geometry of the furnace. The volume of furnace was discretized into 1,335,180 control volumes at which the solution findings were not affected by the grid (see figures 32 and 33).

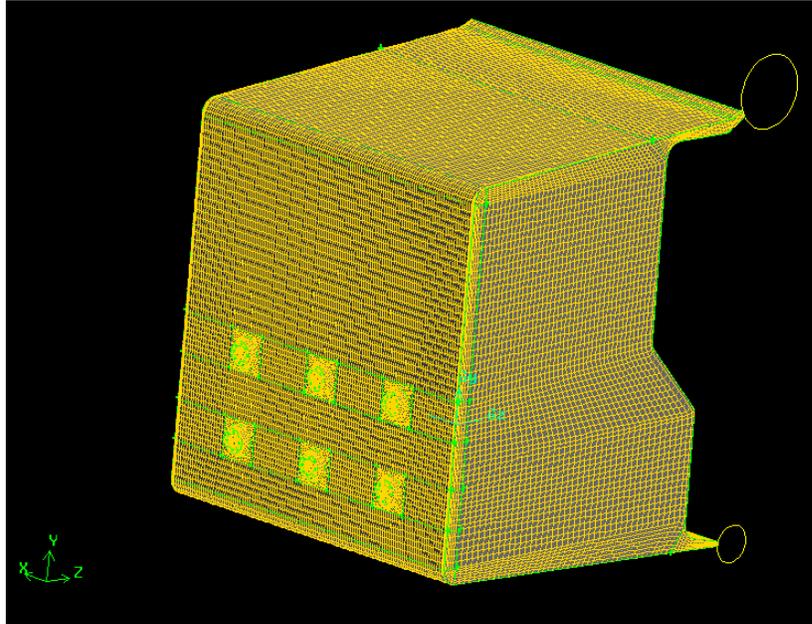


Figure 32 General view of the boiler.

It is a very important issue in the numerical solution to ensure that the obtained results are grid independent and this is what will be shown in the next section. All the walls of furnace are kept at constant temperature of 538.6 K except the bottom wall which is insulated. Fluent package was utilized in the solution procedure. The summary of the models and sub-models in the solution as follows:

- Incompressible ideal gas assumption
- Standard  $k-\varepsilon$  turbulent model
- Temperature-dependent properties
- One step volumetric reaction for species transport

- Finite-rate/Eddy-dissipation turbulence-chemistry interaction
- Discrete Ordinate Radiation Model (DO)

Different gas radiative properties models are tested in the solution. In addition to the models that used in the numerical simulation of the previous chapter, user-defined functions (UDFs) of Leckner and Perry models were coded and compiled to be used in the solution.

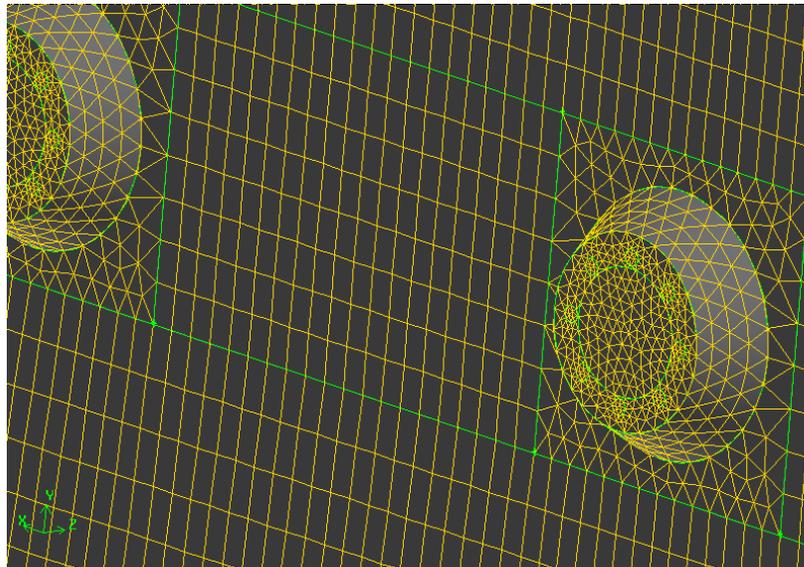


Figure 33 Closer view of the burners

Since it is required in the radiation modeling, the boiler mean beam length must be calculated. By using equation (23), the mean beam length of the furnace is found to be 4.5m. In fact, this feature of boiler geometry has a significant effect on the radiation calculation and the determination of temperature thereafter.

## 5.4 Grid independence study

As mentioned above, all numerical results have to be independent of increasing the discretization level of certain geometry. The variation of the results with different grids must be very small and has insignificant impact on them. Once this condition is satisfied, the results are considered to be grid independent. For this purpose, three different discretized geometries were examined in this study. They are listed in Table 9.

Table 9 Different grids of the furnace

	Grid name	Number of cells	Percentage of increasing
1	Grid-1	935805	----
2	Grid-2	1095555	+17%
3	Grid-3	1335180	+43%

It is found after evaluation and comparison of the obtained results that, in general, the deviation of the results with increasing the level of discretization above 1335180 cells will not exceeds 0.01%. Figure 34 shows the temperature profile for the above mentioned grids along vertical line in the middle between first and second burner near the rear wall i.e.  $z = 6$  m. It is noted that the maximum difference in temperature between Grid-3 and Grid-2 is about 0.02% and this percentage increased to be 0.07% in the comparison between Grid-3 and Grid-1.

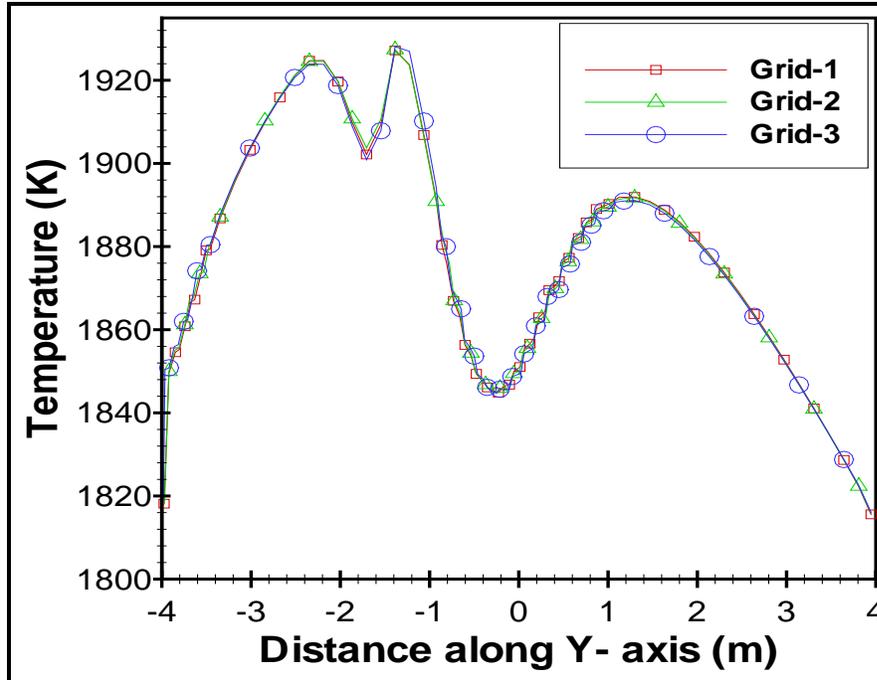


Figure 34 Temperature profile at line ( $x=3.264$ ,  $z=6$ ) along Y-axis

In the same manner, the temperature profiles along Z-axis for the three cases were plotted in figure 35. The maximum variation in temperature between Grid-1 and Grid-3 is calculated to be 1.6% however; it is reduced to 0.2 % in case of Grid-3 compared to Grid-2. Furthermore, total heat flux has been checked in the analysis. Figure 36 demonstrates the total heat flux along X-axis at line located on the left wall above the first row of burners i.e.  $y= 2$  m. it is noted that, the deviation in the results between Grid-3 compared to Grid-2 is around 0.03% while it becomes 0.06% for Grid-1.

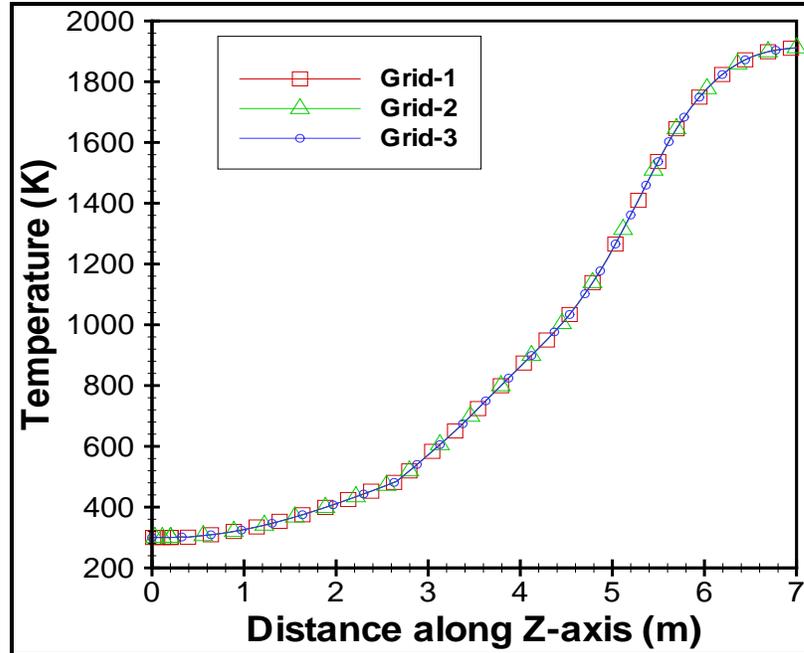


Figure 35 Temperature profile at line (x=4.539, y=-2.135) along Z-axis

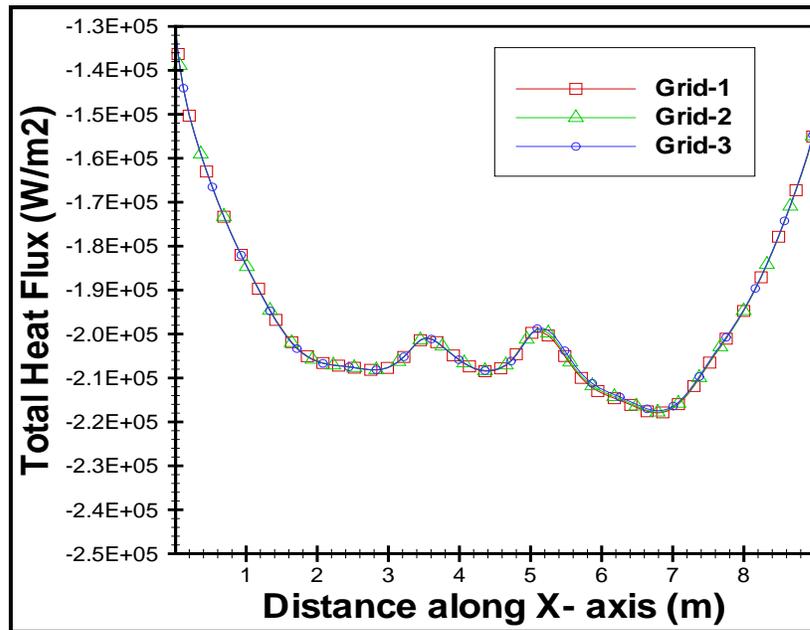


Figure 36 Total heat flux profile at line (y=2, z=0) along X-axis

Moreover, the total heat flux profiles along Z-axis at front wall and y=3m are plotted in figure 37. The difference between Grid-3 and Grid-1 is found to be 0.9%. The deviation

of 0.4% is computed for Grid-2 related to Grid-3. It should be mentioned from the above discussion that the variation in the results if Grid-3 is used in the solution will be insignificant.

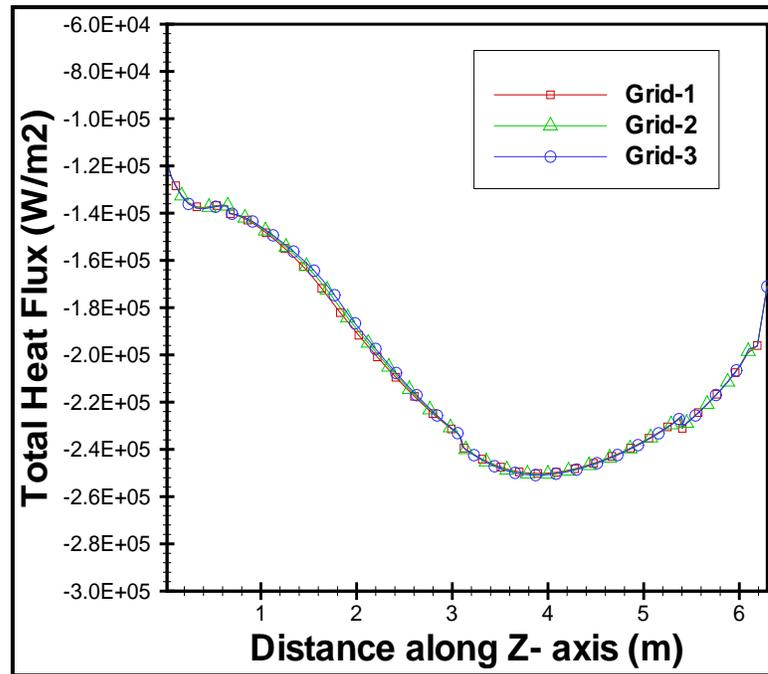


Figure 37 Total heat flux profile at line (x=0, y=3) along Z-axis

## 5.5 Results and discussions

### 5.5.1 Comparisons between gas radiation models

Numerical simulation for air combustion, OF21 combustion and OF27 combustion is obtained by utilizing fluent packaged software. Different gas radiation models are coupled with discrete ordinate method to calculate the radiation heat transfer in the furnace. These models are exponential wide band model (EWBM), Leckner model, Perry model, weighted sum of grey gas model (WSGG) and simple grey gas model (SGG). The solution was converged when the summation of the residual in the governing equation at

all the domain nodes was less than 0.001%. Various absorption coefficients are examined in the analysis to determine the optimal value that reflects the same behavior of the exponential wide band model (EWBM) which is considered the benchmark model. Temperature profiles at different location along Z- axis are plotted for different models. They were evaluated and compared to the reference model to check their accuracy. It is found in the air combustion as shown in figure 39, the weighted sum of grey gas model (WSGG) introduced by Smith [10] predicts higher temperatures by 0.39% compared to exponential wide band model (EWBM). Simple grey gas model (SGG) with absorption coefficient equal to 0.25 (1/m) under-predicts the temperature by 4.5% while the absorption coefficient of 0.18 (1/m) leads to lower results by 3.3%. The deviation in the temperature is reduced to 0.37% in case of the absorption coefficient has a value of 0.1 (1/m). Figure 38 represents the temperature profiles along Z-axis at  $x=4.539\text{m}$ ,  $y=4\text{m}$ ; the maximum difference of 7.2% is computed when the simple grey gas model with absorption coefficient of 0.25 (1/m) is used in the solution. Weighted sum of grey gas model is in good agreement with the benchmark solution, i.e. the deviation less than 0.5%. The more accurate results are obtained when the absorption coefficient of 0.1 (1/m) is chosen in the solution, the error less than 0.35%.

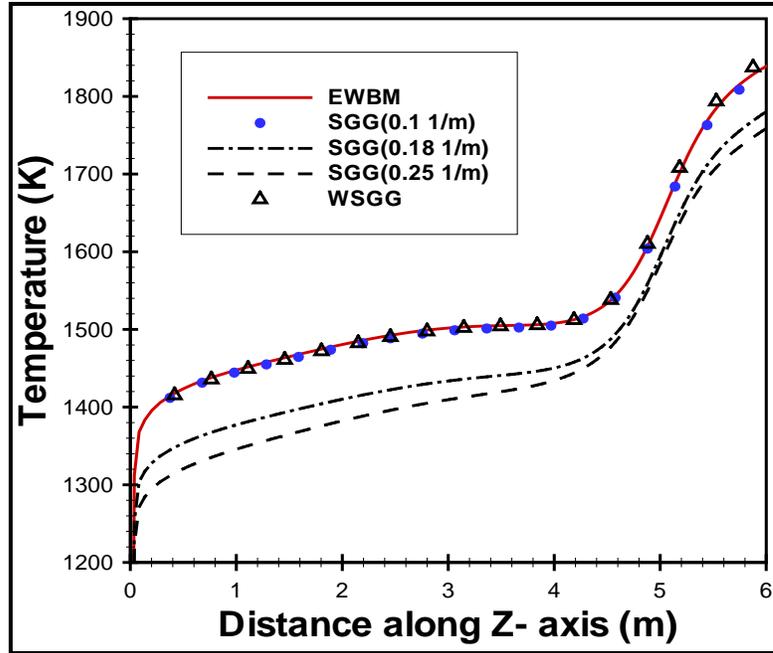


Figure 38 Temperature profile at line ( $x=4.539$ ,  $y=4$ ) along Z-axis, air combustion

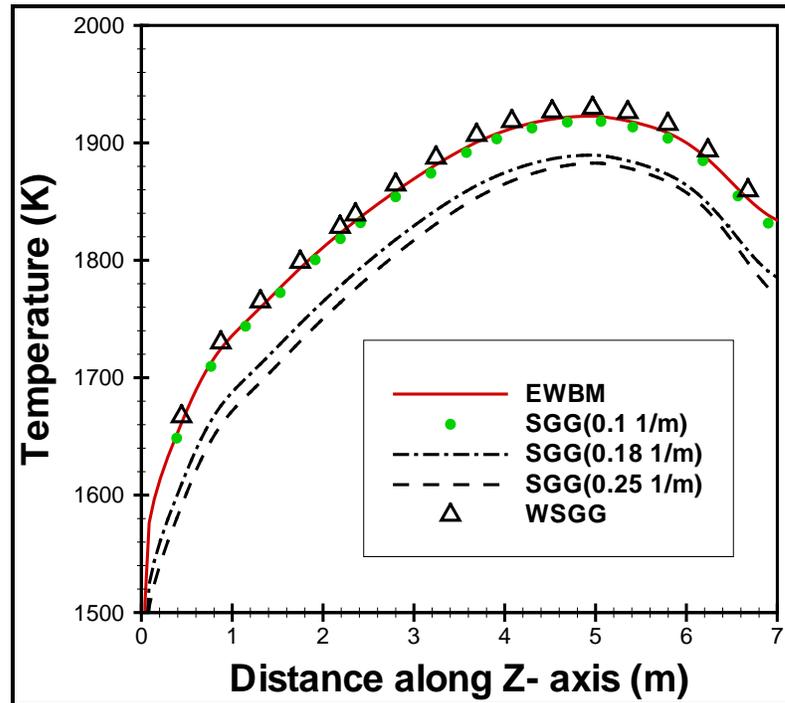


Figure 39 Temperature profile at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis, air combustion

In case of OF21 combustion, the temperature distribution at line located in the middle of the furnace at  $y=4\text{m}$  along Z-axis is represented in the figure 40. Leckner model, Perry model, weighted sum of grey gas model (WSGG) are all over-predicting the temperature by different amounts. The maximum increase in case of Leckner is about 13.5%. The differences of 4.5% and 1% are calculated for Perry model and weighted sum of grey gas model (WSGG), respectively. The appropriate value for the absorption coefficient to get more accurate results is found to be in the range of 0.1 - 0.18 (1/m); the error less than  $\pm 1.4\%$ . Figure 41 represents the temperature profile at line located between first and second columns of burners in the mid-way between the rows of burners along Z- axis (i.e.  $x=3.264\text{m}$ ,  $y=-1.0675\text{m}$ ), the results of four grey one clear weighted sum of grey gas model (WSGG) developed by Andersson et al [26] are in good agreement with those obtained in case of exponential wide band model (EWBM). It over-predicts the temperature by 1.2%. Leckner model leads to increasing in the temperatures by about 7.9% whereas Perry model over-estimates the results by 2.8%. Simple grey gas model with absorption coefficient of 0.1 (1/m) gives more accurate results i.e. the difference less than 0.89%. The results of simple grey gas model with the absorption coefficient equal to 0.25 (1/m) are under-predicted by 2.1%.

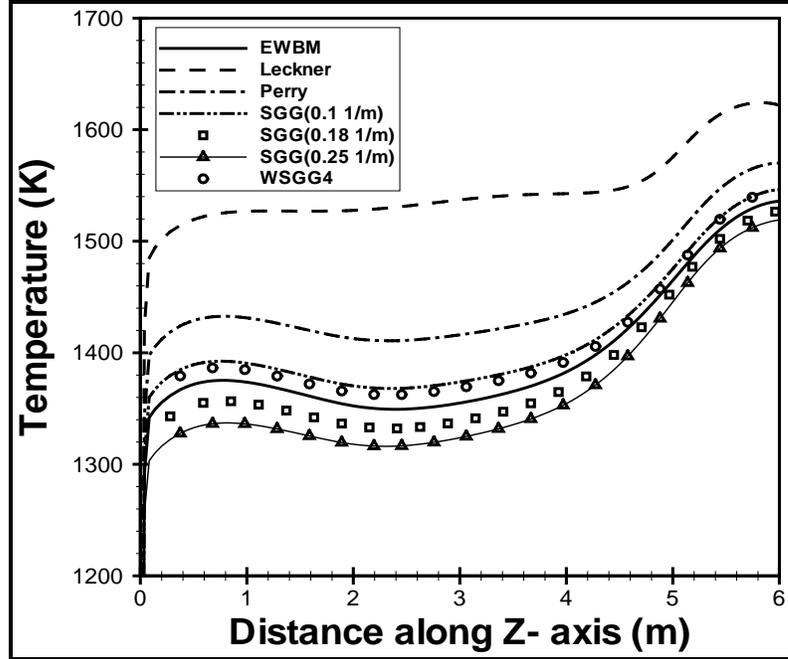


Figure 40 Temperature profile at line (x=4.539, y=4) along Z-axis, OF21 combustion

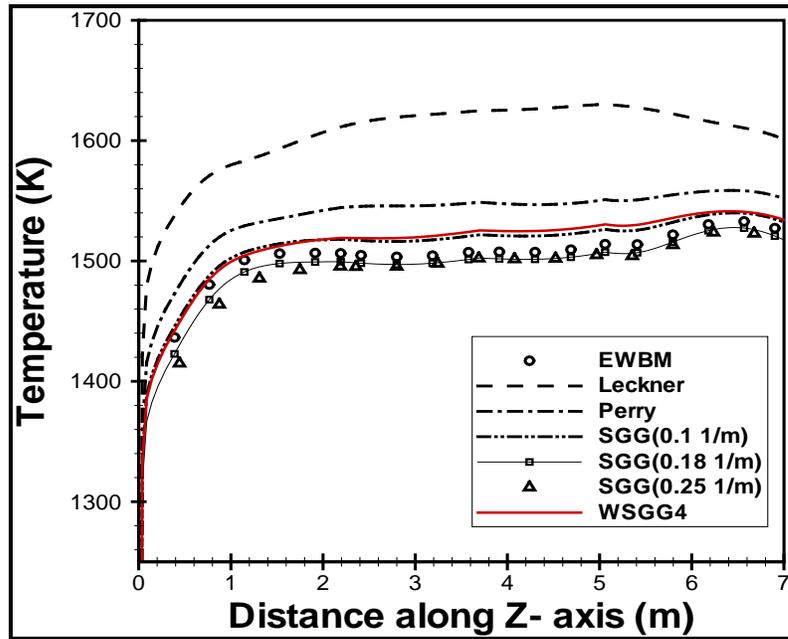


Figure 41 Temperature profile at line (x=3.264, y=-1.0675) along Z-axis, OF21 combustion

Temperature distribution at line located at x=3.264m and y= -1.0675m along Z- axis is represented in figure 42 for OF27 combustion. It is found that Leckner model over-

estimates the temperature by 8.6%. The deviation of 2.9% is computed when Perry model is used. Weighted sum of grey gas model (WSGG) and simple grey gas model (SGG) with absorption coefficient of 0.1 (1/m) over-predict the temperature by 1%. The absorption coefficients of 0.25 (1/m) and 0.18 (1/m) are under-estimating the temperature by 1.8% and 0.79% respectively.

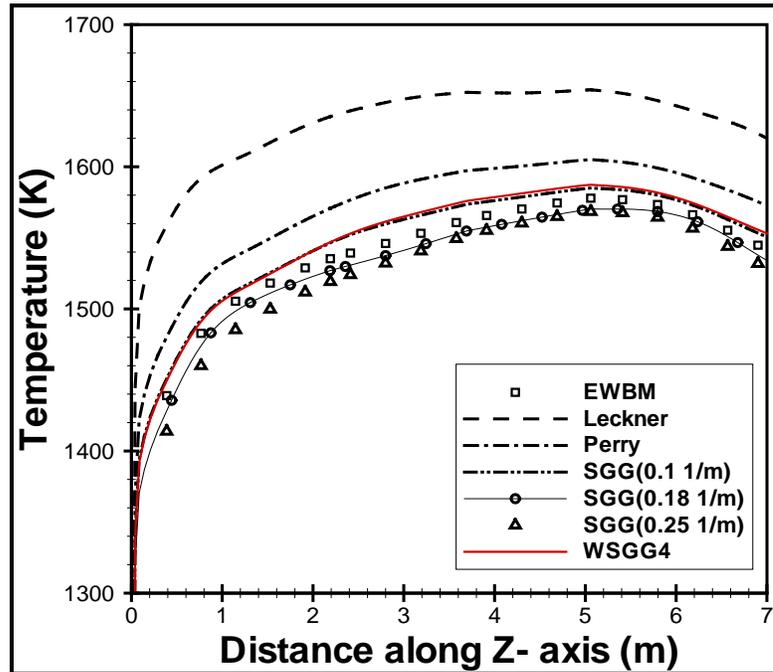


Figure 42 Temperature profile at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis, OF27 combustion

In addition, temperature distribution at  $x=4.539\text{m}$  and  $y=4\text{m}$  along Z- axis is shown in figure 43. In the same trend Leckner model, Perry model, and weighted sum of grey gas model (WSGG) are over-predicting the temperature by 15%, 5% and 1.2% respectively. Simple grey gas model leads to error of 2.8% when absorption coefficient of 0.25 (1/m) is used and this error is reduced to 1.35% in case the absorption coefficient equal to 0.18 (1/m). It should be emphasized that the maximum difference in the result when the

absorption coefficient is selected from the range of 0.25 (1/m) to 0.1 (1/m) will not exceed 4.3%.

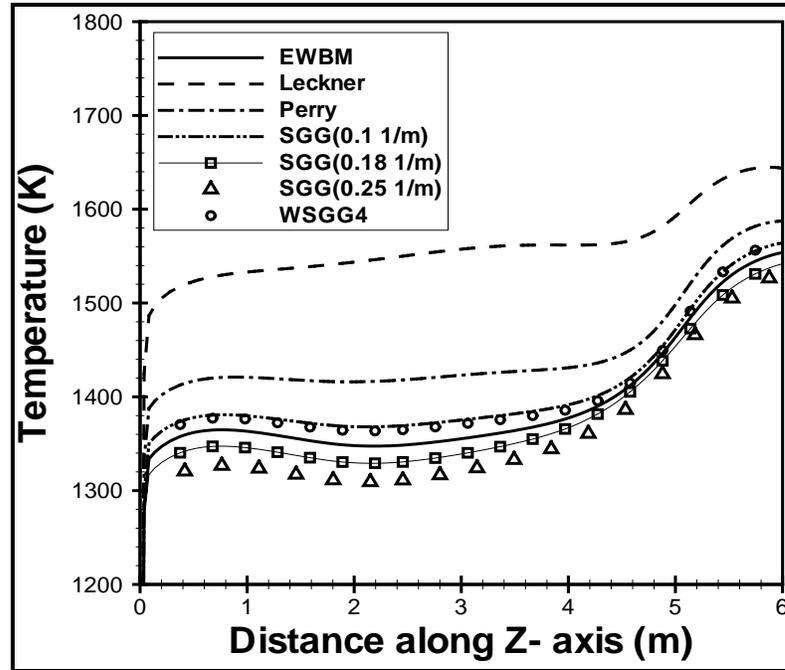


Figure 43 Temperature profile at line (x=4.539, y=4) along Z-axis, OF27 combustion

Weighted sum of grey gas model (WSGG) that developed by Smith [10] is one of the default model available in the Fluent 6.2. Errors of applying this model for oxyfuel combustion is calculated and compared to the exponential wide band model (EWBM). It is found that as shown in figure 44 and figure 45 the predicted  $\text{CO}_2$  and  $\text{H}_2\text{O}$  mole fractions by these two models for OF27 combustion are in good agreement. The maximum difference in  $\text{CO}_2$  mole fraction is less than 0.1% whereas 0.9% in prediction of the  $\text{H}_2\text{O}$  mole fraction.

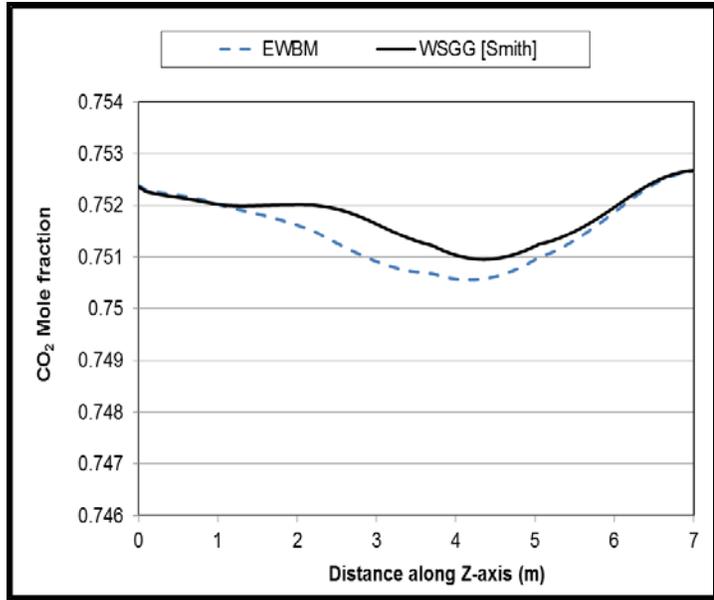


Figure 44 CO<sub>2</sub> mole fraction profile at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion

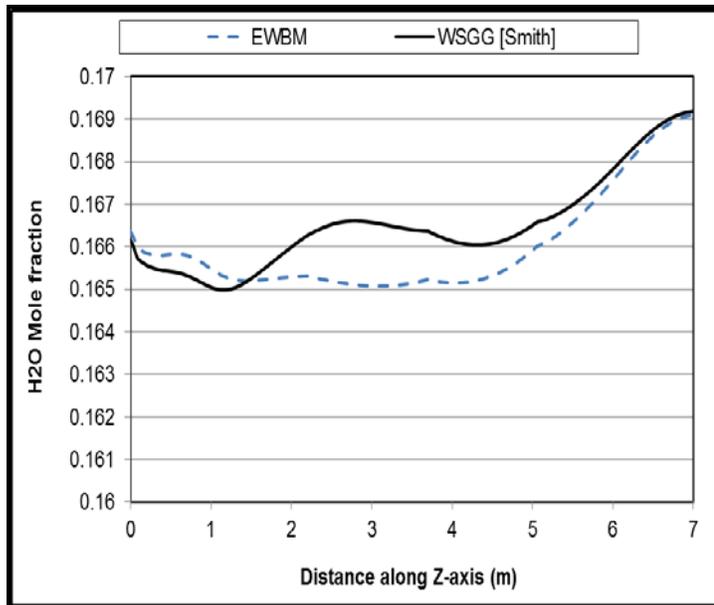


Figure 45 H<sub>2</sub>O mole fraction profile at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion

Incident radiation profiles at line ( $x=3.264$ ,  $y=-1.0675$ ) are plotted in figure 46, the results of weighted sum of grey gas model (WSGG) are lower by around 31% compared to the benchmark model (exponential wide band model).

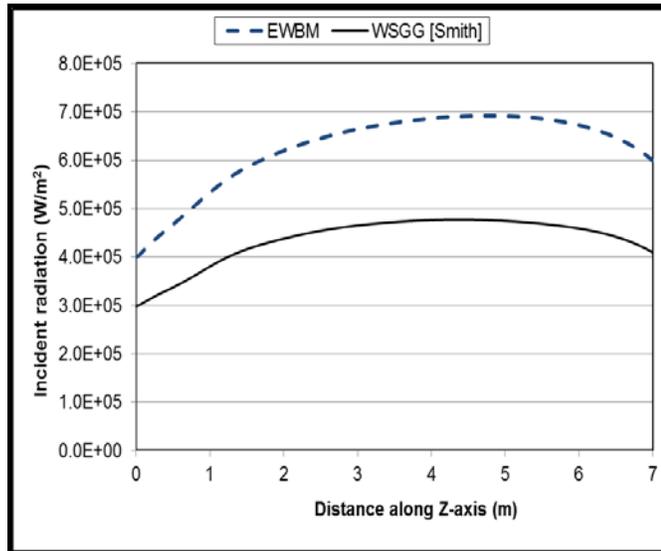


Figure 46 Incident radiation profile at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion

The influence of radiation models is obvious on the total heat flux on the front wall at  $x=0$  and  $y= -1.0675$  as shown in figure 47. Weighted sum of grey gas model (WSGG) is under-predicting the total heat flux by 19%.

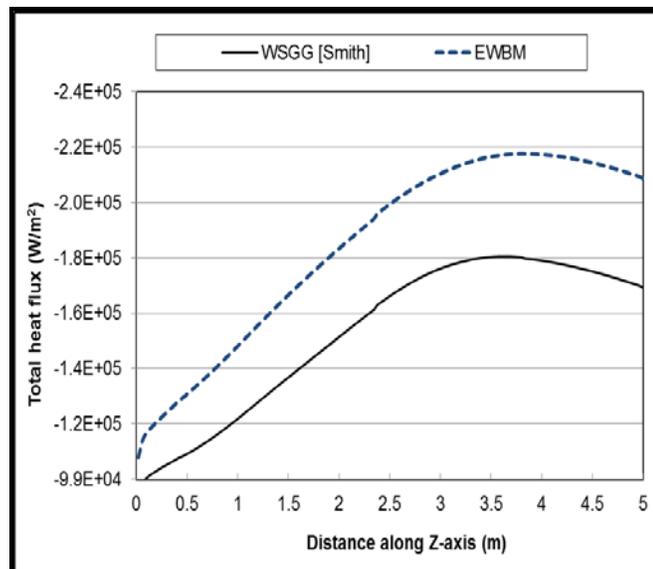


Figure 47 Total heat flux profile at line ( $x=0$ ,  $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion

Temperature profile at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis is shown in figure 48, it is clear that the weighted sum of grey gas model (WSGG) is over-predicting the temperature by 2.14% in comparison to exponential wide band model (EWBM).

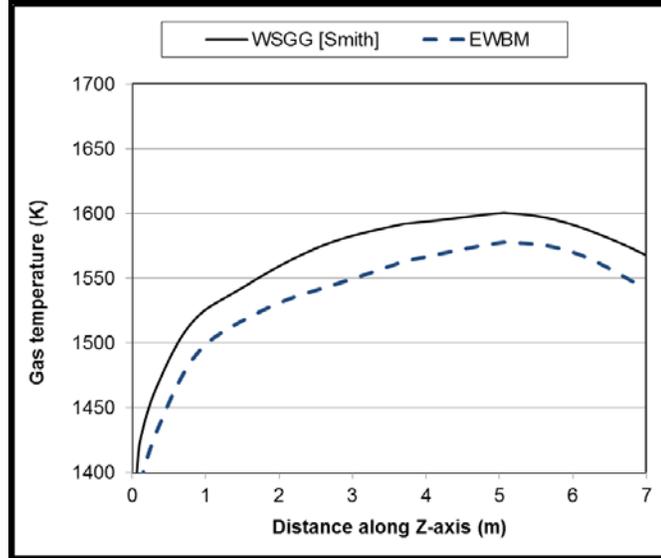


Figure 48 Temperature profile at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for different radiation models, OF27 combustion.

### 5.5.2 Characteristics of Oxyfuel and Air-Fuel Combustion Processes

The effect of replacing  $N_2$  by  $CO_2$  as in the oxyfuel combustion will be discussed for this boiler. It is clear from the temperature profiles in figure 49 that the temperatures are lower in the case of oxyfuel combustion compared to air combustion; reduction reaches up to 21% for OF21 and 18% for OF27. It is thought that the reason of this reduction is attributed to the higher thermal capacity of  $CO_2$  compared to  $N_2$ . Figure 50 demonstrates the specific thermal capacity profiles at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for the air-fuel combustion and the oxyfuel combustion cases, it indicates higher specific

thermal capacity in the case of the oxyfuel combustion in comparison to that of air–fuel combustion. This explains the lower temperature levels in oxyfuel combustion. Increasing the CO<sub>2</sub> recirculation percentage results in higher specific thermal capacity in the combustion products and, thus, resulting in lower temperature levels. In addition to this reason, the radiative emissivity and absorptivity of the CO<sub>2</sub> is higher when it is compared to N<sub>2</sub> which is transparent to the radiation energy and because of this, flame cooling effects exhibited in the oxyfuel combustion cases.

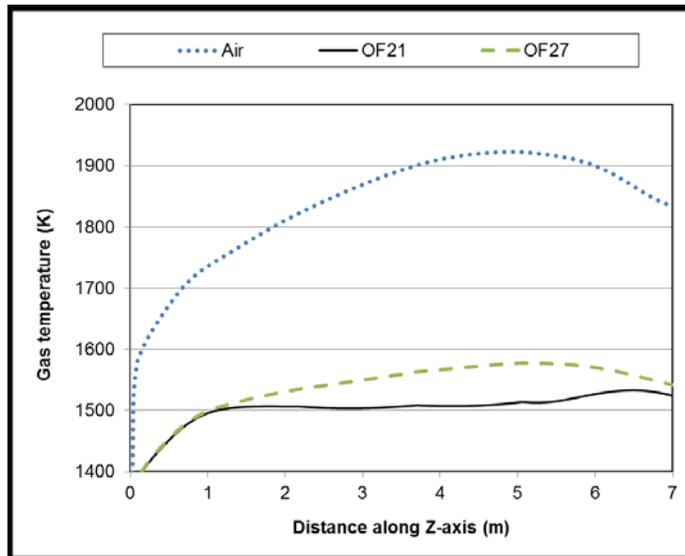


Figure 49 Temperature profiles at line (x=3.264, y=-1.0675) along Z-axis for different combustion cases, EWBM.

Figure 51 shows a comparison between the temperature contours of the air-fuel case and the two oxyfuel combustion cases. The contours are taken in a vertical plane passing through the middle burners. It is clear from the figure that the temperature levels in the oxyfuel combustion cases are lower compared to the air-fuel combustion process. This is expected as it was explained above that the specific heat capacity in the oxyfuel

combustion is higher than that for air-fuel combustion. However, as CO<sub>2</sub> mass percentage is increased, the temperature levels are reduced, as indicated by Figure 51b.

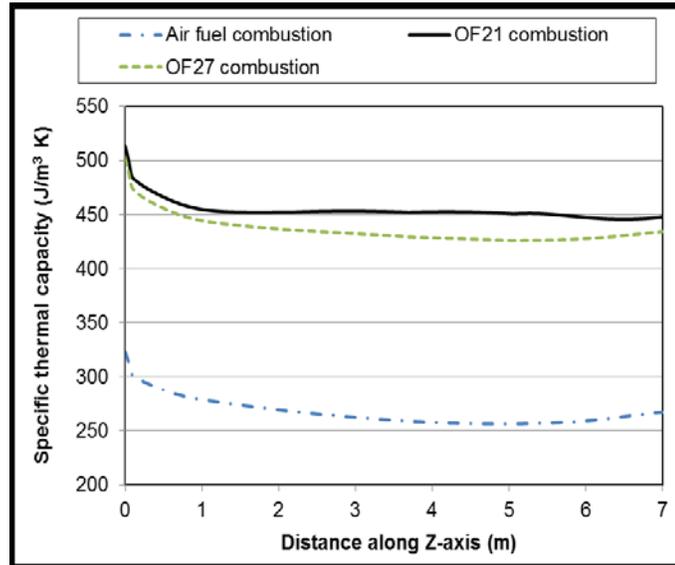
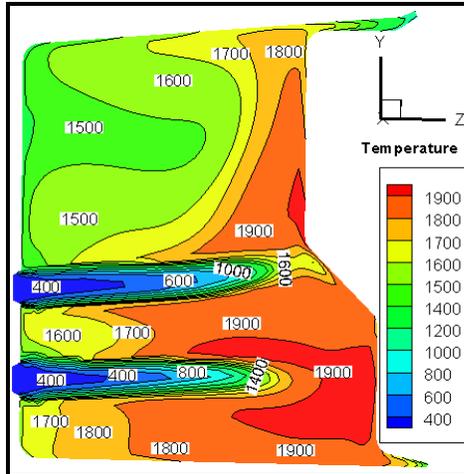
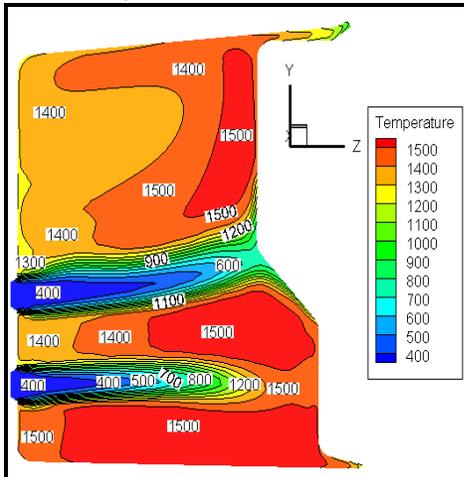


Figure 50 Specific thermal capacity profiles at line (x=3.264, y=-1.0675) along Z-axis for different combustion cases, EWBM.

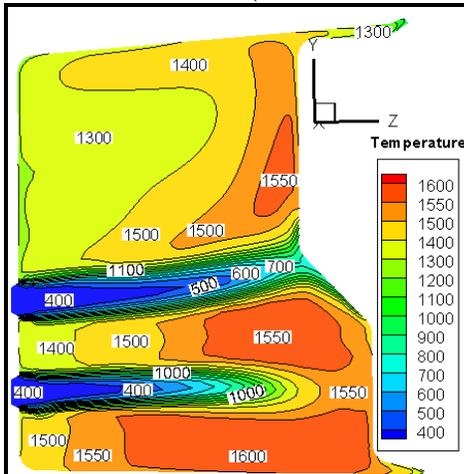
It is noted that the combustion rate is slower in the case of oxyfuel combustion as compared to that of air-fired combustion, and this leading to extending the burning zone causing the flame length to increase. It is also shown that the temperature rise is slower in the oxyfuel combustion cases in comparison to air-fuel combustion and this is attributed to the higher specific heat of CO<sub>2</sub>. Figure 52 shows the contours for temperature of horizontal plane passing through the upper burners (y=0m)



a) Air-fuel combustion

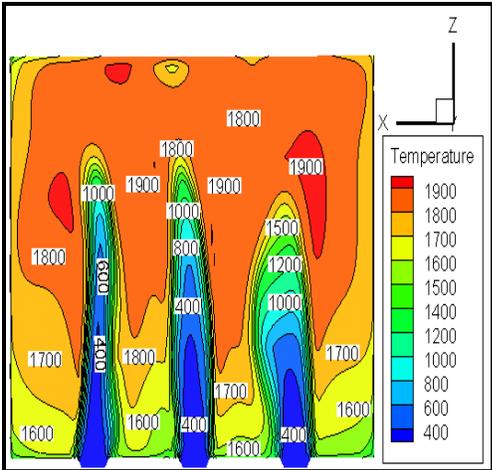


b) Oxyfuel combustion OF21 (21% O<sub>2</sub> and 79% CO<sub>2</sub> by volume)

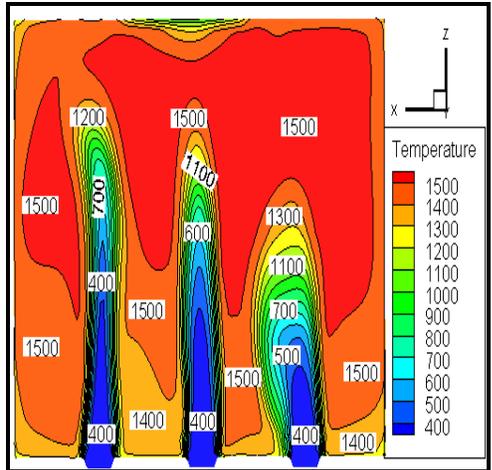


c) Oxyfuel combustion OF27 (27% O<sub>2</sub> and 73% CO<sub>2</sub> by volume)

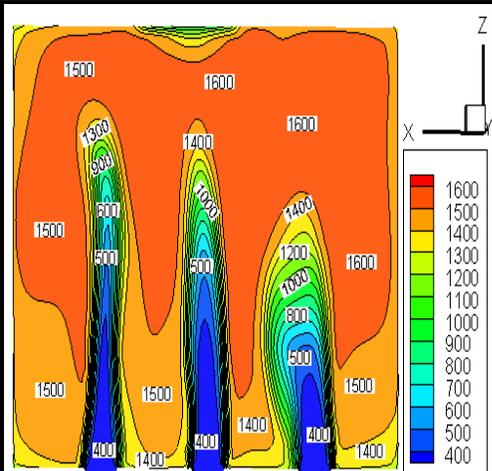
Figure 51 Contours for temperature of vertical plane passing through the middle burners 2 & 5 ( $x=4.539\text{m}$ ), EWBM



a) Air-fuel combustion



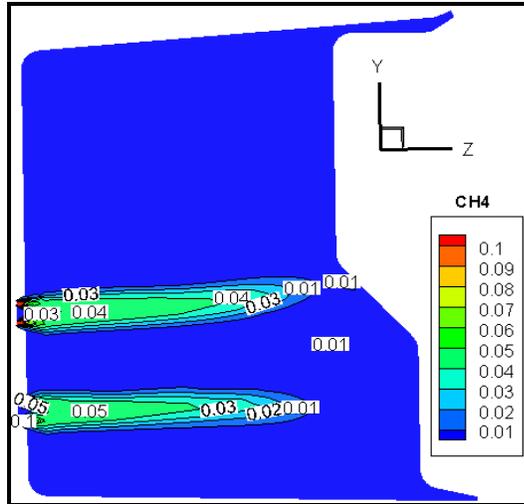
b) Oxyfuel combustion OF21 (21% O<sub>2</sub> and 79% CO<sub>2</sub> by volume)



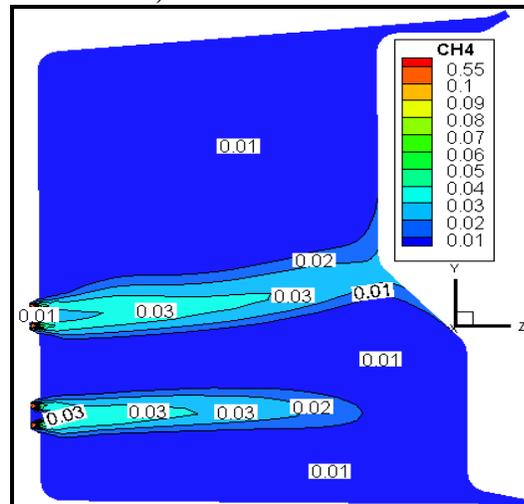
c) Oxyfuel combustion OF27 (27% O<sub>2</sub> and 73% CO<sub>2</sub> by volume)

Figure 52 Contours for temperature of horizontal plane passing through the upper burners (y=0m), EWBM.

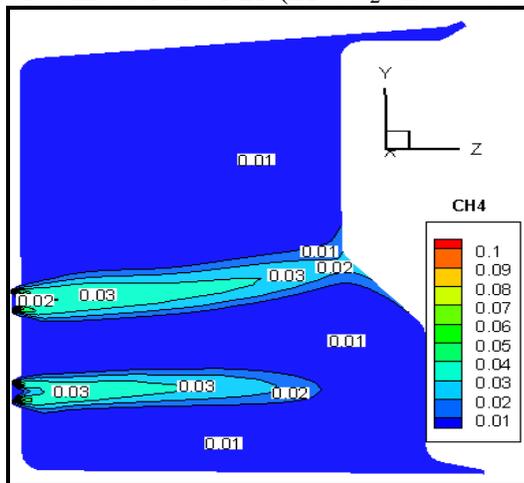
The fuel burning rate is also affected by replacing  $N_2$  by  $CO_2$ . Figure 53 show the contours of  $CH_4$  concentration in the air-fuel combustion and oxyfuel combustion. It is noted that the fuel consumption rate is slower in the oxyfuel combustion cases. Figure 53 b indicates further reduction in fuel burning rate as the  $CO_2$  percentage is increased. In order to clarify the influence of oxyfuel combustion on the burning rate, the turbulent viscosity for the cases of air-fuel combustion and oxyfuel combustion were computed. Figure 54 shows the turbulent viscosity profiles at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for the three combustion cases. It is noted that the higher turbulent viscosity and hence higher mixing rates in the case of the oxyfuel combustion in comparison to those of air-fuel combustion accelerates the reduction in flame temperature and therefore decreases the fuel consumption rate (i.e.  $CO_2$  acts as a cooling gas). This explains the lower levels of temperature in the oxyfuel combustion.



a) Air-fuel combustion



b) Oxyfuel combustion OF21 (21% O<sub>2</sub> and 79% CO<sub>2</sub> by volume)



c) Oxyfuel combustion OF27 (27% O<sub>2</sub> and 73% CO<sub>2</sub> by volume)

Figure 53 Contours for CH<sub>4</sub> mass fraction of vertical plane passing through the middle burners 2 & 5 (x=4.539m), EWBM.

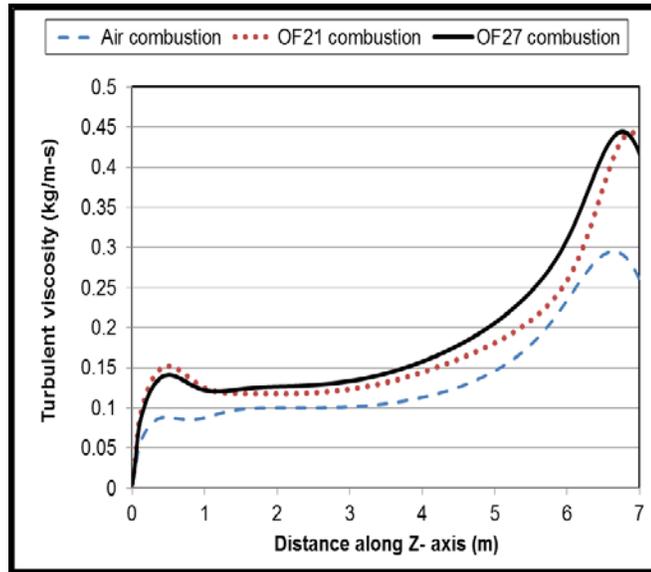
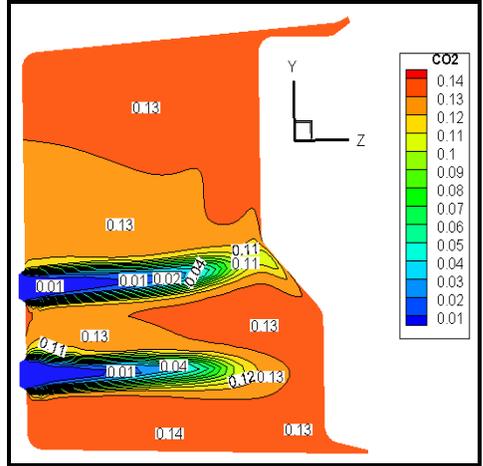
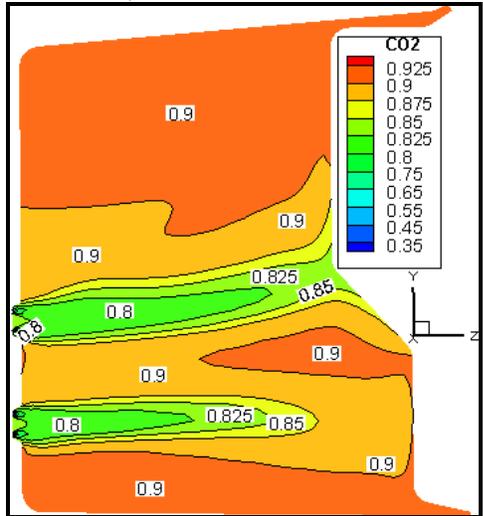


Figure 54 Turbulent viscosity profiles at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for different combustion cases, EWBM.

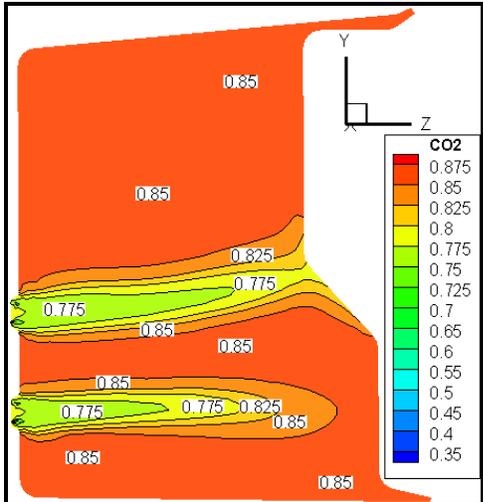
Figure 55 shows the contours of the  $\text{CO}_2$  concentrations of vertical plane passing through the middle burners 2 & 5 ( $x=4.539\text{m}$ ). The  $\text{CO}_2$  concentrations rise from around 14% in the case of air-fuel combustion to above 80% in the oxyfuel combustion in most of the furnace volume. Figure 56 depicts the density profiles at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for the three cases, it indicates higher density in the case of oxyfuel combustion and this is because the molecular weight of  $\text{CO}_2$  is 44, compared to 28 for  $\text{N}_2$ . This explains the smaller gas volume exhibited in the oxyfuel combustion cases. The figure also indicates that the increase in the percentage of  $\text{CO}_2$  recirculation results in higher gas density leading to lower radiation rates.



a) Air-fuel combustion



b) Oxyfuel combustion OF21 (21% O<sub>2</sub> and 79% CO<sub>2</sub> by volume)



c) Oxyfuel combustion OF27 (27% O<sub>2</sub> and 73% CO<sub>2</sub> by volume)

Figure 55 Contours for CO<sub>2</sub> mass fraction of vertical plane passing through the middle burners 2 & 5 (x=4.539m), EWBM.

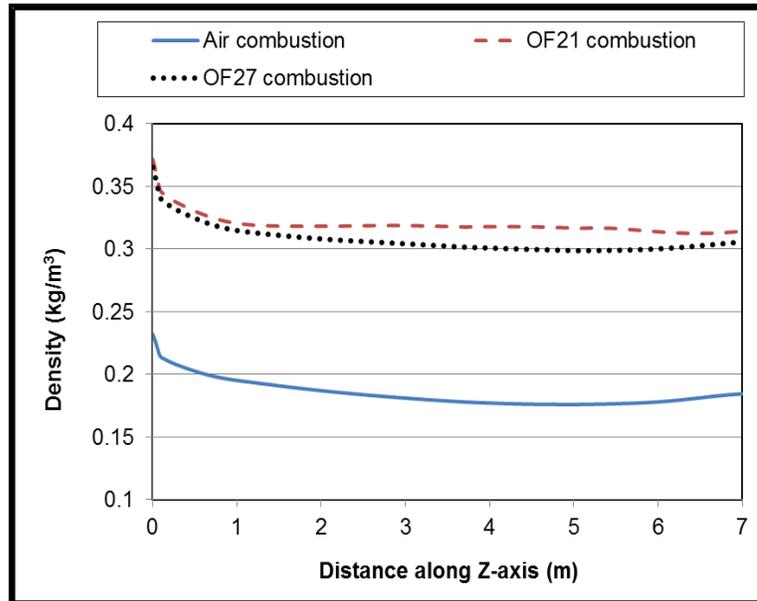
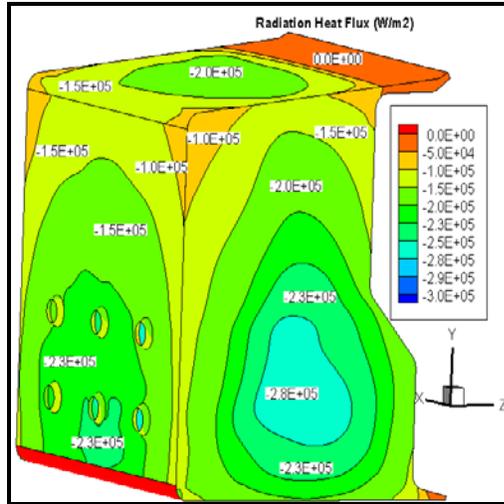
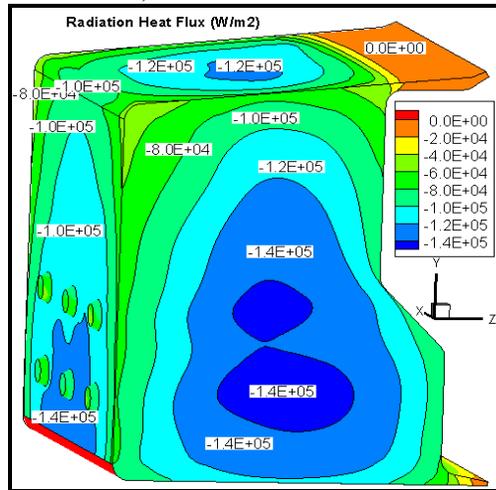


Figure 56 Density profiles at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for different combustion cases, EWBM.

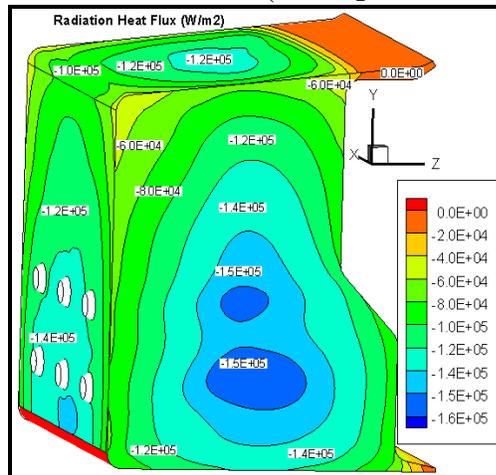
Figures 57 and 58 show the contours of total heat flux (radiative + convective) and radiation heat flux for air-fuel and oxyfuel combustion at the front, top and left walls of the furnace. The negative sign of the heat flux means the flux from the walls to the adjacent gases. Radiation heat flux represents more than 80% of the total heat flux in the furnace in case of air-fuel combustion and about 70% for oxyfuel combustion and this is justifying the importance of selecting the appropriate gas radiation models in order to quantify its value.



a) Air-fuel combustion

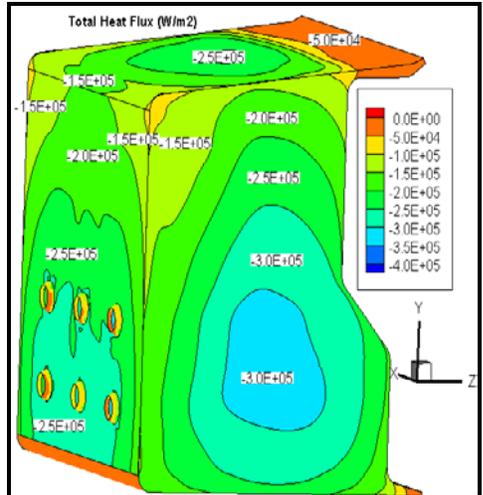


b) Oxyfuel combustion OF21 (21% O<sub>2</sub> and 79% CO<sub>2</sub> by volume)

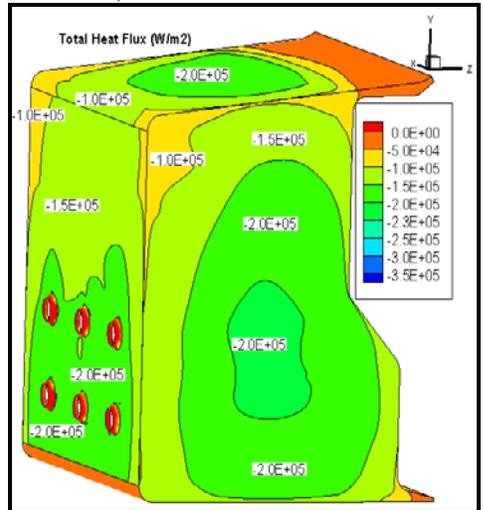


c) Oxyfuel combustion OF27 (27% O<sub>2</sub> and 73% CO<sub>2</sub> by volume)

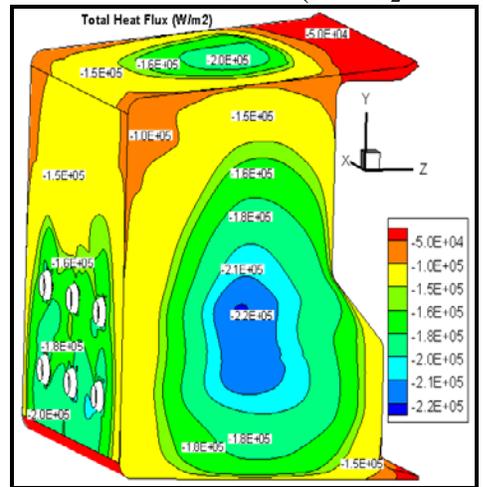
Figure 57 Contours of radiation heat flux for air-fuel and oxyfuel combustion at the front, top and left walls of the furnace, EWBM.



a) Air-fuel combustion



b) Oxyfuel combustion OF21 (21% O<sub>2</sub> and 79% CO<sub>2</sub> by volume)



c) Oxyfuel combustion OF27 (27% O<sub>2</sub> and 73% CO<sub>2</sub> by volume)

Figure 58 Contours of total heat flux for air-fuel and oxyfuel combustion at the front, top and left walls of the furnace, EWBM.

It is found that the heat flux on the walls in case of air-fuel combustion is higher than that for oxyfuel combustion (OF21 and OF27) and this is due to the lower temperature in the oxyfuel combustion. Total heat flux profiles on the front wall ( $x=0$ ) and  $y= -1.0675$  along Z-axis for air-fuel and oxyfuel combustion are plotted in figure 59.

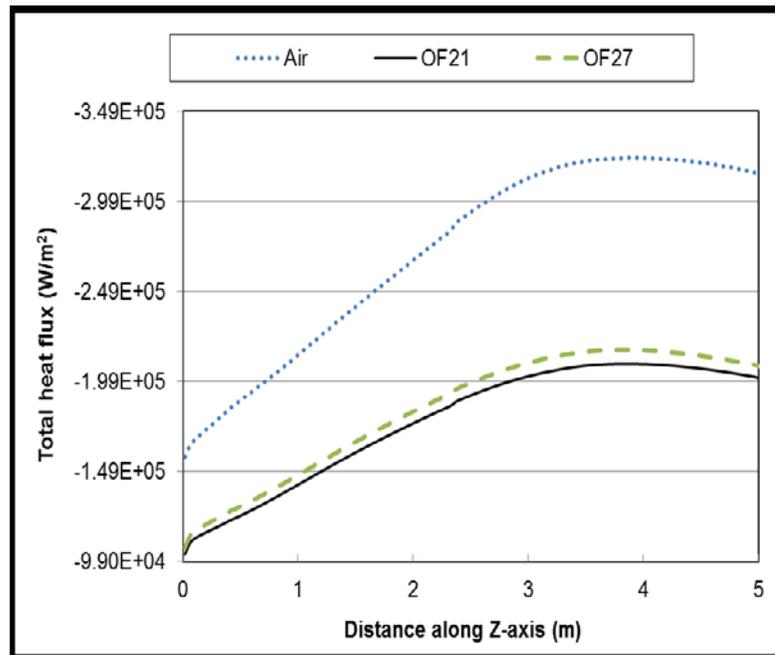


Figure 59 Total heat flux profiles at line ( $x=0, y=-1.0675$ ) along Z-axis for different combustion cases, EWBM.

Incident radiation profiles at line ( $x=3.264, y=-1.0675$ ) along Z-axis for different combustion cases is shown in figure 60. The high incident radiation in air-fuel combustion is translated to high temperature of the combustion products. In case of oxyfuel scenarios and because of replacing  $N_2$  by  $CO_2$ , the incident radiation is reduced due to the cooling effect of  $CO_2$  (i.e. absorbs radiation energy) and this explains the low temperature in such cases.

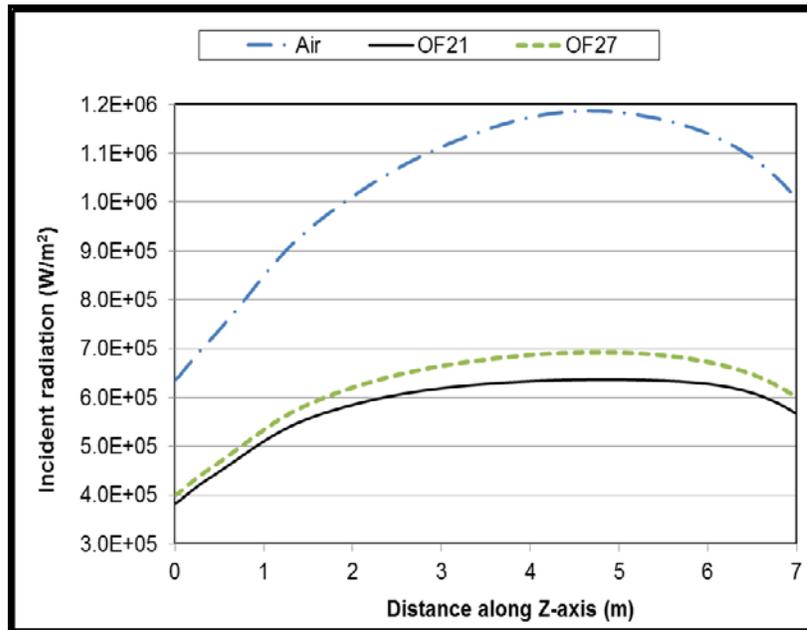


Figure 60 Incident radiation profiles at line ( $x=3.264$ ,  $y=-1.0675$ ) along Z-axis for different combustion cases, EWBM.

# CHAPTER 6

## CONCLUSIONS AND RECOMMENDATIONS

### 6.1 Conclusions

Different gas radiation models have been used in CFD modeling of the typical industrial water tube boiler and their influence on the results was investigated. The characteristics of oxyfuel combustion are compared to those of air-fuel combustion. The influence of the CO<sub>2</sub> circulation on the oxyfuel combustion characteristics was investigated. Based on the discussions and analysis of the results, the following can be concluded:

- Leckner model and Perry model are over-predicting the temperature in comparison to the exponential wide band model.
- Weighted-sum-of-grey-gases model can predict accurate results compared to the benchmark model, however, it requires new parameters for different ratios of H<sub>2</sub>O and CO<sub>2</sub>. It is a computationally efficient option for CFD simulation where there is a need for a simple and accurate gas radiation model.
- One of the limitations of Weighted-sum-of-grey-gases model and Perry model is that they are not applicable for pressure rather than one atmosphere. On the other hand, the exponential wide band model and Leckner model are valid to be used for various pressure and ratios of CO<sub>2</sub> and H<sub>2</sub>O.

- The influence of these models on the prediction of gases concentrations is very limited.
- An optimal value for the absorption coefficient can be determined by comparing the results to the more accurate model (i.e. EWBM).
- The temperature levels are reduced in the case of oxyfuel in comparison to the air-fuel combustion case.
- Increasing the CO<sub>2</sub> recirculation percentage results in higher specific thermal capacity in the combustion products and, thus, resulting in lower temperature levels.
- The fuel consumption rates are slower in the oxyfuel combustion case in comparison to the air-fuel combustion case.
- The energy absorbed is much higher in the case of the air-fuel combustion along the surfaces.

## **6.2 Recommendations for future work**

Oxyfuel combustion is a promising method to capture carbon dioxide (CO<sub>2</sub>) and lower the NO<sub>x</sub> emission. It is shown from the results that the boiler efficiency is dropped in the oxyfuel combustion when oxygen (O<sub>2</sub>) concentration was 27% by volume basis, it is recommended for further study to investigate the influence of increasing the oxygen (O<sub>2</sub>) concentration in the process and its effect on the boiler performance. Furthermore, the soot radiation impact was ignored in this study and therefore it is very important to consider its modeling in the future work. Finally, the design features (shape and size) of

the boiler play a crucial role in the combustion operation and the boiler overall performance under air-fuel or oxyfuel conditions.

## NOMENCLATURE

$I_\nu$	Spectral radiation intensity [ $\text{Wm}^{-2}\text{sr}^{-1}\text{cm}^{-1}$ ]
$I$	Total intensity [ $\text{Wm}^{-2}\text{sr}^{-1}$ ]
$I_b$	Blackbody intensity [ $\text{Wm}^{-2}\text{sr}^{-1}$ ]
$I_{b,\nu}$	Spectral black body intensity [ $\text{Wm}^{-2}\text{sr}^{-1}\text{cm}^{-1}$ ]
$L$	Path length [m]
$c$	Velocity of radiation [m/s]
$A_k$	Total band absorptance of band $k$ [ $\text{cm}^{-1}$ ]
$P_e$	Equivalent broadening pressure
$X$	Density path-length [ $\text{g/m}^2$ ]
$P_E$	Effective pressure [bar]
$a_{e,i}$	Emissivity weighting factors for the grey gas $i$
$b_{e,i,j}$	Emissivity gas temperature polynomial coefficients
$a_{a,i}$	Absorptivity weighting factors
$c_{a,i,j,k}$	Absorptivity polynomial coefficients
$\overline{E}^o_{\nu_{c,k}}$	Blackbody emissive power [ $\text{W}/(\text{m}^2 \cdot \mu\text{m})$ ]

## Greek symbols

$\beta$	Mean line-width-to-spacing parameter
$\varepsilon$	Emissivity
$\varepsilon_c$	Emissivity of carbon dioxide
$\varepsilon_g$	Total gas emissivity
$\varepsilon_w$	Emissivity of water vapour
$\varepsilon_\lambda$	Spectral emissivity
$\zeta$	Partial pressure ratio
$\eta$	Pressure corrected line-width-to-spacing parameter
$\lambda$	Stoichiometric ratio
$\lambda_{\max}$	Pressure correction parameter

$\sigma$	Stefan Boltzmann constant [ $\text{Wm}^{-2}\text{K}^{-4}$ ]
$\tau$	Transmissivity
$\tau_H$	Optical depth at band head
$\tau_k$	Transmissivity of band k
$\tau_{k, i \rightarrow n}$	Transmissivity of band k and path $i \rightarrow n$
$\omega$	Bandwidth parameter [ $\text{cm}^{-1}$ ]
$\varepsilon_0$	Zero partial pressure emissivity
$\kappa_v$	Absorption coefficient [ $\text{m}^{-1}$ ]
$\sigma_v$	Scattering coefficient
$\eta_v$	True emission coefficient
$\Phi_v$	Phase function
$\alpha$	Integrated band intensity [ $\text{cm}^{-1}/(\text{g}/\text{m}^2)$ ]
$\omega$	Band width parameter [ $\text{cm}^{-1}$ ]
$\Delta\varepsilon_{c+w}$	Overlap between bands of $\text{CO}_2$ and $\text{H}_2\text{O}$
$\eta, \mu, \zeta$	Direction cosines
$\Omega$	Solid angle [sr]
$\rho$	Reflectivity of the surface

## References

- [1] <http://www.esrl.noaa.gov/gmd/aggi/>, National Oceanic & Atmospheric Administration, NOAA Research, (visited on December 2010).
- [2] Jordal K, Anheden M, Yan J, Stromberg L., Oxyfuel combustion for coal-fired power generation with CO<sub>2</sub> captures opportunities and challenges, In Seventh conference on greenhouse gas technologies, 2004.
- [3] Wall T. F., Combustion processes for carbon capture, Proceedings of the Combustion Institute 31 (2007) 31–47.
- [4] Hadjipaschalis I., Kourtis G., Poullikkas A., Assessment of oxyfuel power generation technologies, Renewable and Sustainable Energy Reviews 13 (2009) 2637–2644
- [5] Modest M.F., Zhang H., The full-spectrum correlated-k distribution and its relationship to the weighted-sum-of grey- gas method, in: Proceedings of the 2000 IMECE, vol. HTD-366-1, ASME, Orlando, FL, 2000, pp. 75–84.
- [6] Hottel H.C., Sarofim A.F., Radiative Transfer, McGraw-Hill, Inc., New York, 1967.
- [7] Lallemand N., Sayre A., Weber R., Evaluation of emissivity correlations for H<sub>2</sub>O–CO<sub>2</sub>–N<sub>2</sub>/air mixtures and coupling with solution methods of the radiative transfer equation, Progr. Energy Combustion Science, 22 (1996) 543–574.
- [8] Steward F.R., Kocaeffe Y.S., in: U. Grigull, E. Hahne, K. Stephan, J. Straub (Eds.), Proceedings of the 7th International Heat Transfer Conference, vol. 2, Munchen, 1982, pp. 553–558.
- [9] Steward F.R., Kocaeffe Y.S., in: Tien C.L., Carey C.V., Ferrel (Eds.), Proceedings of the 8th International Heat Transfer Conference, Hemisphere Publishing Corporation, New York, 1986, pp. 735–740.
- [10] Smith T.F., Shen Z.F., Fridman J.N., Evaluation of coefficients for the weighted sum of grey gases model, ASME J. Heat Transfer 104 (1982) 602–608.
- [11] Soufiani A., Djavan E., A comparison between weighted sum-of-gray-gases and statistical narrow-band radiation models for combustion applications, Combustion & Flame 97 (1994) 240–250.
- [12] Marakis J.G., Kakaras E., Kakatsios X., Papageorgiou N., Application of the direct numerical integration method in a cylindrical geometry containing non-grey pressurized gases, in: Proceedings Eurotherm # 56, Heat Transfer in Radiating and Combusting Systems-3, 1–3 April 1988, Delphy, Greece, pp. 263–274.

- [13] Zhang L., Soufiani A., Taine J., A spectral correlated and non-correlated radiative transfer in a finite axisymmetric system containing an absorbing and emitting real gas-particle mixture, *Int. J. Heat Mass Transfer* 31 (1988) 2261–2272.
- [14] Modest M.F., The weighted-sum-of-grey-gases model for arbitrary solution methods in radiative transfer, *ASME J. Heat Transfer* 113 (3) (1991) 650–656.
- [15] Denison M.K., Web B.W., A spectral line-based weighted sum-of-grey-gases model for arbitrary RTE solvers, *ASME J. Heat Transfer* 115 (1993) 1004–1012.
- [16] Denison M.K., Web B.W., The spectral line-based weighted sum of grey gases model in non-isothermal and non-homogenous media, *ASME J. Heat Transfer* 117 (1995) 359–363.
- [17] Pierrot L., Riviere Ph., Soufiany A., Taine J., A fictitious gas based absorption distribution function global model for radiative transfer in hot gases, *J. Quant. Spectroscopic Radiative Transfer* 62 (1999) 609–624.
- [18] Pierrot L., Soufiany A., Taine J., Accuracy of narrow-band and global models in H<sub>2</sub>O, CO<sub>2</sub> and H<sub>2</sub>O–CO<sub>2</sub> mixtures at high temperatures, *J. Quant. Spectroscopic Radiative Transfer* 62 (1999) 523–548.
- [19] Trivic D.N., Mathematical modeling of three-dimensional turbulent flow with combustion and radiation, Ph.D. Thesis, Department of Chemical Engineering, University of New Brunswick, Fredericton, NB, Canada, 1987.
- [20] Coelho P.J., Numerical simulation of radiative heat transfer from non-grey gases in three-dimensional enclosures, *J. Quant. Spectroscopic Radiative Transfer* 74 (2002) 307–328.
- [21] Liu F., Numerical solutions of three-dimensional non-grey gas radiative transfer using the statistical narrow-band model, *Trans. ASME* 121 (1999) 200–203.
- [22] Trivic D.N., Modeling of 3-D non-grey gases radiation by coupling the finite volume method with weighted sum of grey gases model, *International Journal of Heat and Mass Transfer* 47 (2004) 1367–1382.
- [23] Buhre B.J.P., Elliott L.K., Sheng C.D., Gupta R.P., Wall T.F., *Prog. Energy & Combustion Science*, 31 (2005) 283–307.
- [24] Modest M.F., *Radiative Heat Transfer*, McGraw- Hill Inc., New York, 1993.
- [25] Kumar R. Renu, Elliott L., Moghtaderi B., Gupta R., Wall T., Differences in Coal Reactivity in Air and Oxy-fuel Conditions and Implications for Coal Burnout. The

Clearwater Coal Conference: The 31st International Technical Conference on Coal Utilization and Fuel Systems, Coal Technology: Whats Next? 2006.

[26] Haynes B.S., Wagner H.G., Soot formation, Progress in Energy Combustion and Science, 7, (1981), 229-273.

[27] Shaddix C.R., Murphy J.J., Coal char combustion reactivity in oxy-fuel applications, Twentieth Pittsburgh Coal Conference, 2003.

[28] Kimura K., Omata K., Kiga T., Takano S., Shikisima S., Energy Convers. Manage. 36 (1995) 805–808.

[29] Okazaki K., Ando T., Energy 22 (1997) 207–215.

[30] Croiset E., Thambimuthu K.V., Fuel 80 (2001) 2117–2121.

[31] Croiset E., Thambimuthu K.V., Palmer A., Chem Can. J., Eng. 78 (2000) 402–407.

[32] Molburg J.C., Doctor R.C., Brockmeier N.F., Plasynski S., CO<sub>2</sub> capture from PC boilers with O<sub>2</sub>-firing. Eighteenth annual International Pittsburgh Coal Conference 2001.

[33] Khare S.P., Wall T.F., Gupta R.P., Elliott L.K., Buhre B.J.P., Oxy-Fuel (O<sub>2</sub>/CO<sub>2</sub>, O<sub>2</sub>/RFG) Technology for Sequestration-Ready CO<sub>2</sub> and Emission Compliance. The Clearwater Coal Conference: The 30th International Technical Conference on Coal Utilization and Fuel Systems, Coal Technology: Yesterday–Today–Tomorrow, 2005.

[34] Khare S, Wall T, Gupta R, Elliott L, Buhre B.J.P., Retrofitting of air-fired pf plant to oxy-fuel: heat transfer impacts for the furnace and convective pass, and associated oxygen production requirements, 5th Asia-Pacific conference on combustion, Adelaide, 17-20 July 2005.

[35] Wall TF, Gupta R, Sheng CD. Calculations of the heat transfer changes for O<sub>2</sub>/CO<sub>2</sub> oxyfuel combustion retrofits of air fired Pf plant with associated recycle ratios and O<sub>2</sub> concentrations fed to the boiler, University of Newcastle, Chemical Engineering, NSW 2308, 2003.

[36] Payne R., Chen S.L., Wolsky A.M., Richter W.F., CO<sub>2</sub> recovery via coal combustion in mixtures of oxygen and recycled flue gas, Combustion Science and Technology, 67 (1989) 1-16.

[37] Chui E.H., Douglas M.A., Tan Y., Fuel 82 (2003), 1201.

[38] Tan Y, Douglas MA, Thambimuthu KV. CO<sub>2</sub> capture using oxygen enhanced combustion strategies for natural-gas power plants. Fuel 81, 2001, 1007–1016.

- [39] Zheng L, Clements B, Runstedtler A., A generic simulation method for the lower and upper furnace of coal-fired utility boilers using both air firing and oxy-fuel combustion with CO<sub>2</sub> recirculation, 27th international technical conference on coal utilization and fuel systems, Clearwater, Florida 2002.
- [40] Zheng L., Clements B., Douglas MA., Simulation of an oxy-fuel retrofit to a typical 400 MWe utility boiler for CO<sub>2</sub> capture, 26<sup>th</sup> international technical conference on coal utilization and fuel systems, Clearwater, Florida 2001.
- [41] Leckner B., Spectral and total emissivity of water vapor and carbon dioxide, *Combustion & Flame* 19 1972, 33–48.
- [42] Andersson Klas, Johnsson Filip, Flame and radiation characteristics of gas-fired O<sub>2</sub>/CO<sub>2</sub> combustion, *Fuel* 86 (2007) 656–668.
- [43] Johansson Robert , Andersson Klas, Leckner Bo, Thunman Henrik, Models for gaseous radiative heat transfer applied to oxy-fuel conditions in boilers, *International Journal of Heat and Mass Transfer* 53 (2010) 220–230
- [44] Leckner B., The spectral and total emissivity of carbon dioxide, *Combustion & Flame* 17 (1971) 37–44.
- [45] Grosshandler W.L., Radiative heat transfer in nonhomogeneous gases: a simplified approach, *Int. J. Heat Mass Transfer* 23 (1980) 1447–1459.
- [46] Soufiani A., Taine J., High temperature gas radiative property parameters of statistical narrow-band model for H<sub>2</sub>O, CO<sub>2</sub> and CO, and correlated-K model for H<sub>2</sub>O and CO<sub>2</sub>, *Int. J. Heat Mass Transfer* 40 (1997) 987–991.
- [47] Rivière Ph., Langlois S., Soufiani A., Taine J., An approximate data base of H<sub>2</sub>O infrared lines for high temperature applications at low resolution, *Statistical narrow-band model parameters*, *J. Quant. Spectroscopic Radiative Transfer* 53 (1995) 221–234.
- [48] Scutaru D., Rosenmann L., Taine J., Approximate intensities of CO<sub>2</sub> hot bands at 2.7, 4.3 and 12  $\mu$ m for high temperature and medium resolution applications, *J. Quant. Spectroscopic Radiative Transfer* 52 (1994) 765–781.
- [49] Andersson K., Johansson R., Johnsson F., Leckner B., Soot formation in propane fired oxy-fuel flames, *Energy Fuels* 22 (2008) 1535-1541.
- [50] Habib M.A., Ben-Mansour R., Badr H.M., Ahmed S.F., Ghoniem A.F., Computational fluid dynamic simulation of oxyfuel combustion in gas-fired water tube boilers, *Computers & Fluids* 56 (2012) 152–165.

- [51] Porter R., Liu F., Pourkashanian M., Williams A., Smith D., Evaluation of solution methods for radiative heat transfer in gaseous oxy-fuel combustion environments, *Journal of Quantitative Spectroscopy & Radiative Transfer* 111 (2010) 2084–2094.
- [52] Johansson Robert, Leckner Bo, Andersson Klas, Johnsson Filip, Account for variations in the H<sub>2</sub>O to CO<sub>2</sub> molar ratio when modeling gaseous radiative heat transfer with the weighted-sum-of-grey-gases model, *Combustion and Flame* 158 (2011) 893–901.
- [53] Hjærtstam Stefan, Johansson Robert, Andersson Klas, Johnsson Filip, Computational Fluid Dynamics Modeling of Oxy-Fuel Flames: The Role of Soot and Gas Radiation, *Energy & Fuels*, ef200983e.
- [54] Raymond Viskanta, *Radiative Transfer in Combustion Systems: Fundamentals & Applications*, Purdue University, USA.
- [55] Siegel Robert, Howell John, *Thermal Radiation Heat Transfer*, Taylor and Francis, New York, 4<sup>th</sup> Edition, 2002.
- [56] Sparrow E. M., Cess R. D., *Radiation Heat Transfer*, McGraw- Hill Inc., New York, 1970.
- [57] Viskanta R, Menguc MP. , *Radiation Heat Transfer in combustion systems*, *Prog Energy Combustion Science*, 1987, 13, 97-160.
- [58] Viskanta R., *Computation of radiative transfer in combustion systems*, *International Journal of numerical methods for heat and fluid flow*, 18, 2008, 415-442.
- [59] Chandrasekhar, S. *Radiative Transfer*, Dover Publications, New York, (1960).
- [60] Edwards, D.K., *Molecular gas band radiation*, *Advances in heat transfer*, 12 (1976), 115-193.
- [61] Liu F., Becker H. A., Bindar Y., *A comparative study of radiative heat transfer modeling in gas-fired furnaces using the simple grey gas and the weighted-sum-of-grey-gases models*, *International journal of heat and mass transfer* 41, 1998, 3357-3371.
- [62] Lallemand, N., Weber, R., *A computationally efficient procedure for calculating gas radiative properties using the exponential wide band model*, *International Journal of Heat Mass Transfer*, 39 (1996), 3273-3286.
- [63] Modak, A.T., *Radiation from products of combustion*, *Fire Research*, (1979) Vol. 1, pp. 339-61.

[64] Hottel, H. C., Noble, J. J., and Sarofim, A. F., 2007, Heat and Mass Transfer, Chapter 5, in Perry's Chemical Engineers' Handbook, 8th Edition, Green, D. W., and Perry, R. H (eds.), McGraw-Hill, New York.

[65] Grosshandler, W. L., 1993, RADCAL: A Narrow-Band Model for Radiation Calculations in a Combustion Environment, NIST Technical Note 1402.

[66] Truelove JS. , A mixed grey gas model for flame radiation, United Kingdom Atomic Energy Authority Report AERER-8494, Harwell, 1976.

[67] Andersson Klas, Characterization of oxy-fuel flames – their composition, temperature and radiation, Ph. D. theses, Chalmers University of Technology, Sweden, 2007.

[68] Fluent 6.2 User's Guide, 2003, Fluent Inc., Center Research Park, 10 Cavendish Court, Lebanon, NH 03766, USA.

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